

Critical Review

Cardiolipin Drives Cytochrome *c* Proapoptotic and Antiapoptotic Actions

Paolo Ascenzi¹, Fabio Polticelli², Maria Marino², Roberto Santucci³ and Massimo Coletta^{3,4}

¹Interdepartmental Laboratory for Electron Microscopy, Via della Vasca Navale 79, Roma, Italy

²Department of Biology, University Roma Tre, Viale Guglielmo Marconi 446, Roma, Italy

³Department of Experimental Medicine and Biochemical Sciences, University of Roma "Tor Vergata," Via Montpellier 1, Roma, Italy

⁴Interuniversity Consortium for the Research on the Chemistry of Metals in Biological Systems (CIRCMSB), Piazza Umberto I 1, Bari, Italy

Summary

Cytochrome *c* (*cytc*) is pivotal in mitochondrial respiration and apoptosis. The heme-Fe-atom of native hexacoordinated horse heart *cytc* (*hhcytc*) displays a very low reactivity toward ligands and does not exhibit catalytic properties. However, on interaction with cardiolipin (CL), *hhcytc* changes its tertiary structure disrupting the heme-Fe-Met80 distal bond. The CL-*hhcytc* complex displays a very low midpoint potential, out of the range required for its physiological role, binds CO and NO with high affinity, facilitates peroxynitrite isomerization to NO₃⁻, and displays peroxidase activity. As a whole, the CL-*hhcytc* complex could play either proapoptotic effects, catalyzing lipid peroxidation and the subsequent *hhcytc* release into the cytoplasm, or antiapoptotic actions, such as scavenging peroxynitrite (*i.e.*, protecting the mitochondrion from reactive nitrogen and oxygen species), and binding of CO and NO (*i.e.*, inhibiting lipid peroxidation and *hhcytc* traslocation). Here, the CL-driven allosteric modulation of *hhcytc* properties is reviewed, highlighting proapoptotic and antiapoptotic actions. © 2011 IUBMB

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Keywords cytochrome *c*; cardiolipin; CO binding; NO binding; peroxynitrite isomerization; lipid peroxidation; apoptosis; allostery.

Abbreviations CL, cardiolipin; *cytc*, cytochrome *c*; *cytc*-Fe(II), ferrous cytochrome *c*; CL-*cytc*, CL-bound cytochrome *c*; *hhcytc*, horse heart cytochrome *c*; *hhcytc*-Fe(III), ferric horse heart cytochrome *c*; *hhcytc*-Fe(II), ferrous horse heart cytochrome *c*; CL-*hhcytc*, CL-bound horse heart cytochrome *c*; CL-*hhcytc*-Fe(III), cardio-

lipin-ferric horse heart cytochrome *c* complex; CL-*hhcytc*-Fe(II), cardiolipin-bound ferric horse heart cytochrome *c* complex; Mb, myoglobin; Mb-Fe(II), ferrous myoglobin.

Eukaryotic cytochromes *c* (*cytc*) are small water-soluble globular heme-proteins that are located within the compartment delimited by the inner and outer mitochondrial membranes playing a pivotal role in mitochondrial respiration and apoptosis (1–5).

In mitochondria, *cytc* is located between the inner and the outer membrane and functions to transfer electrons from Complex III (UQH2-*cytc* reductase) to Complex IV (*cytc* oxidase) of the respiratory chain. It mediates electron transfer through the heme group, which switches between the reduced ferrous form (Fe(II)) and the oxidized ferric (Fe(III)) state. Moreover, in healthy cells, *cytc* inhibits reactive oxygen species formation, thus preventing cell oxidative stress (3, 6–8).

Cytc displays also a central apoptotic role. *Cytc* release into the cytosol is particularly associated with activation of the intrinsic pathway, which responds to intracellular stimuli such as DNA damage and oncogene activation. Multiple cytosolic and mitochondrial proteins regulate the mitochondrial pathway of cell death. Members of the Bcl-2 family regulate events upstream of mitochondria, weighing the prosurvival signals against the stress/damage signals. If the latter prevail, the mitochondrial membrane is permeabilized, leading to deterioration of the bioenergetic functions of mitochondria, overproduction of reactive oxygen species, as well as to the release of *cytc* into the cytosol. Once in the cytosol, in the presence of ATP (and more efficiently in the presence of deoxyATP), *cytc* mediates the allosteric activation and hepta-oligomerization of the adaptor molecule apoptosis-protease activating factor-1, generating the complex known as apoptosome. Each apoptosome can

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Address correspondence to: Paolo Ascenzi, Interdepartmental Laboratory for Electron Microscopy, Via della Vasca Navale 79, Roma I-00146, Italy. Tel: +39-06-5733-3200(2); Fax: +39-06-5733-6321. E-mail: ascenzi@uniroma3.it

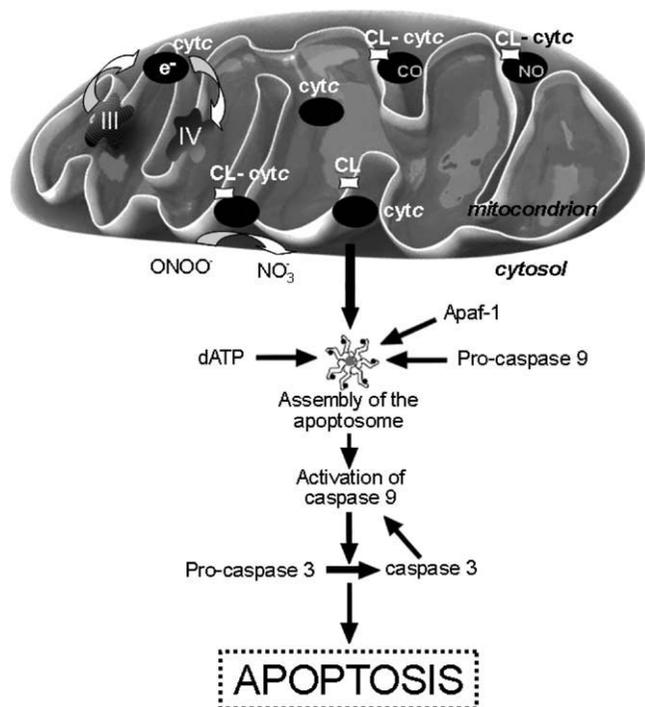


Figure 1. The multiple functions of *cytc*. In the mitochondria, *cytc* mediates electron transfer from Complex III (III) to Complex IV (IV). Mitochondrial *cytc* is placed within the compartment delimited by the inner and the outer membranes or localized into the mitochondrial inner membranes bound to cardiolipin (CL). CL-*cytc* binds CO and NO with high affinity and facilitates peroxynitrite conversion to NO_3^- (antiapoptotic action). CL-*cytc* displays peroxidase activity, and CL being a favorable substrate, this pathway allows *cytc* release into the cytoplasm. Once in the cytosol, in the presence of deoxyATP (dATP), *cytc* mediates the activation of the adaptor molecule apoptosis-protease activating factor-1 (Apaf-1), generating the apoptosome. Apoptosome can recruit caspase 9 favoring proteinase activation. These events induce the catalytic maturation of caspase 3, which mediates the biochemical and morphological features of apoptosis.

recruit seven dimers of caspase 9 favoring proteinase activation. These events, tightly regulated by several heat shock proteins (Fig. 1), allow for the catalytic maturation of caspase 3 and other caspases, which eventually mediate the biochemical and morphological features of apoptosis (9–13).

At least 15% of mitochondrial *cytc* is bound to cardiolipin (CL), an unusual lipid largely confined to the inner mitochondrial membrane (3, 4, 13–20). The interaction with CL is pivotal for switching *cytc* function(s) from mitochondrial respiration to apoptosis. In fact, upon CL binding, *cytc* has been shown (i) to change its tertiary structure disrupting the heme-Fe-Met80 distal bond (the proximal axial ligand being His) and, in some cases, to vary the spin state of the metal (4, 14, 18–22), (ii) to reduce drastically the midpoint potential out of the

range required for its role in the respiratory chain (17), (iii) to display peroxidase activity, using CL as a favorable substrate (15, 23–26), (iv) to bind CO and NO with high affinity (27, 28), and (v) to facilitate peroxynitrite scavenging (*i.e.*, conversion to NO_3^-) (29). All these effects suggest that CL-bound *cytc* (CL-*cytc*) could play either proapoptotic effects, catalyzing lipid peroxidation, and *cytc* release into the cytoplasm, or antiapoptotic actions, scavenging peroxynitrite (*i.e.*, protecting the mitochondrion from reactive nitrogen and oxygen species), and binding CO and NO (*i.e.*, inhibiting lipid peroxidation and *cytc* translocation). Here, the CL-driven *cytc* proapoptotic and antiapoptotic actions are reviewed.

HOW CARDIOLIPIN BINDS TO HORSE HEART CYTOCHROME *c*?

CL, which constitutes about 20% of total lipids of the mitochondrial membrane, is synthesized in the mitochondrion and possesses a unique structure, being composed of four (instead of two, as in most lipids) acyl chains (20, 30).

To the authors' knowledge, no crystal structures are available for the CL-*cytc* complex, thus the proposed binding modes are based on available experimental data and molecular modeling investigations. Although CL/*cytc* recognition is uncertain and still matter of controversy, both regions considered by molecular modeling studies provide suitable sites for CL binding (18, 20, 21).

The first model to be proposed suggests that on CL binding to horse heart *cytc* (hhcytc), one acyl chain of CL protrudes into the protein interior through the hydrophobic channel located close to the Asn52 residue. The insertion of the acyl chain into hhcytc is assumed to be favored by H-bonding between Asn52 and the protonated phosphate group of CL (Fig. 2) (21).

An alternative model asserts that the binding of CL to hhcytc occurs in the region of the Met80-containing loop, and that the acyl chain protrudes into the protein between the hydrophobic strands formed by residues 67–71 and 82–85 after anchoring of the phospholipid to the protein via electrostatic interactions between the deprotonated phosphate group of CL and the Lys72 residue of hhcytc (Fig. 2) (18).

Very recently, it has been hypothesized that CL binds to hhcytc at two distinct regions (20), which are characterized by different affinity for phospholipids (14, 19). According to this model, the acyl chains of CL could be easily accommodated both in the hydrophobic channel in the vicinity of Asn52 and in the region of the Met80-containing loop (Fig. 2). In the latter case, the deprotonated phosphate group of CL could electrostatically interact with the Lys72 and Lys73 residues. The view that two acyl chains bind to hhcytc at distinct sites is in agreement with the observed two-state transition binding process and with the biphasic character of CL-binding kinetics. This hypothesis is very realistic from a stereochemical viewpoint, in that the insertion of only one acyl chain into hhcytc would cause the

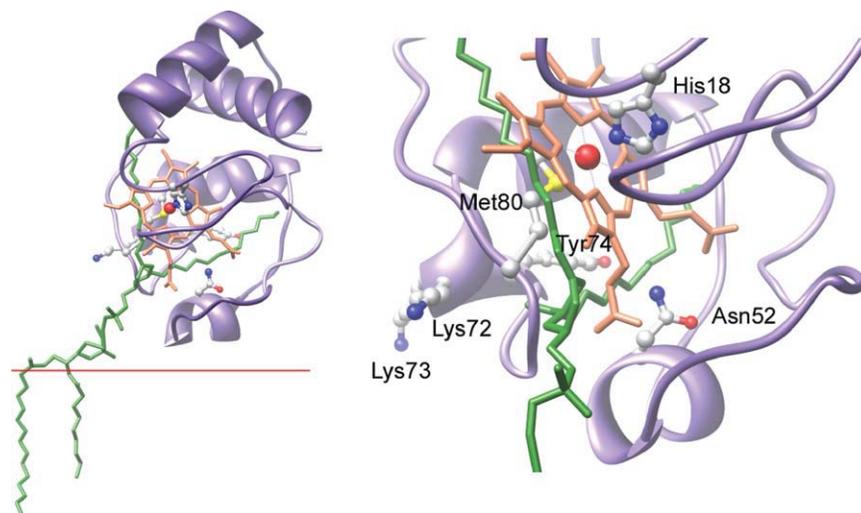


Figure 2. Molecular model of the putative CL-hhcytc complex. The CL molecule is shown in green, and the membrane plane is indicated by a red line. The enlarged view of the CL-hhcytc complex (right panel) shows CL binding to hhcytc by insertion of two of the four acyl chains in the vicinity of Asn52 and Met80. The proximal ligand of the heme-Fe-atom His18 and residues Lys72, Lys73, and Tyr74 are also shown. This CL-binding mode allows the tethering of hhcytc to the membrane as CL can remain membrane-bound through the other two acyl chains. The insertion of a single acyl chain into the hhcytc molecule is unlikely due to stereochemical constraints, which would force the adjacent CL acyl chain to be solvent exposed. [Color figure can be viewed in the online issue, which is available at wileyonlinelibrary.com.]

partial exposure to the solvent of (at least) one of the adjacent acyl chains in CL, the situation disfavored from a solvation energy viewpoint (Fig. 2) (20).

As a whole, CL binding to hhcytc seems to induce a gross conformational change(s) on both the proximal and distal side of the heme, causing the loss of the electron transfer properties of hhcytc and its transformation into a globin- and peroxidase-like heme-protein (15, 17, 27–29).

THE CL-BOUND HORSE HEART CYTOCHROME *c* DISPLAYS PROAPOPTOTIC AND ANTI-APOPTOTIC PROPERTIES

The CL-hhcytc interaction plays an important role in modulating the heme-protein functions. CL-bound cytc shows a non-native tertiary structure and a disrupted heme-Fe-Met80 distal bond, with Lys79 as the likely sixth axial ligand of the heme-Fe-atom at pH > 9 (14, 18–22). CL binding to hhcytc induces a drastically reduced midpoint potential, which falls out of the range required for its physiological role in mitochondrial respiration (17). Further, the cleavage of the distal Fe-Met80 bond endows hhcytc with proapoptotic activity, due to the achievement of peroxidase action (15, 23–26, 31, 32), and anti-apoptotic functions, increasing the affinity for CO and NO (27, 28) and inducing peroxynitrite detoxification properties (29).

Proapoptotic Activity of CL-bound Horse Heart Cytochrome *c*

At the first stage of apoptosis, cytc released into the cytosol binds to apoptosis-protease activating factor-1. This event starts

the process leading to the cleavage of protein substrates and subsequent cell death. Hhcytc release is induced by the dissociation of the CL-hhcytc complex; hhcytc-mediated CL peroxidation is indicated as the process responsible for such an event, although the exact mechanism governing it is not yet fully understood. CL peroxidation is responsible for hhcytc detachment from the mitochondrial membrane. This supported by several evidences, such as the observation that the oxidative degradation of CL occurs in the p53-induced apoptosis (33) and that CL hydroperoxides show a decreased affinity for hhcytc with respect to CL (4, 31). CL oxidation leads to structural changes of hhcytc, such as partial protein unfolding, weak axial binding to the heme iron, and enhanced access of the heme catalytic site to small molecules (*e.g.*, hydrogen peroxide), which result in a modification of the hhcytc catalytic properties, transforming it into a peroxidase (8, 34).

Antiapoptotic Activity of CL-bound Horse Heart Cytochrome *c*

CL binding to hhcytc facilitates CO and NO binding to the heme-Fe-atom and peroxynitrite isomerization (27–29). While native ferrous hhcytc (hhcytc-Fe(II)) is unable to bind CO (35), ferrous CL-hhcytc (CL-hhcytc-Fe(II)), like mammalian myoglobins (Mbs) (36), interacts with CO by a simple second-order process (27). The value of the apparent dissociation equilibrium constant for CL-hhcytc-Fe(II) carbonylation ($\sim 3 \times 10^{-8}$ M) (27) corresponds to that reported for ferrous horse heart Mb (Mb-Fe(II)) generally taken as a molecular model ($\sim 3 \times 10^{-8}$ M) (36). However, both combination and dissociation rate

constants for (de)carbonylation of CL-hhcytc-Fe(II) ($\sim 1 \times 10^7 \text{ M}^{-1} \text{ s}^{-1}$ and 0.18 s^{-1} , respectively) (27) are larger than those reported for horse heart Mb-Fe(II) (de)carbonylation, ($= 5 \times 10^5 \text{ M}^{-1} \text{ s}^{-1}$ and $1.7 \times 10^{-2} \text{ s}^{-1}$, respectively) (36). This suggests a very open heme crevice in CL-hhcytc and a weak proximal His18-Fe bond (27), these characteristics seem to be confirmed also by the very fast dissociation rate of the His18-Fe bond on NO binding to the sixth coordination position of the heme-Fe-atom (28).

The value of the dissociation equilibrium constant for NO binding to hhcytc-Fe(II) ($= 8.2 \times 10^{-6} \text{ M}$) (37) is less favorable than that for CL-hhcytc-Fe(II) nitrosylation ($= 2.5 \times 10^{-11} \text{ M}$) (28), which is similar to that reported for horse heart Mb-Fe(II) ($= 7.1 \times 10^{-12} \text{ M}$) (38).

NO binds to the sixth coordination position of the heme-Fe-atom of hhcytc-Fe(II) by a simple second-order process as Mb (37). Nevertheless, the nitrosylated CL-hhcytc-Fe(II) complex is pentacoordinated with NO as the proximal fifth ligand of the heme-Fe-atom (28).

In CL-hhcytc-Fe(II), NO replaces the proximal His18 residue via an unusually complex kinetic mechanism. Indeed, the nitrosylation of CL-hhcytc-Fe(II) involves seven species: (i) the initial NO-unbound pentacoordinated species (the proximal fifth axial ligand is His18), (ii) the transient ferrous nitrosylated hexacoordinated species (the proximal fifth axial ligand is His18 and the distal sixth axial ligand is NO), (iii) the transient ferrous nitrosylated pentacoordinated species (the proximal His18-Fe bond is cleaved and the distal sixth axial ligand is NO), (iv) the transient ferrous nitrosylated hexacoordinated species (the proximal fifth axial ligand is a weak-field ligand, possibly an intrinsic amino acid residue or a water molecule), (v) the transient ferrous bis-nitrosylated hexacoordinated species (two NO molecules are the fifth and the sixth axial ligands of the heme-Fe-atom), (vi) the transient ferrous nitrosylated pentacoordinated species (the proximal sixth axial ligand is NO) not identical to the final species, and (vii) the final ferrous nitrosylated pentacoordinated species (the proximal sixth axial ligand is NO). The denitrosylation of CL-hhcytc-NO is characterized by: (i) the transient tetraordinated species following the cleavage of the proximal NO-Fe-atom bond, but retaining NO in the protein matrix, (ii) the pentacoordinated species (the proximal fifth axial ligand is His18), but retaining NO in the protein matrix, and (iii) the stable NO-unbound pentacoordinated species (the proximal fifth axial ligand is His18). This suggests a remarkable mobility of the heme environment of hhcytc-Fe(II) induced by CL (28).

Finally, NO induces the reductive nitrosylation of ferric hhcytc (hhcytc-Fe(III)) at $\text{pH} \geq 7$ as reported for ferric heme-proteins, including sperm whale Mb. In fact, the addition of NO to heme-Fe(III) leads to the transient formation of heme-Fe(III)-NO in equilibrium with heme-Fe(II)⁻NO⁺. Then, heme-Fe(II)-NO⁺ undergoes nucleophilic attack by OH⁻ to yield hhcytc-Fe(II), which in turn reacts further with NO to give hhcytc-Fe(II)-NO. This process, occurring *in vitro* under anaerobic conditions, may occur *in vivo* under hypoxia (39).

The unusual features of CO and NO binding to the distal and proximal sides of the heme-Fe-atom, respectively, may represent a new mechanism in the regulation of biological processes by sensing changes in the concentration of CO and NO. Remarkably, a similar situation has been reported for gas discrimination by cytochromes *c'*, a distinct family of class IIa cytochromes found in the periplasm of certain denitrifying, nitrogenfixing, photosynthetic, methanotrophic, and sulfur-oxidizing bacteria (40).

As reported for pentacoordinated ferric heme-proteins, including horse heart and sperm whale Mb (41–43), ferric CL-hhcytc (CL-hhcytc-Fe(III)) quickly catalyzes peroxynitrite scavenging, inducing the formation of nitrate (the process takes 10^{-3} to 1 s) (29). The value of the second-order rate constant for CL-hhcytc-Fe(III)-mediated scavenging of peroxynitrite ($= 3.2 \times 10^5 \text{ M}^{-1} \text{ s}^{-1}$) (29) is similar to those reported for ferric heme-proteins (ranging between $1.2 \times 10^4 \text{ M}^{-1} \text{ s}^{-1}$ and $4.1 \times 10^5 \text{ M}^{-1} \text{ s}^{-1}$) (41–43). However, it has also been shown that peroxynitrite very slowly induces the nitration of the solvent-exposed Tyr74 residue of hhcytc in the absence of CL (the process takes 10 to 30 min). This leads to the cleavage of the Fe-Met80 bond, which is substituted by a weak Fe-Lys72 heme distal ligation (44).

Kinetic and thermodynamic parameters for CO and NO binding to CL-hhcytc-Fe(II) are very favorable, suggesting that the physiological levels of CO and NO may impair CL-hhcytc proapoptotic action inhibiting the *cytc* peroxidase activity (27, 28). Moreover, peroxynitrite scavenging by CL-hhcytc-Fe(III) may protect the mitochondrion from reactive nitrogen and oxygen species helping cell survival (29).

CONCLUSION AND PERSPECTIVES

CL binding to hhcytc induces tertiary changes facilitating the cleavage of the distal heme-Fe-Met80 bond, and this favors the enzymatic activity of CL-hhcytc (40). Data reported so far indicate that CL-hhcytc could act as either a proapoptotic or an antiapoptotic factor, depending on the conditions under which it operates. In particular, CL-hhcytc functions as a proapoptotic factor catalyzing the peroxidative reduction of H₂O₂, which leads to CL peroxidation (31, 34). In contrast, CL-hhcytc exerts an antiapoptotic action facilitating the isomerization of peroxynitrite to nitrate with the consequent scavenging of reactive nitrogen species (29). Further, the reaction of CO and NO with CL-hhcytc-Fe impairs the hhcytc peroxidase activity (27, 28).

As a whole, CL-*cytc* could act as a modulator of the apoptotic cascade depending on the levels and the type of oxidizing species present in the cellular microenvironment. This pathway appears more complicated in that cytosolic *cytc* would bind to several (macro)molecules. Note that *cytc* could interact with neuroglobin, a recently discovered neuroprotectant globin in neurons (45–49). The neuroglobin-*cytc* interaction would be significantly enhanced by the electrostatic interactions between the two proteins, as *cytc* is an unusually basic protein ($\text{pI} =$

10.2) while neuroglobin is an acidic protein (pI = 4.6). Thus, at neutral pH, neuroglobin would be highly negatively charged, whereas cytc would be highly positively charged. The finding that ferrous neuroglobin and ferric cytc could react leads to the hypothesis that neuroglobin can inhibit cytc apoptotic activity allowing cell survival under stress conditions by reducing the heme-Fe-atom of cytc (48, 49). Moreover, several cytosolic and mitochondrial tRNAs specifically associate with cytc (50). Notably, tRNA blunted the ability of cytc to induce apoptosis while degradation of tRNA by a RNase enhanced apoptosis via the intrinsic pathway. The cytc-tRNA interaction (a previously unexpected connection between two ancient molecules) is pivotal in apoptosis, and may represent an evolutionarily conserved connection between metabolism and cell survival (50, 51).

As a whole, the ability of triggering either proapoptotic and antiapoptotic processes renders cytc complexes with (macro)molecules crucial elements for the regulation of the cell fate, thus opening a new (although complex) scenario where the role of cytc appears to be significantly more extensive than what thought so far.

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