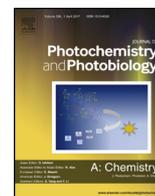




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Invited feature article

A bioinspired dye sensitized solar cell based on a rhodamine-functionalized peptide immobilized on nanocrystalline TiO₂



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ABSTRACT

A dodecapeptide (AMRKLDPAGMH) functionalized with a tetramethylrhodamine (TAMRA) chromophore at the N-terminus was immobilized on nanocrystalline TiO₂. The optical and binding properties of the peptide layer immobilized on the titania surface were characterized by UV–vis absorption, steady-state fluorescence and X-ray photoelectron spectroscopies. Circular Dichroism experiments and Molecular Mechanics calculations showed that the predominant conformation populated by the peptide scaffold brings Arg3, Lys4 and Asp7 in the correct position for linking the TiO₂ surface. Photocurrent generation experiments were therefore carried out to determine the photon-to-current conversion efficiency (IPCE) of a Grätzel-like Dye Sensitized Solar Cell (DSSC), the photoactive unit of which is formed by TAMRA-AMRKLDPAGMH/TiO₂. The measured IPCE amounted to 0.65%, a value that is definitely low, but superior to those previously reported for similar bioinspired DSSCs. This result can be ascribed to the light-harvesting properties of the TAMRA chromophore and to the unique structural properties of the peptide spacer.

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1. Introduction

In the last decade, dye-sensitized solar cell (DSSC) technology has become a realistic alternative to classical silicon technology in solar energy conversion, because of its low cost, easy fabrication and reduced use of hazardous reagents during manufacturing [1].

DSSCs originated from the idea of mimicking photosynthesis, *i.e.* convert sunlight into electrical current by means of light-absorbing dyes (analogously to the light-absorbing chlorophyll molecules in Nature) [2]. They are composed of a conducting glass coated by nanocrystalline titania, functionalized with a dye that, under illumination, injects electrons to TiO₂, a semiconductor with a band gap of 3.2 eV. As in a conventional alkaline battery, the TiO₂ surface works as the anode, while a platinum electrode usually accomplishes the role of the cathode. A suitable electrolytic

solution is inserted between the two electrodes to allow electronic conduction.

Several groups have obtained 8–10% solar energy conversion efficiency using DSSC's based on nanocrystalline TiO₂ [3,4]. To increase the efficiency of the cell, different dye molecules [5], electrolytes [6], electron acceptors [7], co-sensitizers [8], absorber materials like perovskite [9] or different dye-titania linkers have been extensively tested [10].

In biological systems, electron transfer (ET) reactions efficiently occur along a sequential array of electron donor/acceptor moieties surrounded by a polypeptide matrix that is considered as a specialized ET mediator [11]. Helical secondary structures are the most frequently observed motifs in such polypeptide matrixes and are considered to play a crucial role in ET processes [12].

We have already demonstrated that peptide self-assembled monolayers (SAMs) represent the most suitable tool for modifying the surface properties of metallic substrates, paving the way to the realization of hybrid devices (*soft meets hard*) [13,14]. The introduction of SAMs, and in particular the identification of the

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thiol/Au route, has deeply transformed surface chemistry, opening a new pathway for preparing covalently linked thin films [15]. The great plasticity of SAMs brought about by the large choice of end groups that can be linked to the substrate, has greatly improved the number of applications in bio-inspired nanotechnology. Furthermore, it has been demonstrated that α -helical peptides linked on semiconducting TiO₂ are able to modify the electronic properties of the organic/solid interface, enhancing the charge-transfer efficiency of the system [16].

In this connection, a bio-inspired approach to DSSC technology, was recently developed in our laboratory making use of a helical peptide spacer, functionalized by a dye molecule and immobilized on TiO₂ nanoparticles coated by a 5 nanometer-thick layer of gold [17].

In this work, we report on the photocurrent generation properties of a dodecapeptide functionalized with a rhodamine dye on a titania surface. Stable peptide films on TiO₂ surfaces have already been obtained by using peptide chains with a specific binding sequence, namely the hexapeptide motif RKLPGA [18]. Interestingly, it has been found that the interaction of RKLPGA with TiO₂ nanoparticles forced the hexapeptide to kink around the *cis*-P residue, directing R and D to the surface, so that the R and D side chains bound via electrostatic interaction to hydroxyl groups, that can act as either a basis (Ti-O⁻) or acid (Ti-OH₂⁺) groups [19]. The choice of this specific, aqueous-soluble sequence, endowed with several potential anchoring groups, aimed to avoid the dye desorption in the organic solvent used in the photocurrent generation measurements, allowing for the realization of an efficient, long-life cell.

In this work, the RKLPGA motif was inserted between an N-terminal dipeptide, *i.e.* AM, and a C-terminal tetrapeptide, *i.e.* PGMH, to form the dodecapeptide investigated. It should be noted that both the N- and C-terminal end fragments predominantly comprise apolar amino acids. Furthermore, a 5-carboxytetramethylrhodamine (TAMRA) chromophore was covalently linked to the N-terminus of the dodecapeptide, to act as a photosensitizer strongly absorbing in the visible region of the electromagnetic spectrum.

Xanthene dyes and their derivatives have been extensively used for biosensing applications, due to their low toxicity, high absorption coefficient, and high fluorescence quantum yield [20]. These properties make rhodamine-based chromophores promising candidates for applications where high sensitivity and resistance to photolysis are needed.

The molecular structure and acronym (T12) of the compound investigated are shown in Scheme 1.

2. Materials and methods

2.1. Materials

TAMRA-AMRKLPGA (T12) was synthesized by Eurogentech (Belgium) with a purity higher than 95%. All solutions

were prepared using spectroscopic grade solvents (Carlo Erba, Milan, Italy). Water was bi-distilled and de-ionized by passing through a Milli-Q purification system. Other chemicals have all reagent grade quality and were used without further purification. Other chemicals: HEPES (Sigma-Aldrich), sodium sulfate (Sigma-Aldrich), potassium chloride (Sigma-Aldrich).

2.1.1. Preparation of TiO₂/Fluorine-doped tin oxide (FTO) substrates and T12 film preparation

Nanocrystalline titania layers were prepared by adsorbing screen printing Dyesol 18NR-T paste onto a transparent conductive glass substrate (Pilkington, TEC 8, 8 Ω /sq), followed by sintering at 480 °C for 30 min. The final film thickness, controlled by a Veeco Dektak 150 profilometer, was 6 μ m. The size of the photoelectrode was 5 \times 5 mm². After calcination, the photoelectrode was soaked in a TiCl₄ aqueous solution [21].

For all experiments TiO₂/FTO substrates were washed with bi-distilled water and ethanol before immersion in a millimolar peptide solution. Peptide-coated supports were prepared by dipping a clean substrate into a 2 mM peptide HEPES buffer solution (50 mM, pH = 7). After 2 h, the peptide-modified substrate was repeatedly rinsed with the buffer solution to remove physically adsorbed peptides and dried with a gentle flow of Argon.

2.1.2. Preparation of a DSSC cell

Transparent counter electrodes were prepared by firing a solution of Pt-paste (CHIMET) deposited by screen printing on a conductive glass for 15 min at 420 °C. These two electrodes were assembled by using a 60 μ m-thick Surlyn gasket (Dupont 1702). The electrolyte solution was introduced by backfilling through a hole previously realized on the counter electrode.

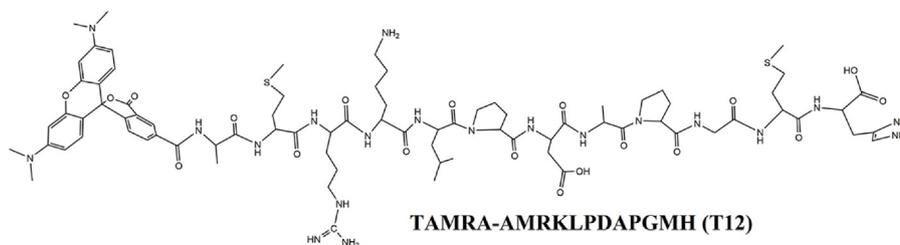
2.2. Methods

2.2.1. Spectroscopy and photoelectrochemistry

All UV-vis absorption experiments were carried out at room temperature with a Cary 100 Scan spectrophotometer (Varian), using quartz cuvettes (Hellma) with optical lengths of 1, 0.5 and 0.1 cm.

Circular dichroism (CD) spectra were recorded in HEPES buffer solution (pH 7) by a J-600 (Jasco) spectrometer, using a 0.1 cm quartz cuvette. Concentrations of peptide solutions were 10⁻⁵ M. Molar ellipticities are reported in terms of molar concentration per residue.

All steady-state fluorescence measurements were carried out on a SPEX Fluoromax-4 (HORIBA, Jobin Yvon) spectrofluorimeter using Single Photon Counting (SPC) acquisition mode. All spectra in solution were recorded by using quartz cuvettes with an optical length of 1 cm. All the fluorescence experiments were carried out on solutions showing absorbance less than 0.03 to minimize inner-filter effect. Fluorescence spectra of peptide films supported on TiO₂/FTO substrates were recorded in transmittance mode on a



Scheme 1. Molecular formula and acronym of the peptide investigated.

support oriented at an angle of 30° with respect to the excitation beam direction.

Fluorescence time decays were recorded using a LifeSpec instrument (Edinburgh Analytical Instruments) equipped with a multi-channel analyzer (MCA). The excitation source was a 460 nm diode laser. The full width at half maximum (FWHM) of the excitation pulse was 1 ns. Experimental time decays were fitted by non-linear least square analysis through iterative reconvolution of a multiexponential trial function with the experimental excitation source profile.

X-ray Photoelectron Spectroscopy (XPS) measurements were recorded with a semi-imaging analyzer MAC 2 (Riber Instruments) operating in the constant pass energy mode. A non-monochromatized Al K α (1486.6 eV) radiation source (9 kV, 700 W) was used. The distance between the sample and the anode was about 40 mm; the illumination area was about 5 × 5 mm², while the take-off angle between the sample surface and the photoelectron energy analyzer was kept at 45°. The energy scale was calibrated with reference to the binding energy of the Ti 2p_{3/2} signal of TiO₂ having a binding energy BE = 458.7 eV. Curve fitting analysis of the C 1s, N 1s and O 1s spectra was done using a standard Gaussian curve fit routine with Shirley background subtraction. The quality of the fit was evaluated by using χ^2 minimization test.

Photocurrent generation experiments were carried out using a Xenon lamp light source equipped with a monochromator (band pass 15 nm); the power density of the incident light was evaluated by a power meter. A PG-310 potentiostat was used as a current detector. The electrochemical cell adopted a standard three-electrode configuration. The electrolyte solution was 0.01 M I₂, 0.5 M KI in acetonitrile. Electronic current vs. applied voltage (I–V) spectra were carried out with a Keithley 2420 Source. I–V curves were measured under an air mass (AM) 1.5G solar simulator (Solar Constant from KHS) at 1 Sun lighting.

2.2.2. Conformational analysis of T12

Molecular Mechanics (MM) calculations were carried out by using the MM4 force field, where solvents effects are only implicitly taken into account [22]. The MM4 results were validated by using the AM1 and PM3 semiempirical methods [23]. Initial conformations were generated assuming a statistical population of folded and extended helical peptide structures by using standard bond lengths and bond angles for all residues. The Pro6 and Pro8 residues were initially assumed to adopt a polyproline type II conformation. The conformational energy was therefore optimized by taking into account electrostatic, non-bonding, hydrogen bond and torsional interactions. Stretching and bending motions, as implemented in the MM4 force field were also taken into account.

3. Results and discussion

3.1. Optical spectroscopy and conformational analysis in solution

The UV–vis absorption and fluorescence spectra of **T12** in HEPES buffer solution (pH = 7) are both reported in Fig. 1. The former shows the typical, broad $\pi \rightarrow \pi^*$ absorption band of rhodamine B featuring an intense absorption peak at $\lambda_{\text{max}} = 550$ nm [24]. From the figure, it can be easily seen that the S₁ → S₀ emission band, peaked at $\lambda_{\text{max}} = 572$ nm, almost mirrors the $\pi \rightarrow \pi^*$ absorption band.

Time resolved fluorescence measurements provided a single lifetime of 2.4 ns for all the **T12** solutions investigated (1–50 μM), a value similar to the reported lifetime of cationic rhodamine B [25].

The conformational properties of **T12** in a HEPES buffer solution (pH 7) were investigated by Circular Dichroism measurements (Fig. 2). The deep minimum observed at around 205 nm ($n \rightarrow \pi^*$ transition), suggests that **T12** attains a statistical population of

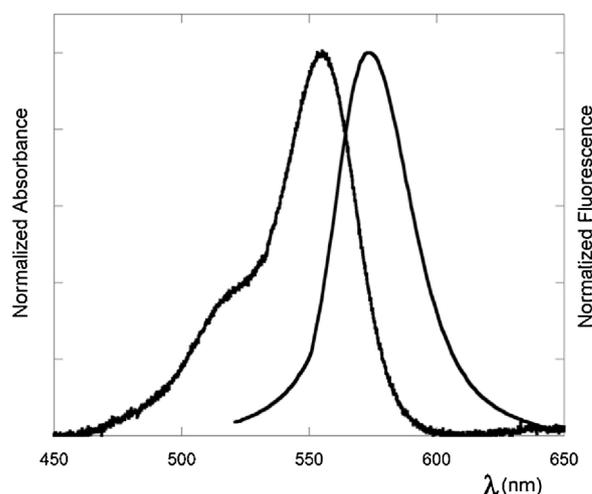


Fig. 1. Absorption and fluorescence emission spectrum of T12 in a HEPES buffer solution (pH = 7) at 1 μM concentration.

conformers with a predominant contribution of polyproline type II-like conformation [26]. The latter is a left-handed helix ($\phi = -75^\circ$; $\chi = 146^\circ$; $\omega = 180^\circ$) with 3 residues per turn and a 3.1 Å rise per residue, characterized by a predominant extended structure [27]. For this unique property, oligoprolines have traditionally been used as molecular spacers in FRET studies, although the rigidity of such spectroscopic rulers has been recently questioned [28].

Molecular Mechanics (MM, AM1, PM3) calculations were carried out to analyze the conformational preferences of **T12**, focusing on two specific issues: i) the localization of groups, *i.e.* peptide side-chains, potentially capable of linking to the TiO₂ surface, and ii) the localization of the TAMRA chromophore and its potential interaction with the surface and/or with other rhodamine groups.

The most stable conformer provided by MM calculations is reported in Fig. 3. It was found that the presence of Pro9 strengthens the propensity of the RKLPGA motif to preferentially adopt an extended conformation mimicking polyproline type II, in agreement with CD results. A very similar W-shaped structure was recently found by Suzuki et al. [29], investigating the binding of RKLPGA to TiO₂ nanoparticles by NMR experiments. In particular,

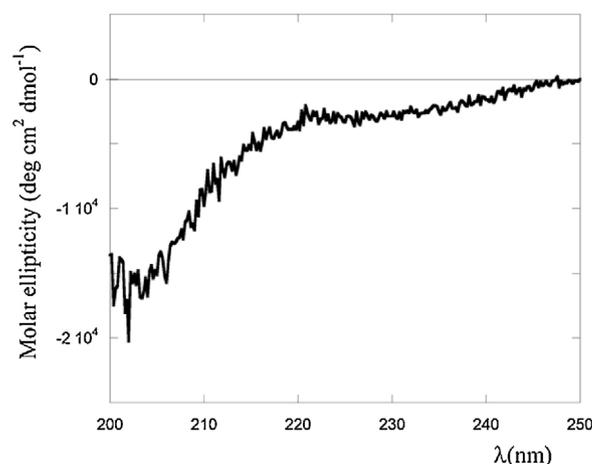


Fig. 2. Circular dichroism spectrum of T12 in HEPES buffer solution (pH 7) at 1 μM concentration.

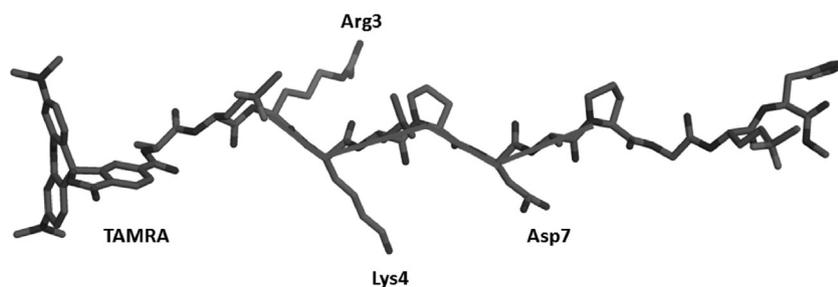


Fig. 3. Most stable conformer of **T12**, as provided by MM calculations. Notice the location of the Arg3, Lys4 and Asp7 side-chains, all suitably positioned for linking to a TiO_2 surface. The orange segment emphasizes the rigid, W-like skeleton of the dodecapeptide investigated.

they found that Arg1 and Lys2, through their N-terminal groups, were principally responsible for anchoring the peptide to the TiO_2 surface. However, they also found that the C-terminal segment of the peptide is rather flexible.

In our case, the presence of Pro9 confers some extra-rigidity to the C-terminal end of **T12**. Furthermore, we found that, besides Arg3 and Lys4, also Asp7 is suitably positioned to bind to positively-charged group ($-\text{OH}_2^+$) on the TiO_2 surface. Regarding the localization of TAMRA, MM calculations suggest that it could be held in the proximity of the TiO_2 surface by the Arg3 and Lys4 anchoring groups. Due to the relative rigidity of the peptide scaffold, the **T12** low-energy conformers are essentially characterized by a different orientation of the TAMRA chromophore. Considering Arg3, Lys4 and Asp7 in the proper position for binding to the TiO_2 surface, the minimum distance from TAMRA to the surface amounts to 3.5 Å, while the most stable conformer shows a minimum distance from TiO_2 of about 5 Å.

This orientationally restricted location and the presence of the sterically hindered N-methyl groups on the TAMRA scaffold, most likely would inhibit short-range rhodamine-rhodamine interactions and block the chromophore in a position in the proximity of the conductive surface, but avoiding the overlap between the TAMRA electronic states and the surface density of states of TiO_2 .

3.2. Optical and X-ray photoelectron spectroscopy of **T12** adsorbed on TiO_2

Interestingly, after deposition on the titania substrate, **T12** film shows a marked blue shift of the $\pi \rightarrow \pi^*$ absorption band ($\lambda_{\text{max}} = 530 \text{ nm}$, Fig. 4A), and a rather intense fluorescence emission spectrum (Fig. 4B), quite similar to the monomer emission (Fig. 2). These results strongly suggest, in agreement

with the computational results, that TAMRA is not interacting with the manifold of the semiconductor surface states, otherwise his fluorescence should be almost totally quenched. The fact that the TAMRA emission retains the shape of the monomer emission, also strongly suggests that TAMRA-TAMRA interactions are negligible. Self-association in solution or at the solid/liquid interface is a frequently encountered phenomenon in dye chemistry, owing to attractive van der Waals interactions between the aromatic moieties. The resulting aggregates exhibit distinct changes in the absorption band as compared to the monomeric species, depending on the aggregation geometry [30]. For example, H-type aggregates are characterized by a sandwich-like configuration of the aromatic moieties, giving rise to hypsochromically-shifted absorption bands. In this case, the symmetry-allowed transition involves the electronic ground state and the high energy excitonic state [31], with a consequent shift of the absorption band to the blue side of the spectrum. In this type of aggregates, the relaxation to the ground state is very fast and generally occurs between excitonic states with vanishing dipole transition moments. As a result, most of the excitation energy is dissipated through non-radiative pathways, and H aggregates usually show a very low fluorescence.

In summary, optical spectroscopy results indicate that TAMRA is not undergoing fluorescence quenching interactions with the TiO_2 surface, and that it is not involved in H-type stacking interactions with nearby TAMRA moieties, as well.

These findings suggest that TAMRA is positioned near the TiO_2 surface, allowing for efficient electron transfer, but at a sufficiently large distance to avoid overlapping between the chromophore and the semiconductor surface electronic states. This situation would promote efficient through-space electron transfer, inhibiting charge recombination or relaxation through energy dissipation.

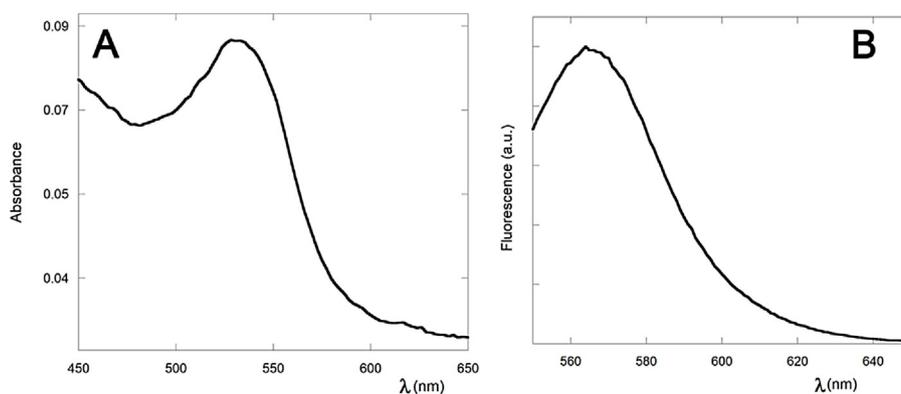


Fig. 4. UV-vis absorption (A) and fluorescence emission spectrum (B) of **T12** supported on TiO_2 .

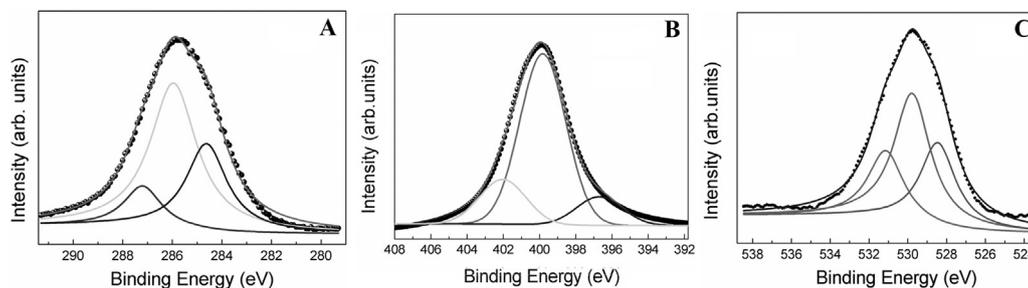


Fig. 5. X-ray photoelectron spectra of **T12** on TiO₂. A: C 1s; B: N 1s; C: O 1s.

It may be questioned about the contribution of through-bond pathways to the ET efficiency. From the calculations we obtained that the TAMRA distance from the surface ranges between 3.5 and 5 Å, while the through bond distance considering the first linkage through the anchoring nitrogen atoms of the Arg3 side-chain is 19.8 Å. The dependence of ET rate constant on the donor-acceptor distance can be described through the equation:

$$k_{ET} = k_0 e^{-\beta R_{DA}}$$

In a theoretical work, Shin et al. [32], showed that in the case of the polyproline type II conformation $\beta = 0.7 \text{ \AA}^{-1}$. Assuming $\beta = 1.1 \text{ \AA}^{-1}$ for through-space ET ($R_{DA} = 5 \text{ \AA}$), this means that through-space ET should be favored with respect to through-bond ($R_{DA} = 19.8 \text{ \AA}$) ET by a factor larger than $4 \cdot 10^3$.

X-ray Photoelectron Spectroscopy (XPS) measurements were carried out to characterize the **T12**·TiO₂ binding interaction. Extended XPS surveys of the TiO₂ surface carried out before and after peptide deposition reveal a marked attenuation of the Ti contribution coming from the substrate (Fig. S1, Supporting information), suggesting the formation of a thick **T12** layer. Furthermore, upon **T12** linkage, evidences of some oxidation processes can be obtained by the decrease of the Ti 2p_{3/2} lines in the range 455–470 eV, attributed to Ti ions in the lower oxidation states [33].

Fig. 5 shows the contributions from the C, N and O elements composing the peptide film and the associated curve-fittings after adsorption on TiO₂.

The C1s core level spectrum consists of three components (Fig. 5A), one located at 284.6 eV, assigned to carbon atoms involved in C–C/C–H bonds, a second one at 286 eV (C–OH groups), that is the main contribution to the signal, and a third signal at 287.2 eV that originates from the C=O groups.

In the N1s XPS spectrum three different N components can be observed (Fig. 5B): one at 402 eV, that can be assigned to H_x⁺N-species, a second one at 399.8 eV (H₂N-) [34], and a minor contribution at 396.7 eV, that most likely originates from Ti–N contacts formed during the film deposition [35], respectively.

The O1s spectrum (Fig. 5C) also features three components, one ascribable to the TiO₂ substrate (529.4 eV), the second one to

carbonyl O groups on the surface (532 eV) and the third one to the surface hydroxyl groups formed during the dissociative adsorption reaction (530.4 eV), respectively [36].

The results of the curve fitting procedure for all relevant XPS signals are summarized in Table 1.

3.3. Photoelectrochemical experiments

Photocurrent generation (PG) experiments were carried out on a TiO₂/FTO electrode modified by adsorption of a **T12** peptide film. Interestingly, a very high anodic current was generated upon illumination in the visible region of the **T12**/TiO₂/FTO modified electrode in the presence of the I[−]/I₃[−] redox pair in solution.

Fig. 6A shows repeated light-on/light-off cycles carried out for different excitation wavelengths.

When the peptide layer is irradiated in the range comprised between 470 and 600 nm, **T12** gives rise to an electron transfer process from its excited state to the conduction band of TiO₂ (Fig. 6B). After that, the I[−]/I₃[−] redox pair in solution transfers an electron to the oxidized TAMRA, producing a net, anodic electronic current.

As TiO₂ absorbs light only in the UV-range, the excitation band-gap being 3.2 eV, the strong photocurrent signal measured in the visible region can be safely ascribed to the TAMRA dye functionalizing the peptide. This is confirmed by the photocurrent action spectrum, i.e. the excitation wavelength dependence of the photocurrent response of the **T12**/TiO₂/FTO electrode, shown in Fig. 7. The latter strictly overlaps the absorption spectrum of **T12**, indicating that TAMRA is responsible for triggering the photoinduced electron transfer process.

The incident photon to current efficiency (IPCE) of the **T12**/TiO₂/FTO photoactive unit can be evaluated through the equation:

$$IPCE(\%) = 100 \cdot \frac{i(A/cm^2) \cdot 1240}{I(W/cm^2) \cdot \lambda}$$

where i is the photocurrent density (A/cm²), I is the incident power density (W/cm²) and λ is the incident photon-wavelength (nm). The IPCE value at $\lambda_{exc} = 550 \text{ nm}$ and under null bias potential is $0.55 \pm 0.05\%$. This promising result urged us to investigate the

Table 1
X-rays Photoelectron Spectroscopy energies of **T12** on TiO₂.

Element	Binding Energy (eV)	Signal emitters	Element	Binding Energy (eV)	Signal emitters
C 1s	284.6	C–C/C–H	O 1s	529.4	TiO ₂
	286.0	C–OH		530.4	Ti–OH
	287.2	C=O		532.0	C–O
N 1s	399.8	NH ₂	Ti 2p	459.0	TiO ₂
	402.0	NH ₃ ⁺		464.5	TiO ₂

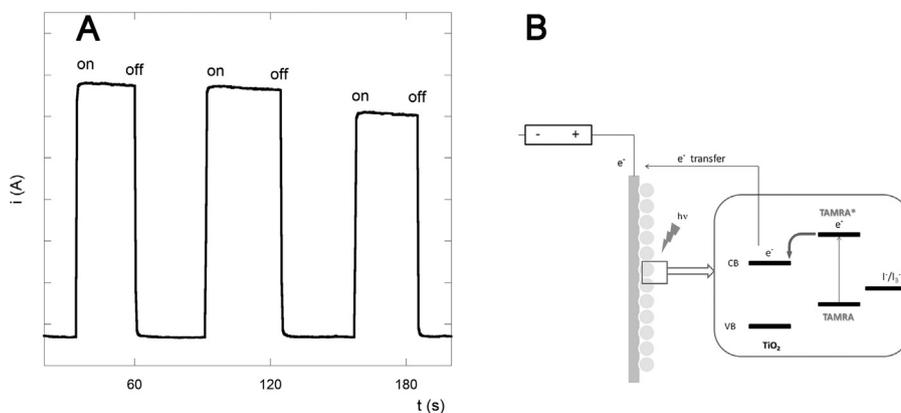


Fig. 6. Photocurrent generation experiments carried out on **T12**/TiO₂ immobilized on an FTO electrode. (A) Photocurrent steps at null bias potential ($\Delta V = 0$ V) under illumination at $\lambda = 550, 560$ and 570 nm (from left to right in the panel). (B) Schematic representation of the energy levels and electron transfer steps involved in the photocurrent process.

possibility of using **T12** for building a prototype Dye Sensitized Solar Cell (DSSC).

3.4. Construction of a **T12**/TiO₂ dye sensitized solar cell

Bio-inspired DSSC's using photosynthetic proteins for the design of new-concept devices for photoelectric conversion have been already realized [37–39]. Unfortunately the low stability of natural proteins once extracted from their biological environment, severely limits their use.

Peptides have been shown to feature unique electron transfer properties, in terms of superior ET efficiency over very long distances, ET asymmetry and robustness over many photoinduced ET cycles [40]. In this connection, we recently reported on PG experiments carried out using a gold electrode modified by a SAM

formed by a helical thiolated peptide [41]. At that time, we were able to show that a helical peptide scaffold suitably functionalized with an antenna chromophore could enhance light-induced electron-hole separation (*antenna effect*).

Following the same approach, we built a prototype DSSC using a TiO₂/FTO electrode coated with a **T12** film as a working electrode and an FTO conducting glass coated with platinum as a counter electrode. The latter was placed directly on the top of the working electrode, using a Surlyn film. PG experiments carried out with this cell showed high stability after several cycles of measurements and an efficiency of $0.65 \pm 0.05\%$ under simulated solar light illumination, definitely higher than the efficiencies measured in a DSSC using rhodamine as the photoactive element [42] or natural dye extracts [43,44].

In particular, the photovoltaic performance of Rhodamine B was already investigated by Choopun et al. using ZnO as the conductive substrate [45]. In that case, they obtained an efficiency of 0.18%, more than 3 times lower than that one measured by us for the **T12**/TiO₂ system.

This is most likely due to the increased energy gap between the LUMO energy of the dye and the conduction band of the metal oxide (Fig. 6B), enhancing the charge transfer rate constant and therefore, the photoconversion efficiency [46]. The operative parameters of the **T12**-based DSSC, reported as the average of the results obtained during four days-long experiments, are reported in Table 2, in comparison with the analogous DSSC using bare TiO₂ nanoparticles as a working electrode.

The voltages measured under open circuit conditions (V_{oc} 's) using a TiO₂/FTO electrode with or without the peptide layer, are almost the same. This finding indicates that the peptide layer does not affect the efficiency of charge recombination, contrasting with what found for vertically-arranged self-assembled monolayers formed by 3₁₀-helical peptide building blocks [41]. In that case, charge recombination is inhibited by the electrostatic field associated to the peptide helix dipole, that preferentially promotes ET from the C- to the N-terminus. As suggested by CD experiments and MM calculations, **T12** attains a polyproline type II conformation that was shown to contribute almost negligibly to the stabilization of charge-transfer states, due to the weak electrostatic field associated to peptide backbone attaining this specific conformation [32]. Besides that, in analogy with what we have recently found for a peptide film horizontally layered on a gold surface by two side-chain dithiol groups [14], the topology of the **T12**/TiO₂ integrated system, strongly favors trough-space ET, limiting the role of the peptide chain as a rigid scaffold holding

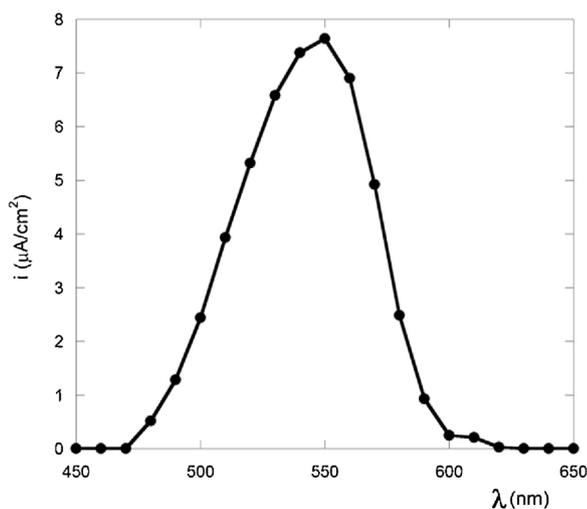


Fig. 7. Photocurrent action spectrum of the **T12**/TiO₂/FTO electrode, showing the excitation wavelength dependence of the photocurrent intensity.

Table 2
Operative parameters of the **T12** based DSSC.

Sample	V_{oc} (V)	J_{sc} (mA/cm ²)	Efficiency (%)	Fill Factor
T12 /TiO ₂	0.45 (± 0.05)	2.3 (± 0.1)	0.65 (± 0.05)	66%
TiO ₂	0.50 (± 0.1)	0.9 (± 0.1)	0.29 (± 0.1)	64%

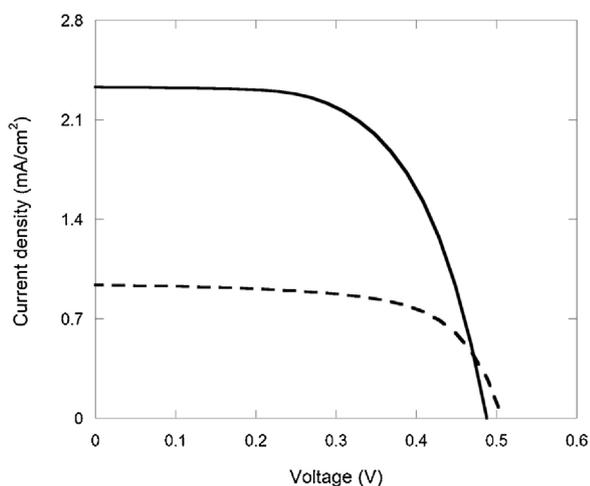


Fig. 8. Current density vs voltage response of the **T12**/TiO₂/FTO electrode (continuous line) and of the bare TiO₂/FTO electrode (dotted line).

the TAMRA chromophore in the most convenient position for efficient ET. On the other hand, the presence of several anchoring groups contributes to prevent aggregation, but limits the coverage of TiO₂, diminishing light harvest.

Interestingly, **T12**-based DSSC shows definitely higher short-circuit density currents (J_{sc}), i.e. the current that flows through the external circuit when the electrodes of the solar cell are short circuited. This parameter therefore represents the maximum current that can be delivered by a solar cell, and is directly related to the generation and collection of light-generated electron/hole carriers.

This finding is further emphasized by the data shown in Fig. 8, where the current-voltage (I/V) dependence, obtained with the **T12**/TiO₂/FTO thin layer cell under an air mass (AM) 1.5G solar simulator at 1 Sun lighting, was reported in comparison with the photoresponse of a bare TiO₂/FTO electrode.

From the data reported in Fig. 8, it can be easily seen that the current density at all applied voltages is definitely larger for the **T12**/TiO₂/FTO-coated electrode, the current-density limit at short circuit (J_{sc}) being 3-times larger than the one measured for the bare TiO₂/FTO electrode. This result emphasizes the role of the peptide scaffold as a rigid spacer, preventing charge-recombination for structural reasons, due to the electronic properties of the polyproline type II conformation attained by the peptide chain [27,28,32], and the topology of the **T12**/TiO₂ film, characterized by the presence of multiple side-chain linkers, that constrain the peptide chain to attain a relatively rigid location horizontally layered with respect to the TiO₂ surface.

4. Conclusions

The formation of a stable peptide film on TiO₂ and its ability to generate an electronic current under continuous electromagnetic excitation were clearly demonstrated. The peptide investigated comprised a core sequence (RKLPGA) well-known to be able to specifically bind TiO₂ by means of several side-chain groups, i.e. the arginine and lysine protonated amines and the aspartic carboxylate groups. The role of the peptide is to hold the TAMRA chromophore in the most convenient position for efficient ET and to modulate the electronic properties of the TiO₂ surface. In particular, the peptide layer, acting as a rigid scaffold, avoids that the photoexcited rhodamine chromophores would dissipate their

excitation energy by mixing to the manifold of the TiO₂ electronic states.

Our results indicate that peptide scaffolds can be considered promising materials for the development of molecular components to be used in DSSCs. We demonstrate that a **T12**-based DSSC shows an efficiency of 0.65%, which is awfully low if compared with the efficiency of silicon-based solar cells, but superior to previously reported studies on similar bio-inspired systems.

In any case, the evidence that photosensitizers embedded in a peptide matrix give rise to a current density definitely higher than bare TiO₂ materials, is, in our opinion, of relevant interest. This is, to the best of our knowledge, the first example of a peptide-based DSSC, where the peptide is directly linked to the TiO₂ surface.

The key advantages of using peptides in bioelectronics are their well-defined molecular structure, facile molecular design and arrangement of functional groups along the molecule, unique templating and self-organizing properties, and stability after several cycles of measurements.

Furthermore, at the end of life of the dye and/or of the peptide, the cell can be easily regenerated by reduction of the peptide-metal bond and incubation of the electrode in a peptide solution.

A specific result of this work is the possibility to finely tuning the electronic properties of the metal or semiconductor electrodes by covalent deposition of *vertically* or *horizontally* layered peptide films. In the former case, helically arranged peptide layers can assume an active role, favoring directional ET from the C- to the N-terminus, while in the latter is predominant the structural role of the peptide chain as a conformationally well-defined spacer.

We do hope that the results presented here will prompt new efforts to improve the efficiency of peptide-based DSSC, optimizing the parameters that determine the photocurrent generation efficiency, from the harvest of photoexcitation by a suitable molecular antenna to the cell current production.

Notes

The authors declare no competing financial interest.

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Appendix A. Supplementary data

Supplementary data associated with this article can be found, in the online version, at <http://dx.doi.org/10.1016/j.jphotochem.2017.07.027>.

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