Determination of Interproton Distances from NOESY Spectra in the Active Site of Paramagnetic Metalloenzymes: Cyanide-Inhibited Horseradish Peroxidase

Marco Sette,^{†,§} Jeffrey S. de Ropp,[‡] Griselda Hernandez,[†] and Gerd N. La Mar^{*,†}

Contribution from the Department of Chemistry and Nuclear Magnetic Resonance Facility, University of California, Davis, California 95616

Received January 11, 1993

Abstract: Two dimensional (2D) nuclear Overhauser effect (NOE) or NOESY experiments are performed on cyanideinhibited horseradish peroxidase in order to assess the prospects for obtaining quantitative interproton distances for the hyperfine shifted and paramagnetically relaxed active site signals in an intermediate sized (\sim 44 kDa) paramagnetic metalloenzyme. This protein represents an ideal test case for such experiments because a series of structurally defined proton pairs on the heme and axial His 170 have been previously assigned. The relaxation properties of hyperfine shifted signals relevant to the experimental setup of 2D experiments and interpretation of both 1D and 2D NOE data are also investigated. NOESY spectra as a function of mixing time show that quantitative rise curves can be obtained that clearly differentiate between primary and secondary NOEs even among the most strongly relaxed protons, but this requires very short mixing times in the range 0.5-3.0 ms. The sensitivity of the weak cross peaks at these short mixing times is improved by the relatively rapid pulse repetition rate and concommitant increase in the number of scans allowed by the rapid relaxation of active site protons. The paramagnetic relaxation influence, as well as the size of the protein, results in rise curves that are linear to only 1.5 ms for geminal protons and to only 3-5 ms for more weakly dipolar coupled proton pairs. However, the cross peak intensities in the linear region are shown to yield cross relaxation rates and internuclear distances for a series of assigned and orientationally invariant proton pairs that are in good agreement with their known distances. The patterns of NOESY rise curves are used both to determine the orientations of one propionate and both vinyl groups relative to the heme and to show that the axial His exhibits an orientation relative to its helix that is similar but not identical to that in crystallographically characterized cytochrome c peroxidase. Selective and nonselective 1D as well as 2D selective relaxation rate measurements for hyperfine shifted signals show that only selective or intrinsic relaxation rates can be used to optimize the setup of NOESY experiments and interpret 1D NOE data. The results of the study indicate that NOESY spectra can be expected to yield valuable quantitative structural information on the hyperfine shifted active site residues in a variety of cyanide-inhibited heme peroxidases.

Introduction

2D NMR can provide both the resonance assignments and the interproton distances that allow the determination of the solution molecular structure of diamagnetic biopolymers.^{1,2} Similar 2D NMR studies on analogous paramagnetic systems, particularly near the active site, were originally thought to be impossible, since the paramagnetic-induced relaxation results in substantial diminution of the nuclear Overhauser effect (NOE) and shortcircuits the development of coherence. Recent studies on a variety of paramagnetic metalloproteins, however, have shown that the major 2D experiments are surprisingly effective in detecting both scalar and dipolar connectivities.³ To date, however, virtually all studies have emphasized qualitative rather than quantitative aspects of the 2D data. Thus, both COSY and NOESY experiments have been utilized primarily to provide assignments, and few attempts has been made to extract distances from NOESY data on the active site of a paramagnetic protein. In the instances where interproton distances have been determined,4-9 the results have relied on the 1D steady-state or truncated NOE.

- On leave from the Department of Chemical Sciences and Technologies, University of Rome, "Tor Vergata".
 (1) Wüthrich, K. NMR of Proteins and Nucleic Acids; Wiley & Sons:
- New York, NY, 1986.
- (2) Robertson, A. D.; Markley, J. L. In *Biological Magnetic Resonance*; Berliner, L. J., Reuben, J., Eds.; Plenum Press: New York, 1990; Vol. 9, pp 155-176.
- (3) La Mar, G. N.; de Ropp, J. S. In Biological Magnetic Resonance;
 Berliner, L. J., Reuben, J., Eds.; Plenum Press: New York, in press; Vol. 12.
 (4) Thanabal, V.; de Ropp, J. S.; La Mar, G. N. J. Am. Chem. Soc. 1987, 109, 265-272.

The intensity (volume) of a NOESY cross peak between proton sets i $(n_i \text{ protons})$, and j $(n_j \text{ protons})$ with intrinsic relaxation rates ρ_i , ρ_j , is given by:¹⁰⁻¹³

$$a_{ij} = \left(\frac{n_i n_j}{n_i + n_j}\right) \frac{V_0 |\sigma_{ij}|}{2D} (1 - e^{-2D\tau_m}) (e^{-(R-D)\tau_m})$$
(1)

for which the initial term in τ_m represents an exponential buildup due to cross relaxation and the second term in τ_m represents a decay by a "leakage" rate, R - D, where $R = 1/2(\rho_i + \rho_j)$, $\Delta D = |\rho_i - \rho_j|$, $E = [(\Delta R)^2/4 + \sigma_{ij}^2]^{1/2}$, and $V_0 = V_{0i} + V_{0j}$ is the total magnetization (total intensity or volume of the diagonal peaks for *i*, *j* at $\tau_m = 0$). The cross relaxation rate σ_{ij} , in the slow motion limit, yields¹⁴ r_{ij} , *i.e.*:

- (5) Thanabal, V.; de Ropp, J. S.; La Mar, G. N. J. Am. Chem. Soc. 1987, 109, 7516-7525.
- (6) Banci, L.; Bencini, A.; Bertini, I.; Luchinat, C.; Piccioli, M. Inorg. Chem. 1990, 29, 4867-4873. (7) Scarrow, R. C.; Pyrz, J. W.; Que, L., Jr. J. Am. Chem. Soc. 1990, 112,
- 657-665. (8) Dugad, L. B.; La Mar, G. N.; Banci, L.; Bertini, I. Biochemistry 1990,
- 29, 2263-2271 (9) Cheng, H.; Grohmann, K.; Sweeney, W. J. Biol. Chem. 1990, 265, 12388-12392.
- (10) Macura, S.; Huang, Y.; Suter, D.; Ernst, R. R. J. Magn. Reson. 1981, 43. 259-281.
- (11) Ernst, R. R.; Bodenhausen, G.; Wokaun, A. Principles of Nuclear Magnetic Resonance in One and Two Dimensions; Clarendon Press: Oxford, UK, 1987.
- (12) Neuhaus, D.; Williamson, M. The Nuclear Overhauser Effect in Structural and Conformational Analysis; VCH Publisher: New York, NY, 1989.

(13) Yip, P. F. J. Magn. Reson. 1990, 90, 382-383.

(14) For convenience, we consider only the magnitude of the cross relaxation rate in 2D experiments; for HRP-CN, all σ are negative.

^{*} Address correspondence to this author.

[†] Department of Chemistry.

[†]NMR Facility

$$|\sigma_{ii}| = 0.1 \gamma^4 \hbar^2 r_{ii}^{-6} \tau_c$$
 (2)

At very short $\tau_m \ll (2D)^{-1}$, $(R - D)^{-1}$, the expression in (1) is simplified to:^{12,13}

$$\mathbf{a}_{ij}|_{\tau_{\mathrm{m}\to 0}} = \left(\frac{n_i n_j}{n_i + n_j}\right) \mathbf{V}_0 |\sigma_{ij}| \tau_{\mathrm{m}}$$
(3)

This initial linear slope of the NOESY cross peak buildup is the only quantitative method for extracting distances.

There exist numerous metalloproteins which exist solely in paramagnetic functional states that exhibit reasonably well resolved ¹H NMR spectra¹⁵⁻¹⁷ and for which the determination of interproton distances near the active site will provide valuable information on the detailed stereochemistry of functional residues. A particular class of these proteins are the heme peroxidases that exist in a high-spin ferric resting state which is readily inhibited by cyanide ligation to yield a low-spin ferric derivative. These proteins oxidize a variety of substrates at the expense of hydrogen peroxide and appear to rely on highly conserved distal catalytic residues to allow the facile formation of the activated complexes called compounds I and II.¹⁸⁻²⁰ The enzymes are large by 2D NMR standards, 34-150 kDa, and detailed crystal structures for only one member, cytochrome c peroxidase (CcP) have been reported.^{21,22} Numerous 1D and 2D ¹H NMR NOE studies on the hyperfine shifted resonances of cyanide complexes of 34-kDa CcP,^{23,24} ~44-kDa horseradish peroxidase^{4,5,25,26} (HRP), 42-kDa lignin peroxidase,^{27,28} 46-kDa manganese peroxidase,29 78-kDa lactoperoxidase,30 and 155-kDa myeloperoxidase³¹ have been carried out. Successful ¹H NMR studies have also been extended to genetically engineered variants of CcP and HRP.³²⁻³⁴ The varied proteins reflect a remarkably conserved electronic structure that results in similar nonselective T_1 s of 3-140 ms for coordinated heme α -substituents and histidine protons. However, while these enhanced relaxation rates for active site residues suggest that the decay of NOESY cross peaks will be rapid, the relatively large size of the proteins also guarantees reasonably rapid cross peak rise curves.³ Hence a heme peroxidase constitutes a test case for the realistic prospects for extracting distances from NOESY data for active site residues of a class of

- (15) La Mar, G. N., Horrocks, W. D., Jr., Holm, R. H., Eds. NMR of Paramagnetic Molecules; Academic Press: New York, 1973.
 (16) Satterlee, J. D. Annu. Rep. NMR Spectrosc. 1986, 17, 79-178.
- (17) Bertini, I.; Luchinat, C. NMR of Paramagnetic Molecules in Biological
- Systems; Benjamin/Cummings: Menlo Park, CA, 1986.
- (18) Dunford, H. B.; Stillman, J. S. Coord. Chem. Rev. 1976, 19, 187-251.
 (19) Dunford, H. B. In Peroxidases in Chemistry and Biology; Everse, J., Everse, K. E., Grisham, M. B., Eds.; CRC Press: Boca Raton, FL, 1991; Vol.
 2, pp 1-23.
 (20) Will a V. C. L. D.
- (20) Welinder, K. G. In Plant Peroxidases 1980-1990: Progress and Prospects in Biochemistry and Physiology; Gaspar, T., Penel, C., Greppin, H., Eds.; in press.
- (21) Poulos, T. L.; Kraut, J. J. Biol. Chem. 1980, 255, 8199-8205.
- (22) Finzel, B. C.; Poulos, T. L.; Kraut, J. J. Biol. Chem. 1984, 259, 13027-13036.
- (23) Satterlee, J. D.; Erman, J. E. Biochemistry 1991, 30, 4398-4405.
 (24) Satterlee, J. D.; Russell, D. J.; Erman, J. E. Biochemistry 1991, 30, 9072-9077.
- (25) Thanabal, V.; de Ropp, J. S.; La Mar, G. N. J. Am. Chem. Soc. 1988, 110, 3027-3035.
- (26) de Ropp, J. S.; Yu, L. P.; La Mar, G. N. J. Biomol. NMR 1991, 1, 175-190.
- (27) de Ropp, J. S.; La Mar, G. N.; Wariishi, H.; Gold, M. H. J. Biol. Chem. 1991, 266, 5001-5008.
- (28) Banci, L.; Bertini, I.; Turano, P.; Tien, M.; Kirk, T. K. Proc. Natl. Acad. Sci. U.S.A. 1991, 88, 6956–6960.
- (29) Banci, L.; Bertini, I.; Pease, E. A.; Tien, M.; Turano, P. Biochemistry 1992, 31, 10009-10017.
- (30) Thanabal, V.; La Mar, G. N. Biochemistry 1989, 28, 7038-7044.
 (31) Dugad, L. B.; La Mar, G. N.; Lee, H. C.; Ikeda-Saito, M.; Booth, K. S.; Caughey, W. S. J. Biol. Chem. 1990, 265, 7173-7179.
- (32) Satterlee, J. D.; Erman, J. E.; Mauro, J. M.; Kraut, J. Biochemistry 1990, 29, 8797–8804.
- (33) Banci, L.; Bertini, I.; C., Turano, P.; Ferrer, J. C.; Mauk, A. G. Inorg. Chem. 1991, 39, 4510-4516.
- (34) Veitch, N. C.; Williams, R. J. P.; Bray, R. C.; Burke, J. F.; Sanders, S. A.; Thorneley, R. N. F.; Smith, A. T. *Eur. J. Biochem.* **1992**, 207, 521–531.

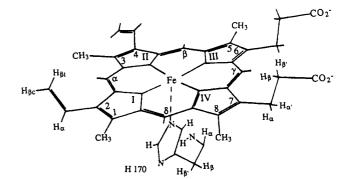


Figure 1. Structure and labeling scheme of the heme and axial His 170 as described in the crystal structure of $CcP^{21,22}$ The vinyl orientations indicated are those determined for HRP-CN.^{4,48}

paramagnetic proteins for which success should also allow much more quantitative structural comparison among genetic variants and/or point mutants.

Our reference protein selected for this study is the cyanide complex of horseradish peroxidase isozyme C (HRP–CN), a 308 amino acid glycoprotein with a $\sim 23\%$ carbohydrate content.^{18–20} In the absence of a crystal structure, ¹H NMR has served as the major tool for delineating the active site structure.^{4,5,25,26,34,35} The properties that make HRP-CN the ideal candidate for our study are that all of the resonances for the two bound endogenous ligands, the heme and axial His 170 (Figure 1), have been unambiguously identified,^{4,5,26,34} the protein is remarkably stable, and HRP–CN is available in large quantities. Hence the functionality of many groups in HRP–CN, including some of the interproton distances, are already known and serve as a test for our measured values. The heme and axial His resonances, moreover, provide some of the broadest and most rapidly relaxed protons in the protein and therefore serve as the most serious test for obtaining distances.

2D NOESY studies of paramagnetic metalloproteins have infrequently utilized more than one mixing time,³ and those mixing times were generally selected on the basis of the nonselective T_1 of resonances of interest since the maximum cross peak intensity occurs for $\tau_{\rm m} \sim$ spin lattice relaxation time.^{11,12} In only a few cases have more than one τ_m been used, and no detailed rise curve has been reported. As an extension of our investigation of the prospects for deriving distances from such rise curves, we also explore the basis for selecting the mixing time for a single or a limited series of NOESY spectra designed to optimize intensity of selected cross peaks. The particular questions we seek to answer in this study are the following. (1) Can cross peak rise curves for a significant number of dipolar contacts for a paramagnetically relaxed proton be quantitated? (2) How serious are secondary NOEs and spin diffusion for mixing times commonly used (20-50 ms) to collect NOESY maps for low-spin ferric heme peroxidases? (3) For primary NOEs, is it practical to quantitatively measure cross peak intensity for the short mixing times over which the rise curves are linear? (4) Do the linear portions of rise curves yield relative cross relaxation rates and distances consistent with known geometry? (5) Can absolute distances be obtained? (6) Which practically measurable relaxation property serves as the best basis for selecting mixing times for optimal cross peak sensitivity? And lastly, (7) does the geometric information on the heme and axial ligand provide an interpretation for HRP-CN on the basis of the crystal coordinates^{21,22} for CcP-CN?

Experimental Section

Sample Preparation. Horseradish peroxidase (HRP) was purchased from Boehringer-Mannheim as a lyophilized salt-free powder; the protein is 98% isozyme C with an RZ (Reinheitszahl, A_{403}/A_{280}) value of 3.1.

⁽³⁵⁾ de Ropp, J. S.; La Mar, G. N.; Smith, K. M.; Langry, K. C. J. Am. Chem. Soc. 1984, 106, 4438-4444.

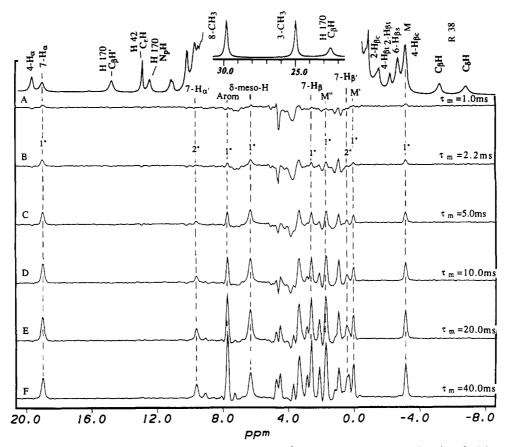


Figure 2. Slices through the 8-CH₃ diagonal parallel to the f_2 axis for HRP-CN in ²H₂O at 35 °C, pH 7.0 as a function of mixing time: (A) 1.0 ms, (B) 2.2 ms, (C) 5.0 ms, (D) 10 ms, (E) 20 ms, and (F) 40 ms. The identical vertical scaling is used for all slices. The cross peaks of interest are labeled by their identity if known; otherwise the symbols 1°, 2° are used simply to designate whether the rise curve has its intercept at $\tau_m = 0$ (primary NOE) or at $\tau_m > 0$ (secondary NOE). Note the excellent sensitivity for cross peaks to resonances labeled Arom and M, M', M" are to residues previously proposed 5,26 as Phe and Ile, respectively.

Protein samples were prepared to be 3 mM in 99.9% ${}^{2}H_{2}O$. A solution of potassium cyanide (Mallinckrodt) prepared in the same solvent was added to generate HRP-CN. The solution pH was adjusted to 6.8 with dilute ${}^{2}HCl$ or NaO ${}^{2}H$; the pH value was not corrected for the isotope effect. The majority of previous 2D and 1D NOE data^{4,5,25,26} on HRP-CN have been collected at temperatures well above 40 °C because of the much narrower lines and superior spectral resolution. However, since few other heme peroxidases exhibit the remarkable thermal stability of HRP, the present exploration of 2D NMR methods on HRP-CN is restricted to a more representative temperature, *i.e.*, 35 °C, such that the results from the study can be extrapolated to other systems.

NMR Data Collection. Data were obtained on a GE-NMR Ω -500 spectrometer operating at 11.75 T (500 MHz for ¹H). NOESY data were collected in a 10-mm ¹H probe from Cryomagnet Systems, Inc.; the T1 data were collected in a 5-mm GE-NMR ¹H probe. All NOESY data sets^{10,36} were obtained under identical conditions: 352 scans were collected in 512 blocks (t_1 acquisition time 8 ms) using the hypercomplex method with a repetition rate of 5 scans/s for a total acquisition time of 10 h. The recycle time was minimized to allow the maximum number of scans per unit time to optimize sensitivity.³ The value of 200 ms is consistent with the convention of using a recycle time of the order 1.5 T_1 , since the longest T_1 s for the heme α -substituents and axial His resonances are known to be <140 ms.^{5,26} A total of 1024 complex points in t_2 were collected over a 31.25-kHz bandwidth for a 33-ms acquisition time in t_2 . The residual solvent line was presaturated with a low-power transmitter pulse. The basic NOESY sequence was utilized without any composite pulses in order to maximize the bandwidth of excitation.³ To maintain a constant pulse recycle time, the predelay was varied to compensate for any changes in the mixing time.

Nonselective T_1 , $(\rho^{nsel})^{-1}$, data were obtained by the basic $180^\circ - \tau - 90^\circ$ pulse sequence using 8192 complex points over a 31.25-kHz bandwidth. Selective T_1 s, $(\rho^{nsel})^{-1}$, were obtained by the saturation recovery method,

utilizing a low-power 30-ms selective saturation of the resonance of interest by the decoupler. The nonselective, ρ^{nsel} , and selective, ρ^{sel} , relaxation rates were obtained from the initial slope³⁷ of the semilogarithmic plot of the fractional deviation of the z-magnetization from equilibrium versus the relaxation delay time τ . Steady-state NOE measurements were made by saturating the resonance of interest with a 100-ms selective decoupler pulse. Data were acquired by interleaving a block of scans with saturation on-resonance with an equal block of scans off-resonance. Difference spectra and reference spectra were plotted to the same scale and the NOE determined from:¹²

$$\eta_i\{j\} = \frac{I - I_0}{J_0} = \frac{\sigma_{ij}}{\rho_i} \tag{4}$$

where j is the resonance saturated, with intensity J_0 prior to saturation, which yields an NOE to peak *i*, with intensity I and I_0 with and without complete saturation of *j*.

NMR Data Processing. Raw 2D data sets were transferred to an IRIS 4D/35 and processed using the Hare Research software package Felix (version 1.1). 45°-shifted sine-bell-squared apodization was applied in t_2 over 1024 points, and 30°-shifted sine-bell-squared apodization was employed in t_1 over 256 points. The t_2 data were phase-corrected and base line straightened, and the t_1 data were zero-filled twice (for a final data matrix of 1024 × 1024 points) and, after Fourier transformation, phase-corrected with a small zero-order phase correction. Cross peak heights for rise curves (Figures 3–6) were obtained by plotting f_2 slices through the diagonal peak of interest and measuring cross peak heights in centimeters. All slices were plotted at the same arbitrary but uniform scale; uncertainties in cross peak heights are ± 0.1 units. Cross peak intensities between methyls and single protons were scaled by dividing¹³ by 3. Cross peak and diagonal peak volumes were obtained using the appropriate Felix subroutine. R_t values, the selective T_1^{-1} by diagonal

⁽³⁶⁾ States, D. J.; Haberkorn, R. A.; Ruben, D. J. J. Magn. Reson. 1982, 48, 286-292.

⁽³⁷⁾ Granot, J. J. Magn. Reson. 1982, 49, 257-270.

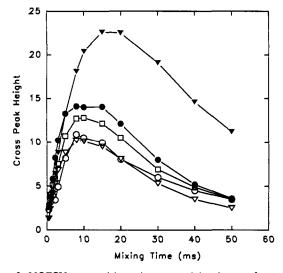


Figure 3. NOESY cross peak intensity versus mixing time, τ_m , for strongly coupled (geminal) proton pairs $7H_{\alpha} \rightarrow 7H_{\alpha'}$ (∇), $4H_{\beta c} \rightarrow 4H_{\beta t}$ (∇), His 170 $C_{\beta}H \rightarrow C_{\beta}H'(\Box)$, Arg 38 $C_{\beta}H \rightarrow C_{\beta}H'(\bullet)$, and Arg 38 $C_{\delta}H \rightarrow C_{\delta}H'$ (O). The vertical scaling is identical for all data. The data are for the f_2 slices through the diagonal of the better resolved proton of the pair.

decay,^{10,38} were determined from the slope of the plots of diagonal peak volume vs mixing time. The cross relaxation rate was determined by rearranging eq 3 to eq 5 and using the cross peak volume at the largest $\tau_{\rm m}$ value for which the cross peak rise curve was judged strictly linear; this $\tau_{\rm m}$ value is designated $\tau_{\rm m}^*$.

$$|\sigma_{ij}| = \frac{a_{ij}(\tau_{\rm m}^{*})}{q(V_{0i}/n_i)\tau_{\rm m}^{*}}$$
(5)

where q = 1 for a pair of protons and q = 3 for a proton interacting with a methyl group; V_{0i} is the diagonal volume at $\tau_m = 0$ for peak *i*. For geminal protons, τ_m^* values as short as 1.5 ms were used, while for more weakly coupled proton pairs, τ_m^* values up to 3 ms were used. The use of τ_m^* rather than a linear fit of a_{ij} values is dictated by the fractionally larger uncertainties of a_{ii} for $\tau_{\rm m} < \tau_{\rm m}^*$.

Results

Rise Curves. Representative f_2 slices of the NOESY map are illustrated in Figure 2 for the 8-CH₃ diagonal for mixing times of 1, 2.2, 5, 10, 20, and 40 ms; the reference trace for HRP-CN is shown along the top of Figure 2. The excellent sensitivity achieved is apparent by the clear detection of several cross peaks to weakly coupled protons with resolved signals even for $\tau_{\rm m} = 1$ ms. Several of the signals of interest in this report are labeled by their known assignments,^{4,5,26} and the symbols 1° and 2° are given to designate whether the signal appears to rise monotonically from $\tau_m = 0$ (primary NOEs) or develops only after some delay at $\tau_m > 0$ (secondary NOEs). The great wealth of information in this figure is made obvious by the detection of over a dozen cross peaks for the heme methyl, of which the majority, but not all, appear to be primary. Peaks in the unresolved 10-0 ppm region are of no interest in this report unless they are known to arise^{4,26} from the heme or axial His 170. Essentially all the significant primary NOEs for these weakly coupled protons are identified and can be quantitated for τ_m as short as 3 ms. Moreover, clearly secondary NOEs that develop only in the 3-5 ms window, such as those marked $7H_{a'}$ and $7H_{b'}$ in Figure 2, can be identified (see below). However, while it is obvious that there is sufficient sensitivity to detect and quantify cross peak intensity at short mixing times, it is also clear from Figure 2 that there are limitations to identifying and quantifying cross peak intensity at short mixing times in the window of the intense diamagnetic

envelope, 0–8 ppm, where t_1 ridges³⁹ and other artifacts tend to obscure the cross peaks when they are weak (*i.e.*, $\tau_{\rm m} \leq 2.2$ ms). At large $\tau_{\rm m}$, the artifacts make only minor contributions to the slices. Nevertheless, it is apparent that cross peak intensity is more difficult to quantify for peaks in the diamagnetic window than for resolved resonances (see below).

More impressive is the detectability of cross peaks involving strongly dipolar coupled proton pairs such as geminal methylene groups, for which the cross peak intensity at $\tau_m = 0.5$ ms (not shown) exceeds that for most of the resolved peaks at $\tau_m = 5.0$ ms for the weakly coupled protons in Figure 2. The intensity profiles (in arbitrary units, but the same units for all graphs in this report) as a function of $\tau_{\rm m}$ for cross peaks for expected geminal proton pairs, the propionate 7-H_as, 4-vinyl H_bs, axial His 170 C_{β} Hs, and Arg 38 C_{β} Hs and C_{δ} Hs, are illustrated in Figure 3. The cross peaks are designated $i \rightarrow j$ when the slice in the f_2 dimension through the diagonal of peak i is analyzed. Essentially the same rise curves are obtained for the cross peak symmetric about the diagonal (not shown). The curve for the unresolved propionate $7H_{\beta}s$ does not yield useful points⁴⁰ for $\tau_m \leq 2.2$ ms, but the steep increase in intensity that characterizes a geminal proton pair, however, is readily recognized in spite of the absence of the data at short τ_m (not shown). A noteworthy observation is that for most of the geminal proton pairs in Figure 3 the rise curves reach a maximum at $\tau_{\rm m} \sim 10$ ms, which is at a significantly shorter time than that expected from nonselective T_1 for any of the resonances of interest (see below). The maximum cross peak intensity at $\tau_{\rm m}$ ~20 ms for the 4H_bs is similarly at a much shorter time than the nonselective T_1 s. Nonselective T_1 s at 50 and 55 °C have been previously reported;5,26 the presently determined values for the resolved resonances at 35 °C are listed in Table I. The consequence of the cross peak maxima occurring at such short τ_m is that the rise curves for geminal protons in the active site are strictly linear to τ_m only ~1.5 ms.

The remaining cross peak intensity profiles with τ_m are considered in terms of the substituents on and flanking each of the pyrroles that yield resolved resonances, as given in Figures 4 and 5, and for the axial His 170 (Figure 1), as shown in Figure All rise curves represent monotonic increases to a maximum, and then monotonic decreases, except for those involving vicinal protons for the vinyl groups (see below). Figure 4A presents the rise curves for pyrrole II with the resolved 3-CH₃ and 4-vinyl signals. Within the vinyl group, the $4H_{\beta c} \rightarrow 4H_{\beta t}$ geminal rise curve has been considered above. The $4H_{\alpha} \rightarrow 4H_{\beta c}$ and $4H_{\alpha} \rightarrow 4H_{\beta t}$ rise curves exhibit a behavior different from that of other resonances considered; the intensity at short $\tau_{\rm m}$ oscillates with $\tau_{\rm m}$ (Figure 4A). A similar behavior is observed for $2H_{\alpha} \rightarrow 2H_{\beta c}$ and $2H_{\alpha} \rightarrow 2H_{\beta t}$ (not shown). This oscillatory behavior is due to zeroquantum coherence via the scalar interaction and reflects the fact that these proton pairs have strong scalar coupling relative to their dipolar coupling.^{10,11} Such oscillations have been characterized in diamagnetic proteins for proton pairs which show strong scalar and weak dipolar coupling.⁴¹ This is the first case for the detection of such oscillations in a paramagnetic compound. This oscillation precludes the detection of the initial linear proton of the rise curve. The $4H_{\alpha} \rightarrow 4H_{\beta t}$ rise curve should reflect a secondary NOE even with the oscillation; the observed curve is consistent with this expectation. This scalar modulation can be

⁽³⁸⁾ Mirau, P. A.; Bovey, F. A. J. Am. Chem. Soc. 1986, 108, 5130-5134.

⁽³⁹⁾ Mehlkopf, A. F.; Korbee, D.; Tiggelman, T. A.; Freeman, R. J. Magn. Reson. 1984, 58, 315-323. Turner, C. J.; Patt, S. L. J. Magn. Reson. 1989, 85, 492-505. Derome, A. E.; Williamson, M. P. J. Magn. Reson. 1990, 88, 177-185.

⁽⁴⁰⁾ Cross peaks to the diamagnetic region frequently suffered from distorted intensities due to t_1 ridges and other artifacts. Cross peaks a_{ij} , where resonance i was in the diamagnetic region and resonance j shifted outside, resonance i values in the damagnetic region and resonance j sinice oblishe, were quantified by using the a_{ij} volume at the f_1 frequency of i, placing the cross peak outside the region of most artifacts. (41) Kumar, A.; Wagner, G.; Ernst, R. R.; Wüthrich, K. J. Am. Chem. Soc. 1981, 103, 3654–3658.

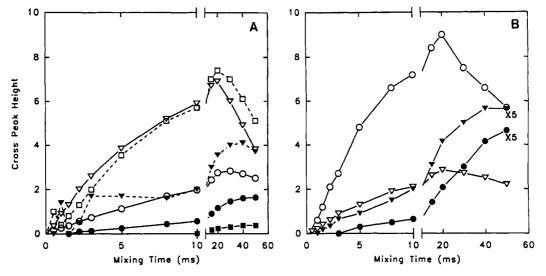


Figure 4. NOESY cross peak buildup curves as a function of mixing time, τ_m , for resonances of pyrroles I and II of the heme of HRP-CN in ²H₂O at 35 °C, pH 7.0. The f_2 slices were used in each case, and the slice is identified by the convention: diagonal \rightarrow cross peak; the same scaling is used, but peaks involving methyls¹³ are divided by 3. Panel A presents data for pyrrole II: $3-CH_3 \rightarrow 4H_{\beta t}$ (O), $3CH_3 \rightarrow 4H_{\beta c}$ (\oplus), $4H_{\alpha} \rightarrow 3CH_3$ (\blacksquare), $4H_{\alpha} \rightarrow 4H_{\beta t}$ (\square). Note the oscillation of the intensity (shown by dashed lines) for $4H_{\alpha} \rightarrow 4H_{\beta t}$ and $4H_{\alpha} \rightarrow 4H_{\beta t}$ at short mixing times which results from zero-quantum coherence.^{10,41} Note that both the $4H_{\alpha} \rightarrow 3CH_3$ (\blacksquare) and $3CH_3 \rightarrow 4H_{\beta c}$ (\square) NOEs are secondary. Panel B presents data for the pyrrole I/II junction: $3CH_3 \rightarrow 2H_{\beta t}$ (\triangledown), $3CH_3 \rightarrow 2H_{\beta c}$ (\square) (both with vertical expansion $\times 5$ compared to other plots), α -meso-H $\rightarrow 3CH_3$ (\triangledown), and α -meso-H $\rightarrow 2H_{\beta t}$ (O). Primary NOEs are designated by open markers and secondary NOEs by closed markers.

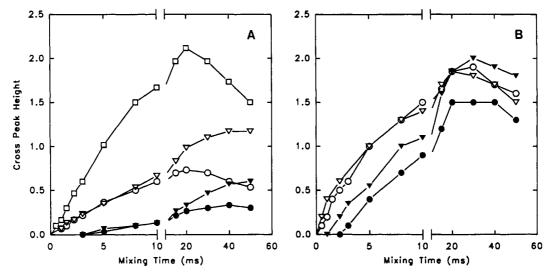


Figure 5. NOESY cross peak intensity (in arbitrary units) as a function of mixing time, τ_{m} , for pyrrole IV of HRP-CN in ²H₂O at 35 °C, pH 7.0. Panel A, cross peaks involving the 8CH₃ group: δ -meso-H \rightarrow 8CH₃ (\Box), 7H_{$\alpha}<math>\rightarrow$ 8CH₃ (O), 7H_{$\alpha'}<math>\rightarrow$ 8CH₃ (\bullet), 7H_{$\beta'}<math>\rightarrow$ 8CH₃ (∇), and 7H_{$\beta'}<math>\rightarrow$ 8CH₃ (∇), and 7H_{$\beta'}<math>\rightarrow$ 8CH₃ (∇). Panel B, vicinal cross peaks within the 7-propionate group: 7H_{$\beta'}<math>\rightarrow$ 7H_{$\alpha'} (<math>O$), 7H_{$\beta'}<math>\rightarrow$ 7H_{$\alpha'} (<math>\bullet$), and 7H_{$\beta''}<math>\rightarrow$ 7H_{α''} (∇). Note that in both panels the open markers reflect primary NOEs while the closed markers show secondary NOEs.</sub></sub></sub></sub></sub></sub></sub></sub></sub></sub>

removed by varying the mixing time randomly by a small increment;¹⁰ such experiments, however, were not attempted on this system.

The 3-CH₃ \rightarrow 4H_{β t} and β -meso-H \rightarrow 4H_{α} rise curves in Figure 4A reflect primary NOEs with a significant linear portion (to $\tau_{\rm m}$ = 3 ms). While $4H_{\alpha} \rightarrow 3$ -CH₃ and 3-CH₃ $\rightarrow 4H_{\beta c}$ cross peaks are observed at long $\tau_m > 5$ ms, the rise curves in Figure 4 clearly extrapolate to 0 for $\tau_m > 0$, dictating that the NOEs are secondary. The α -meso-H \rightarrow 3-CH₃ rise curve shown in Figure 4B is primary, as expected, and yields a significant initial linear portion to 5 ms. However, 3-CH₃ yields an intense cross peak not only to α -meso-H but also to a proton which resonates very close to α -meso-H^{4,5,26} (not shown). The fact that both rise curves involving meso-H reach maxima at shorter $\tau_{\rm m}$ (~20 ms) than those for the pyrrole substituents is likely due to the expected more rapid paramagnetic relaxation of meso-Hs. Previous 1D NOE studies had reported⁴ a dipolar contact between the 3-CH₃ and 2-vinyl H_{θ}s. The rise curves for 3-CH₃ \rightarrow 2-H_{β t} and 3-CH₃ \rightarrow 2-H_{β c} in Figure 4B, however, suggest that these NOEs are secondary, since the

intercepts are at $\tau_m \sim 1$ and 3 ms, respectively. This conclusion is supported by the observation of a much steeper rise curve for a primary NOE between $2H_{\beta t}$ and α -meso-H and between α -meso-H and 3-CH₃, indicating that the weak 3-CH₃ $\rightarrow 2H_{\beta c}$ cross peak at $\tau_m > 1$ ms arises because of the indirect path $2H_{\beta t}$ $\rightarrow \alpha$ -meso-H \rightarrow 3-CH₃ (see Discussion).

The cross peak rise curves for the pyrrole IV resonances are given in Figure 5. The δ -meso-H \rightarrow 8-CH₃ rise curve is primary, as expected, as is the rise curve for the 7H_{α} \rightarrow 8-CH₃ (Figure 5A). In contrast, the rise curve 7H_{$\alpha'}<math>\rightarrow$ 8-CH₃ clearly reflects a secondary NOE; hence 8-CH₃ is closer to 7H_{α} than 7H_{α'}. Similarly, the 7H_{$\beta'}<math>\rightarrow$ 8-CH₃ rise curve reflects a primary NOE, while that for 7H_{$\beta'}<math>\rightarrow$ 8-CH₃ indicates a secondary NOE; hence 8-CH₃ is closer to 7H_{β} than 7H_{β'}. The 7H_{α}s and 7H_{β}s each yielded the expected steep rise curves for geminal protons (7H_{α} shown in Figure 3). Further information on the propionate orientation can be gleaned from the propionate vicinal dipolar contacts shown in Figure 5B; primary NOEs are observed for the</sub></sub></sub>

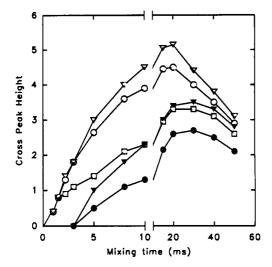


Figure 6. NOESY cross peak intensity (in arbitrary units) as a function of mixing time, τ_m , for vicinal protons of the axial His 170 of HRP-CN in ²H₂O at 35 °C, pH 7.0; C_{\beta}H \rightarrow C_{\alpha}H (\odot), C_{\beta}H \rightarrow NpH (\odot), C_{\beta}H' \rightarrow C_{\alpha}H (\checkmark), C_{\beta}H' \rightarrow C_{\alpha}H (\checkmark), C_{\beta}H' \rightarrow NpH (\heartsuit), and C_{\alpha}H \rightarrow NpH (\square). Note that open markers reflect primary NOEs while closed markers represent secondary NOEs.

Table I. ¹H NMR Spectral Parameters for HRP-CN^{a,b}

| peak | shift ^c | $\rho^{\text{nsel } d}$ | ρ ^{sel e} | R⁄ | $\Delta \rho_i^{g}$ | $\Delta \rho_i'$ | $ ho^{ m sel}/ ho^{ m nsel}$ | | | |
|--------------------------|--------------------|-------------------------|--------------------|-----|---------------------|------------------|------------------------------|--|--|--|
| 2-H _{βc} | -1.53 | i | | | | | | | | |
| 2-H _{st} | -2.62 | i | | | | | | | | |
| 3-CH₃ | 24.96 | 7.3 | 23 | 26 | 16 | 19 | 3.2 | | | |
| 4-H _a | 19.67 | 5.8 | 48 | 50 | 42 | 44 | 8.3 | | | |
| 4-H _{βc} | -3.09 | i | | | | | | | | |
| 4-H _{st} | -2.14 | i | | | | | | | | |
| 6-H ₈ | -2.66 | i | | | | | | | | |
| $7 - H_{\alpha}$ | 19.03 | 6.4 | 67 | 87 | 61 | 81 | 10 | | | |
| 7-H _a ' | 9.71 | i | | | | | | | | |
| 8-CH3 | 29.78 | 9.2 | 29 | 29 | 20 | 20 | 3.2 | | | |
| His 170 C _β H | I 22.48 | 18 | 74 | 99 | 56 | 81 | 4.1 | | | |
| His 170 C _β H | I' 14.83 | 15 | 56 | 95 | 41 | 80 | 3.7 | | | |
| His 170 N _p H | I 12.52 | 13 | 43 | 50 | 30 | 37 | 3.3 | | | |
| His 170 C_{α} H | I 9.86 | i | | | | | | | | |
| His 42 C,H | 12.97 | 6.3 | 13 | 16 | 7 | 10 | 2.1 | | | |
| Arg 38 C _s H | -5.23 | 9.8 | 69 | 107 | 59 | 97 | 7.0 | | | |
| Arg 38 C _s H | -6.82 | 30 | 90 | 94 | 60 | 64 | 3.0 | | | |

^a 3 mM HRP in ²H₂O at 35 °C, pH 7.0. ^b The shifts at 35 °C for nonresolved heme signals, in ppm, are 1-CH₃, 2.68; 2-H_a, 5.08; α -meso-H, 1.17; β -meso-H, 6.93; 5-CH₃, 6.45; γ -meso-H, 4.78; 7H_β', 0.52; 7H_β, 2.68; δ -meso-H, 6.44. ^c Chemical shift, in ppm from DSS; uncertainty ±0.01 ppm. ^d Nonselective relaxation rate, $(1/T_1^{usel})$, in s⁻¹, obtained from the initial slope from a nonselective inversion-recovery experiment; uncertainty ±10%. ^e Selective relaxation rate, $(1/T_1^{usel})$, in s⁻¹, obtained from a selective saturation-recovery experiment; uncertainty ±10%. ^f Decay constant, in s⁻¹, for the diagonal intensity in NOESY spectra as a function of mixing time; uncertainty $\oplus 10\%$. ^g $\Delta \rho_i = \rho_i^{sel} - \rho_i^{nsel}$. ^h $\Delta \rho_i'$

pairs $7H_{\beta} \rightarrow 7H_{\alpha}$ and $7H_{\beta'} \rightarrow 7H_{\alpha'}$, while the pairs $7H_{\beta} \rightarrow 7H_{\alpha}$, $7H_{\beta} \rightarrow 7H_{\alpha'}$ exhibit secondary NOEs.

The cross peak rise curves for the vicinal protons of the axial His 170 resonances (excluding the imidazole ring, whose resonances are ~400-Hz broad and whose dipolar connectivities can be detected only by 1D NOEs)⁵ are shown in Figure 6; the geminal $C_{\beta}H \rightarrow C_{\beta}H'$ rise curve has been presented in Figure 3. The C_{β} Hs each give only one other primary NOE, $C_{\beta}H \rightarrow C_{\alpha}H$ and $C_{\beta}H' \rightarrow N_{P}H$, with the large $C_{\beta}H \rightarrow N_{P}H$ and $C_{\beta}H' \rightarrow C_{\alpha}H$ NOEs at long τ_{m} clearly secondary. The rise curve for $C_{\alpha}H \rightarrow N_{P}H$ is harder to quantitate because both resonances are in or close to the intense diamagnetic envelope.⁴⁰ However, the intercept appears to occur at $\tau_{m} = 0$, and hence the large NOE is judged primary. Each of the primary NOEs yields initial linear portions for the rise curves only to ~2.2 ms for distance estimation (see below).

Distance Determination. The determination of σ_{ij} (eq 5) requires the diagonal volume at $\tau_m = 0$, V_{0i} , and the cross peak volumes at a $\tau_m \neq 0$ value which is clearly on the strictly linear portion of the rise curves in Figures 3-6. The normalized diagonal volumes of $\tau_{\rm m} = 0$, V_{0i}/n_i , (which must correlate with proton intensity in a nonsaturated trace) for the nine clearly resolved resonances for which this determination could be made yield a range 35.4-48.7 in arbitrary units, for a mean value $V_{0i}/n_i = 42.1$ \pm 6.7, *i.e.*, an uncertainty \pm 15%. This value was used in all subsequent calculations for the diagonal of both resolved and unresolved resonances. The volumes of the cross peaks, $a_{ii}(\tau_m^*)$, at the longest $\tau_{\rm m}$ for which the rise curve is strictly linear (at $\tau_{\rm m}^{*}$), for resonaces of interest are listed in Table II, along with the τ_m^* values. The $a_{ij}(\tau_m^*)$ values for a proton pair are similarly judged reliable⁴² to $\pm 15\%$, which results in σ_{ij} values with uncertainties $\pm 30\%$ and r_{ij} values reliable to $\pm 5\%$. The first seven entries in Table II are for proton sets for which r_{ii} values are essentially independent of protein constraints and hence can be considered known. They serve as calibration for the ability to obtain distances. For geminal proton pairs,43 the maximum in the rise curve near $\tau_{\rm m} \sim 10$ ms dictates $\tau_{\rm m}^*$ values ≤ 1.5 ms to insure linearity. The resulting $a_{ij}(\tau_m^*)$ values are of the order .05–0.1 V_{0i}/n_i (Table II). For the more weakly coupled proton pairs, the rise curves reached maximum at larger $\tau_{\rm m}$, and hence longer $\tau_{\rm m}^*$ (~2.2-5 ms) could be used. The σ_{ij} s resulting from eq 5 are listed in Table II. For the known geminal proton pairs, the σ_{ij} ranges from 38.3 to 58.6 s⁻¹ for an average 49.0 \pm 10.5. We use this average as reference for the relative σ_{ii} values for the proton pairs in Table II and, since $r_{ij} \propto \sigma_{ij}^{-1/6}$, obtain relative r_{ij} values for all proton pairs. By setting $r_{ij} = 1.77$ Å for the average geminal $\sigma_{ij} = 49.0 \text{ s}^{-1}$, the relative r_{ij} leads to estimates of r_{ij} for all proton pairs, as listed in Table II. For the likely rigid four methylene groups, this yields r_{ii} in the range 1.72–1.85 Å.

The rise curves for the fixed geometry δ -meso-H \rightarrow 8-CH₃ (Figure 5A) and α -meso-H \rightarrow 3-CH₃ (Figure 4B) exhibit very similar slopes, indicating that they reflect very similar σ_{ii} , r_{ii} , as is expected; the 5-CH₃ and β -meso-H resonate too close to each other²⁶ under the diamagnetic envelope to allow the characterization of a rise curve. Since 3-CH₃ yields NOEs to both the α -meso-H peak (not shown) and a partially overlapping resonance,^{4,5} a quantitative σ_{ij} was obtainable only from δ -meso-H \rightarrow 8-CH₃. The observed $|\sigma_{ij}| \sim 3.2 \text{ s}^{-1}$, which yields $r_{ij} \sim 2.8$ Å, is very close to that expected for this fixed geometry, 2.85 Å. Thus the "calibration" data for the heme and known geminal protons dictate that relative distance can be quantitatively determined using the very short linear portions of the rise curves. The relative cross relaxation rates and distances for several pairs of strongly relaxed protons⁴² whose geometry depends on the orientation of side chains are obtained from the linear portion of the rise curves and the values listed in Table II. For the axial His 170, distances 2.1-2.2 Å are obtained for C_dH':N_PH and $C_{\beta}H:C_{\alpha}H$. The $r(4H_{\alpha};\beta$ -meso-H) is estimated at 2.2 Å, as shown in Table II.

The estimated σ_{ij} for the three methylene groups that can reasonably be expected to be immobile within the protein matrix, the His C_βH:C_βH', propionate 7H_α:7H_α', and Arg 38 C_βH:C_βH', is 49 ± 10 s⁻¹, and using $r_{ij} = 1.77$ Å translates to a correlation time $\tau_c = 27 \pm 5$ ns by eq 2. This value compares reasonably

⁽⁴²⁾ Each dipolar set of protons i,j gives rise to two NOESY cross peaks which, under sufficiently slow pulse recycle times, are equal in intensity. However, if i,j have different T_1 s, the rapid pulsing conditions employed here can lead to suppression of the cross peak at the f_1 frequency of the slower relaxing resonance (Williamson, M. P.; Neuhaus, D. J. Magn. Reson. 1987, 72, 369–375). The difference in cross peak intensities for α -substituents is <20% and hence does not significantly contribute to uncertainties in σ_{ij} , for differences of >10%, only the larger value was used to obtain distance. Larger differences in cross peak intensities are observed for β -substituents, and hence no attempts were made to extract σ_{ij} .

⁽⁴³⁾ The Arg 38 $C_{\delta}H$, $C_{\delta}H'$ geminal pair yielded a well-defined rise curve based on peak height (Figure 3), but the base line artifacts precluded the determination of the peak volumes.

Table II. Cross Relaxation Rates and Distances Determined in HRP-CN from NOESY Cross Peak Intensity^a

| residue | proton pairs $i \rightarrow j$ | $a_{ij}(\tau_{\rm m}^*)^b$ | $\tau_{\rm m}^*,$ ms | σ_{ij} , s ^{-1 c} | rel r _{if} | rel r _{ij} , ^e | r _{ij} , A | expected r _{ij} , Å |
|---------|--|----------------------------|----------------------|-----------------------------------|---------------------|------------------------------------|---------------------|------------------------------|
| His 170 | C _β H→C _β H′ | 2.83 | 1.5 | 44.8 | 0.90 | 1.02 | 1.81 | 1.77 |
| | C _β H′→C _β H | 3.57 | 1.5 | 56.5 | 1.14 | 0.98 | 1.73 | 1.77 |
| Arg 38 | C _β H→C _β H′ | 3.67 | 1.5 | 58.1 | 1.17 | 0.97 | 1.72 | 1.77 |
| | C _β H′→C _β H | 2.65 | 1.5 | 42.0 | 0.84 | 1.03 | 1.82 | 1.77 |
| Heme | 7H _a →7H _a ′ | 2.42 | 1.5 | 38.3 | 0.77 | 1.04 | 1.85 | 1.77 |
| | $7H_{\alpha}' \rightarrow 7H_{\alpha}$ | 3.70 | 1.5 | 58.6 | 1.18 | 0.97 | 1.72 | 1.77 |
| | δ-meso-H→8-CH ₃ | 1.20 | 3.0 | 3.2 | 0.064 | 1.58 | 2.80 | 2.85 |
| His 170 | C _α H→C _β H | 1.33 | 2.2 | 14.4 | 0.29 | 1.23 | 2.18 | |
| | N _p H→C _β H′ | 1.59 | 2.2 | 17.2 | 0.35 | 1.19 | 2.11 | |
| | C _∂ H′→N _P H | 1.40 | 2.2 | 15.1 | 0.30 | 1.22 | 2.16 | |
| heme | β -meso-4 H_{α} | 1.29 | 2.2 | 13.9 | 0.28 | 1.23 | 2.18 | |

^a 3 mM in ²H₂O, pH 7.0 at 35 °C. ^b Volume of cross peak (in arbitrary units, but the same as for diagonal, $V_{0i}/n_i = 42.1$) for mixing time τ_m^* (longest τ_m where rise curve is linear); uncertainty ±15%. ^c Cross relaxation rate, σ_{ij} , obtained via eq 5; uncertainty ±30%. ^d Relative σ , obtained by scaling σ value obtained via eq 5 by the mean σ for the known four geminal pair data, $\bar{\sigma}(CH_2) = 49.0 \text{ s}^{-1}$. ^e Rel $r_{ij} = (\text{rel } \sigma)^{1/6}$, via eq 2; uncertainty ±5%. ^f Obtained from rel r_{ij} using $r_{ij} = 1.77$ Å for geminal protons; uncertainty ±5%.

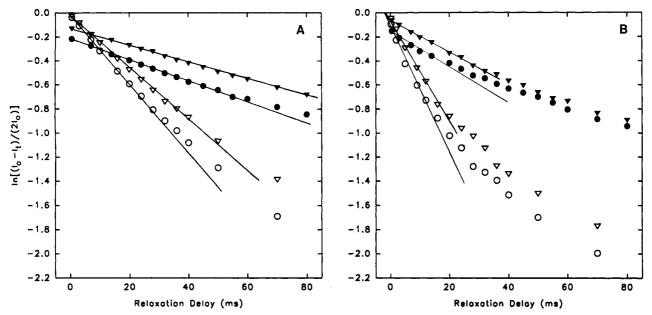


Figure 7. Semilogarithmic plots of fractional recovered intensity vs relaxation delay, τ , for nonselective inversion-recovery data (solid markers) and selective saturation-recovery data (open markers). (A) Plots for 8-CH₃ (\bullet and O) and 3-CH₃ (∇ and ∇), and (B) plots for His 170 β (\bullet and O) and His 170 β' (∇ and ∇). Note that only the initial portions of the plots are linear and that the selective data show greater slope (larger ρ).

well with that estimated by the Stokes-Einstein equation for a 44-kDa globular protein¹² (22 ns) and that estimated from the quadrupolar relaxation⁴⁴ of deuterium-labeled HRP-CN (28 ns). The use of the τ_c from the Stokes-Einstein equation, together with the observed σ_{ij} , yields a geminal $r_{ij} = 1.72$ -Å separation. Thus we conclude that the initial rise curves for a paramagnetic low-spin ferric heme peroxidase or similar sized hemoprotein can be used to estimate absolute distances, in particular those distances that will uniquely identify geminal protons.

Relaxation Rates. In Figure 7 we present the semilogarithmic plots for the recovery of magnetization versus relaxation delay time, τ , for nonselective ($\rho^{nsel} = (T_1^{nsel})^{-1}$) and selective ($\rho^{sel} = (T_1^{sel})^{-1}$) experiments for the heme 8-CH₃ and 3-CH₃ peaks (panel A) and the axial His C_βH and C_βH' peaks (panel B). It is clear that the recoveries are not single exponential³⁷ and that computer fits to straight lines over an arbitrary long time will yield ρ (or T_1) estimates that are much slower (longer) than the initial slopes. The initial slopes yield relaxation rates as listed in Table I, where similar data for the other resolved peaks are also included. The initial slope was generally maintained only for a time shorter than the $T_1 = \rho^{-1}$ determined from the initial slope. The degree of curvature in the plots varies significantly for the resolved resonances, with the largest observed for geminal proton pairs. Difference in T_1^{nsel} patterns from those reported earlier^{5,26} is due to the earlier failure to emphasize data collection at much shorter times. Note that the nonselective T_1 s are virtually useless for predicting the mixing time for optimal cross peak intensity. The selective saturation magnetization recovery curves in Figure 7 exhibit substantially steeper slopes and larger ρ s (shorter T_1 s); after a time interval $\sim T_1^{sel}$, significant curvature that could yield longer T_1^{sel} estimates by a computer least-squares fit is also apparent. In general, ρ^{sel} is longer than ρ^{nsel} by factors between 2 and 10, which emphasizes that the simpler nonselective T_1 data cannot substitute for the T_1^{sel} . The correlation between $(\rho^{sel})^{-1}$ and τ_m at maximum cross peak intensity is much better than that for the $(\rho^{nsel})^{-1}$, as should be expected.¹⁰⁻¹²

Another measure of the intrinsic relaxation rate is provided by the initial rate of the decay of the diagonal intensity with mixing time in a NOESY experiment.^{10-12,38} The diagonal intensity is given by:¹⁰⁻¹²

$$a_{ii}(\tau_{\rm m}) = \frac{V_{0i}|\sigma_{ij}|}{2D} (1 + e^{-2D\tau_{\rm m}}) e^{-(R-D)\tau_{\rm m}}$$
(6)

which, for very short $\tau_{\rm m} \ll (2D^{-1})$, $(R - D)^{-1}$, yields:

$$a_{ii}(\tau_{\rm m}) = V_{0i} \frac{|\sigma_{ij}|}{D} (1 - R_i \tau_{\rm m})$$
 (7)

where R_i is the effective relaxation rate and V_{0i} is the equilibrium magnetization (diagonal volume at $\tau_m = 0$). Plots of a_{ii} vs τ_m

⁽⁴⁴⁾ La Mar, G. N.; Thanabal, V.; Johnson, R. D.; Smith, K. M.; Parish, D. W. J. Biol. Chem. 1989, 264, 5428-5434.

were found linear for τ_m to ~3 ms and the resulting slopes listed as R_i in Table I. R_i and ρ_i^{sel} differ significantly primarily for geminal protons, for which R_i is generally larger than ρ_i^{sel} by a factor of 1.3–1.7. It is noteworthy, however, that it is R_i^{-1} that correlates best with the τ_m for the maximum intensity of a cross peak.

Steady-State NOEs. Steady-state NOE measurements for HRP-CN have been reported at 55 °C, where the lines are narrowers and resolution, particularly in the upfield portion, is significantly improved.^{4,5,25} We report here only representative steady-state NOEs at the present temperature of interest, 35 °C, in order to provide a comparison of distances obtained by 1D and 2D NOE methodology. Saturation of the axial His C_βH and C_βH' signals yields the steady-state NOEs $\eta(C_{\beta}H \rightarrow C_{\beta}H') = -0.68 \pm 0.07$ and the reciprocal value, $\eta(C_{\beta}H' \rightarrow C_{\beta}H) = -0.53 \pm 0.06$. The selective relaxation rates for these two signals given in Table I yield¹⁴ $|\sigma_{ij}| = 38 \pm 4$ and $39 \pm 4 \text{ s}^{-1}$, respectively, via eq 4. The use of the nonselective relaxation rates in Table I, on the other hand, yields cross relaxation rates of $\sim 10 \pm 1 \text{ s}^{-1}$ in each case.

Discussion

Evaluation of NOESY Data. The present NOESY data on HRP-CN reveal that it is indeed possible to obtain reliable and quantitative rise curves that discriminate between primary and secondary NOEs and provide a sufficient initial linear portion to obtain both the relative and an estimate of the absolute cross relaxation rates for a variety of proton pairs that reasonably closely correlate with known values. While the commonly used 20-ms mixing time in isoelectronic, isostructural complexes^{26-29,33,34} provides a good single value that allows observation of the majority of the primary NOEs, a large number of secondary NOEs are also observed which could lead to incorrect assessments of proximity between proton pairs. the crucial limitation to both the differentiation of primary and secondary NOEs and the quantification of an initial linear portion to obtain distance is sensitivity. The practicality of this study, which in all entailed nearly 20 NOESY maps, each requiring 10 h, was facilitated by the availability of sufficient protein to provide a 3 mM sample to adequately fill a 10-mm tube. Comparison of sensitivity characteristics of this 10-mm probe with a state-of-the-art 5-mm ¹H probe indicates that the NOESY spectrum is obtainable in 20 h per $\tau_{\rm m}$ value in the latter probe, which is not significantly larger than the data collection time needed for a diamagnetic system. The inherently lower intensity of cross peaks in the paramagnetic system is offset, in large part, by the ability to use a significantly shorter pulse sequence repetition time (i.e., ~ 1.5 $T_{1s} \sim 200$ ms), allowing a much larger number of scans per block than for a comparable diamagnetic system.³ However, the faster repetition rate also will cause saturation of some hyperfine shifted protons such as β -protons on the heme (vinyl, propionate $H_{\beta}s$), precluding the determination of reliable distances involving these protons from the present data.42

Valuable qualitative information obtained from the detailed rise curves and not obtainable from the most commonly used $\tau_m \sim 20-30$ ms is stereospecific assignments of methylene groups. Thus, the primary NOE 7-H_a \rightarrow 8-CH₃ and secondary NOE 7H_a' \rightarrow 8-CH₃ and the His 170 primary NOE C_gH \rightarrow C_aH and secondary NOE C_gH' \rightarrow C_aH differentiate the 7- α -methylene and His 170 methylene protons, respectively. Such stereospecific assignments will be valuable for the detailed analysis of dipolar shifts in terms of the magnetic axes of the heme iron.⁴⁵ The present study indicates that NOESY data should allow the stereospecific assignments and determination of detailed stereochemistry relative to the heme of key noncoordinated catalytic residues in the heme pocket.^{5,25,26} Such studies, however, must involve further extensions of scalar correlation experiments,^{2,26} as well as improvements of the NOESY spectra to minimize the artifacts in the diamagnetic region,^{39,40} since the resonances from noncoordinated residues, while hyperfine shifted, generally exhibit shifts insufficient to resolve a significant number of peaks outside the diamagnetic envelope. Such studies to address these problems are in progress.

In the absence of bond correlation spectra, the NOESY cross peak intensities can clearly identify immobile methylene groups. At mixing times of 0.5-2.0 ms, they dominate the NOESY map and provide enough intensity to allow a semiquantitative estimate of σ_{ij} and hence r_{ij} ; other proton pairs exhibit <20% of the intensity of geminal pairs in this linear region. Conversely, NOESY maps collected at >30 ms do not necessarily allow a distinction between geminal and nongeminal proton pairs. This is because the intrinsic relaxation rate for methylene protons is so much faster than that suggested by nonselective T_1 s that the cross peaks will have decayed significantly and do not necessarily exceed that for a more weakly coupled proton pair with slower intrinsic relaxation rate. Certainly the nonselective T_1 s do not directly provide useful information for selecting an optimal mixing time for a low-spin heme peroxidase; selective T_1 s serve as a more appropriate basis (Table I). The reason for this is that cross relaxation can easily exceed the paramagnetic contribution to the relaxation process (see below).

Orientation of Side Chains. The NOESY rise curves provide more definitive information on structure than steady-state NOEs and single mixing time NOESY spectra. The cis orientation of the 4-vinyl and the trans orientation of the 2-vinyl group, previously proposed on the basis of limited data,^{4,26} are quantitatively confirmed. However, the 2-vinyl $H_{\beta t} \rightarrow 3-CH_3$ NOE, previously judged primary⁴ by time-dependent 1D NOEs, is shown to be secondary, and the pathway for this transfer is shown to proceed $2H_{\beta t} \rightarrow \alpha$ -meso-H \rightarrow 3-CH₃. The reason for this misinterpretation of the 1D NOEs, and a basic limitation of 1D NOEs of any type on a paramagnetic system,^{3,46} is that the time needed to sufficiently saturate a resonance limits times for investigation to >10 ms. The secondary NOE is shown here to develop in less than half this time. A molecular model of protohemin clearly shows that in any *trans* orientation, the 2-H_{β t} will always be closer to α -meso-H than 3-CH₃. For the cis-oriented 4-vinyl group, the 2.2-Å $4H_{\alpha}$: β -meso-H distance, together with the molecular modeling of these two distances as a function of the vinyl/heme dihedral angle, suggests an angle of $45 \pm 20^{\circ}$ for the 4-vinyl group.⁴⁷ It is interesting that, while solution ¹H NMR has indicated 2-vinyl/4-vinyl trans/cis orientations in lignin peroxidase^{27,28} and all derivatives of HRP,^{4,48} the orientation for the two vinyls in $CcP^{21,22}$ are *cis* and *trans* for the 2-vinyl and 4-vinyl group, respectively.

The NOE pattern among the vicinal protons of the 7-propionate group (Figure 5B) dictates an orientation with very similar disposition of $7H_{\alpha}$: $7H_{\beta}$ and $7H_{\alpha}'$: $7H_{\beta}'$ pairs, *i.e.*, the $H_{\alpha}-C_{\alpha}-C_{\beta}-H_{\beta}$ and $H_{\alpha}'-C_{\alpha}-C_{\beta}-H_{\beta}'$ dihedral angles θ are both either ~0° or ~60°. However, the cos² θ dependence of scalar coupling⁴⁹ and the previous observation²⁶ of intense COSY peaks only for the proton pairs $7H_{\alpha}-7H_{\beta}'$ and $7H_{\alpha}'-7H_{\beta}$ dictate that the propionate chain is extended (*i.e.*, $H_{\alpha}-C_{\alpha}-C_{\beta}-H_{\beta}$ and $H_{\alpha}'-C_{\alpha}-C_{\beta}-H_{\beta}'$ dihedral angles each ~60°) with the carboxylate group pointed away from the heme rather than toward the iron. The axial His 170 NOESY cross peak rise curves reveal an orientation with $C_{\beta}H$ closer to $C_{\alpha}H$ and $C_{\beta}H'$ closer to the N_PH, with the large hyperfine shifted $C_{\beta}H$ closer to $C_{\alpha}H$; these two

⁽⁴⁶⁾ Lecomte, J. T. J.; Unger, S. W.; La Mar, G. N. J. Magn. Reson. 1991, 94, 112-122.

⁽⁴⁵⁾ Emerson, S. D.; La Mar, G. N. Biochemistry 1990, 29, 1556–1566. Rajarathnam, K.; La Mar, G. N.; Chiu, M. L.; Sligar, S. G. J. Am. Chem. Soc. 1992, 114, 9048–9058.

⁽⁴⁷⁾ The dihedral angle is defined as 0° for an in-plane, cis orientation.
(48) Thanabal, V.; de Ropp, J. S.; La Mar, G. N. J. Am. Chem. Soc. 1986, 108, 4244–4245.

⁽⁴⁹⁾ Karplus, M. J. Chem. Phys. 1959, 30, 11-15.

distances are estimated at ~2.1-2.2 Å. The axial His 175 in the CcP-CN crystal structure^{21,22} reveals a similar orientation with one C_βH close to N_PH and the other closer to C_αH; the observed distances are 2.6 and 2.3 Å, revealing a very similar but not identical orientation of the axial His side chain with respect to the helical backbone.

Relaxation Studies. In a macromolecule exhibiting cross relaxation, the recovery of magnetization is not a single exponential^{3,37} and a " T_1 " can only be extracted from the initial linear portion of a semilogarithmic plot of $I - I_0$ as a function of time. A spin *i* with magnetic moment *I*, with intrinsic relaxation rate ρ_i (both diamagnetic and paramagnetic contributions) interacting with spins *j* with magnetic moment *J* via cross relaxation σ_{ij} , obeys the equation:^{12,37}

$$\frac{dI}{dt} = -\rho_i (I - I_0) - \sum^j \sigma_{ij} (J - J_0)$$
(8)

The initial linear portions of a nonselective and selective T_1 experiment yield recovery rates, respectively:

$$\rho_i^{\text{nsel}} = \rho_i + \sum^j \sigma_{ij} \tag{9}$$

$$\rho_i^{\text{set}} = \rho_i \tag{10}$$

The intrinsic relaxation rate has both diamagnetic and paramagnetic components,^{3,37} $\rho_i = \rho_i(\text{dia}) + \rho_i(\text{para})$. Since σ_{ij} is negative for a macromolecule,¹² we always have $\rho^{\text{sel}} > \rho^{\text{nsel}}$, as observed in Table I. Because σ_{ij} s are so large in a large macromolecule (approaching $-\rho_i(\text{dia})^{12}$), ρ^{sel} , ρ^{nsel} diverge strongly as a macromolecule increases in size. While nonselective T_1 s may approximate intrinsic T_1 s in a small molecule (or in a molecule where the paramagnetic contribution to ρ_i is much larger than $\Sigma \sigma_{ij}$), the correlation fails in large molecules with only moderate paramagnetic contribution to ρ_i , such as low-spin heme peroxidases.³

In Table I we list ρ_i^{sel} measured by saturation recovery, R_i obtained from NOESY diagonal decay,^{10,38} the nonselective ρ_i^{nsel} , and the differences $\Delta \rho_i = \rho_i^{sel} - \rho_i^{nsel}$ and $\Delta \rho_i' = R_i - \rho_i^{nsel}$. Also included are the ratios ρ^{sel}/ρ^{nsel} . For a relatively isolated proton such as the distal His 42 C_eH in ²H₂O, the difference in relaxation rates is small, which is consistent with the detection of only a few relatively weak NOEs.⁵ On the other hand, for all geminal partners, $\Delta \rho_i = \rho_i^{sel} - \rho_i^{nsel}$ equals or slightly exceeds σ_{ij} to its geminal partner. Thus the difference in selective and nonselective relaxation rates finds a ready correlation with the degree of cross relaxation experienced by a proton.^{3,12} From the practical side these observations have two consequences. Nonselective T_1 values will be useless for estimating optimal NOESY mixing times or interpreting steady-state NOEs (see below). However, a possible use of $\rho^{\text{sel}}, \rho^{\text{nsel}}$ is to identify geminal proton pairs in larger paramagnetic macromolecules by inspecting the difference in ρ^{sel} and ρ^{nsel} for signals that approach the expected geminal σ for the particular sized protein.

For selection of τ_m for optimal sensitivity for NOESY cross peaks, R_i^{-1} obtained by decay of the NOESY diagonal is the appropriate indicator.¹⁰⁻¹² However, the selection of τ_m can still be optimized prior to embarking on potentially very long 2D data collection. Since the diagonal peaks are typically 1 order of magnitude more intense than any cross peak in the linear regime, a fast set of NOESY spectra to estimate R_i can be collected at the rapid pulsing rate with minimal scans (32 scans as opposed to the 352 needed for data in Figure 2), reducing the data collection time by a factor ≥ 10 .

Interpretation of Steady-State NOEs. The His $C_{\beta}H:C_{\beta}H'$ steady-state NOE data yield¹⁴ $|\sigma_{ii}| \sim 40 \pm 4 \, \text{s}^{-1}$ using the selective relaxation rate for the respective signals, which is in reasonable agreement with the values (~49 \pm 10 s⁻¹) determined by the NOESY cross peak rise curves. One likely reason for the smaller σ_{ij} from steady-state NOEs than NOESY cross peaks is that steady state is not achieved for 50-100 ms, and during this time multiple secondary NOEs develop and the two spin approximations become invalid. The NOESY-determined cross relaxation rates are all derived from data collected at very short τ_m (1.5-3.0 ms), which is generally before any secondary NOEs are observed. The relationship between ρ^{sel} and ρ^{nsel} , as it relates to interpretation of steady-state NOEs, is not their difference but their ratio, ρ^{sel} ρ^{nsel} , which is included in Table II. The use of the nonselective relaxation rates in determining σ_{ii} for the His C_bHs via eq 4, on the other hand, yields values that seriously underestimate the cross relaxation rates (by a factor \sim 4–6) and hence significantly overestimate distance (by $\sim 30\%$). The degree that steady-state NOE interpretations of σ_{ij} are incorrect by using nonselective versus selective relaxation rates is given by the factor ρ^{sel}/ρ^{nsel} listed in Table I. For nongeminal proton pairs for which the ratio $\rho^{\rm sel}/\rho^{\rm nsel}$ is particularly large (*i.e.*, 4H_a, 3CH₃), distances obtained using ρ^{nsel} would result in substantial errors in r_{ij} . Hence, while steady-state NOEs clearly demand the use of the selective rather than the nonselective relaxation rate, the use of the proper rate appears to yield distances that are similar to those obtained from NOESY data.

Acknowledgment. The authors are indebted to A. D. McPherson for experimental assistance. M. Sette is grateful to M.U.R.S.T. for a fellowship to carry out these studies. This research was supported by a grant from the National Institutes of Health, GM 26226, and the NMR spectrometer was purchased, in part, by funds provided by the National Institutes of Health, RR-04795, and the National Science Foundation, BBS-88-04739.