Articles

# **Quinoxalinylethylpyridylthioureas (QXPTs) as Potent Non-Nucleoside HIV-1 Reverse Transcriptase (RT) Inhibitors. Further SAR Studies and Identification** of a Novel Orally Bioavailable Hydrazine-Based Antiviral Agent

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Quinoxalinylethylpyridylthioureas (QXPTs) represent a new class of human immunodeficiency virus type 1 (HIV-1) non-nucleoside reverse transcriptase (RT) inhibitors (NNRTIs) whose prototype is 6-FQXPT (6). Docking studies based on the three-dimensional structure of RT prompted the synthesis of novel heteroarylethylpyridylthioureas which were tested as anti-HIV agents. Several compounds proved to be potent broad-spectrum enzyme inhibitors and significantly inhibited HIV-1 replication in vitro. Their potency depends on the substituents and the nature of the heterocyclic skeleton linked to the ethyl spacer, and structure-activity relationships are discussed in terms of the possible interaction with the RT binding site. Although the new QXPTs analogues show potent antiviral activity, none of the compounds tested overcome the pharmacokinetic disadvantages inherent to ethylpyridylthioureidic antiviral agents, which in general have very low oral bioavailability. Through an integrated effort involving synthesis, docking studies, and biological and pharmacokinetic evaluation, we investigated the structural dependence of the poor bioavailability and rapid clearance within the thioureidic series of antivirals. Replacing the ethylthioureidic moiety with a hydrazine linker led to a new antiviral lead, offering promising pharmacological and pharmacokinetic properties in terms of antiviral activity and oral bioavailability.

## Introduction

Among the antiviral agents so far proposed for the treatment of human immunodeficiency virus type 1 (HIV-1) infection,<sup>1</sup> the non-nucleoside reverse transcriptase inhibitors (NNRTI) represent an intriguing class of therapeutic agents. Although NNRTIs are highly specific and less toxic than nucleoside inhibitors (Chart 1, zidovudine (AZT), 1),<sup>2</sup> their therapeutic effectiveness is limited by relatively rapid emergence of drug-resistant HIV-1 strains and by the onset of skin rashes. NNRTIs (Chart 1) include structurally unrelated subclasses of compounds that bind a common allosteric

site, adjacent to the NRTI (nucleoside reverse transcriptase inhibitor) binding site, by a similar threedimensional arrangement. HEPT and TIBO derivatives  $(2)^3$  were the first compounds described in this class, followed by nevirapine  $(3)^4$  and efavirenz (4), both of which have been approved for clinical use for the therapy of acquired immunodeficiency syndrome (AIDS). Recently the PETT class of RT inhibitors has been described. The representative compound of this series is trovirdine (5), a potent RT inhibitor that, however, has low oral bioavailability.<sup>5-7a,b</sup> In the course of our research aimed at identifying novel antiviral agents.<sup>8</sup> we discovered that members of a series of quinoxalinylethylthioureas inhibited the replication of HIV-1 in different cell lines.<sup>9</sup> First, we identified the tricyclic quinoxaline system as a critical structural requirement for optimal interaction with the NNRTI binding site, and 6-FQXTP (6), a potent broad-spectrum secondgeneration NNRTI but characterized by unsatisfactory oral bioavailability, was selected as the lead compound.9

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Chart 1



**Chart 2.** Planned Modifications to 6-FQXPT and the Newly Designed Antiviral Agents



8 The New Antiviral Lead

To discover analogues with greater potency and better pharmacokinetic properties, we extended our structureactivity relationship (SAR) studies to assess the effect of a variety of substituents and heterocyclic systems (7) in place of the quinoxalinyl scaffold of 6-FQXPT (Chart 2). Potent broad spectum RT inhibitors with high antiviral efficacy were identified (12 and 15). Despite these promising results, studies performed to evaluate the pharmacokinetic properties of the new compounds were disappointing. None of the tested compounds showed an improved pharmacokinetic profile over the known thioureidic antiviral agents in terms of oral bioavailability. Thus, starting from 6, the objective of the present study was also the development of QXPTrelated antiviral agents with better oral bioavailability to bypass the pharmacokinetic disadvantages inherent to the thioureas. Exploiting synthesis, molecular modeling, and biological evaluation, our SAR studies (Chart 2) led to the identification of the quinoxaline derivative 8, representative lead of a new class of RT inhibitors,



offering satisfactory oral bioavailability and higher antiviral activity than that of nevirapine. Furthermore, this novel antiviral agent readly crosses the blood-brain barrier and has potent synergistic activity with AZT. Herein we report the synthesis and biological evaluation of a series of heterocyclic analogues of 6-FQXPT (**12a**– **g**, **15a**–**c**) and the development of **8**. Pharmacokinetic and synergistic studies for the most active analogues were also done.

## Chemistry

Syntheses were accomplished as schematized in Schemes 1 and 2. The desired products 12a-g and 15a-c were prepared starting from the corresponding key amino intermediates 11a-g and 14a-c following a similar synthetic procedure using 2-(5-bromopyridyl)-isothiocyanate, which is in turn prepared from 2-amino-5-bromopyridine and 1,1'-thiocarbonyldiimidazole or thiophosgene.<sup>9</sup> The azido intermediates 10a-g and

Scheme 2



13a-g were obtained starting from the lactams 9a-g by direct alkylation with tosylethyl azide, while 13h (R = 6-F, A = CH, X = N, Y = C, and Z = CH) and **13i** (R = 7.8-diMe, A = CH, X = N, Y = C, and Z = CH) were previously synthesized.<sup>9</sup> The lactams **9a**-g were prepared according to previously described procedures.<sup>10–14</sup> Usually the O-alkylated products were obtained in 10-25% yield. After examination of several reduction methods, we found that treatment of azides **10a**-g and **13d**, h, i with 1,3-propanedithiol-triethylamine in methanol<sup>15</sup> provided the corresponding amines 11a-g and 14a-c. Last, by treatment of 11a-g and 14a-c with 2-(5-bromopyridyl) isothiocyanate in the presence of triethylamine we obtained the thioureas **12a**–g and **15a**–c. The syntheses of compounds **8** and 18a,b are outlined in Scheme 2. Lactam 16<sup>10</sup> was transformed to the hydrazine derivatives 17 by means of phosphorus oxychloride and hydrazine. Treatment of **17** with triphosgene and 2-amino-5-bromopyridine provided 18a, while treatment of 17 with 5-bromopyridyl-2-isothiocyanate (in turn obtained from 2-amino-5bromopyridine and 1,1'-thiocarbonyldiimidazole) furnished 18b. Moreover, reaction of 17 with pyrazinecarboxylic acid or pyridine-2-carboxylic acid, in the presence of triphenylphosphine and Aldrithiol-2,16 gave 8 and 19, respectively, in good overall yield.

#### **Biological Results**

**Enzymatic Assays. In Vitro SAR Studies.** The newly described ethylpyridylthioureas 12a-g and 15a-c, and the analogue **8**, were tested in an in vitro HIV-1 RTwt assay to evaluate their potential as anti-HIV drugs. The results are summarized in Table 1 as  $K_i$  values. In addition, in order to evaluate the potential of these antienzymatic derivatives as novel broad-spectrum antiviral agents, the whole set of active compounds was also tested on a panel of RT mutants including L100I, Y181I, V106A, one of the most commonly identified mutations associated with dipyridodiazepinones, and K103N, a mutant associated with resistance to pyridones and nevirapine as well. The

**Table 1.** Inhibition of HIV-1 Wild-Type RT and HIV-1 Mutant HIV-1 RT Enzymes Containing the Single Amino Acid Substitutions L100I, V179D (for **15b,c**), K103N, V106A, Y181I, and Y188L by Compounds **8**, **12a–g**, **15a–c**, and **18a,b**, and **19** 

	HIV-1 RT $K_i (\mu M)^a$ poly rA (dT) <sub>12-18</sub>						
compd	WT	L100I	V179D	K103N	V106A	Y181I	Y188L
8	0.19	0.3		NA	0.8	2.5	1.8
12a	0.029	0.7		0.65	0.04	1.0	1.2
12b	0.04	0.8		0.4	0.06	0.9	0.2
12c	0.022	0.1		NA	0.03	1.6	0.45
12d	0.02	0.16		0.15	0.02	0.24	0.9
12e	0.009	0.095		0.25	0.05	0.4	0.13
12f	0.25	1.0		0.24	0.35	1.4	5.0
12g	0.015	0.1		0.021	0.33	0.52	0.4
15a	0.058	0.15		0.35	0.029	0.3	0.7
15b	0.053	0.12	0.07	0.21	0.12	0.09	0.1
15c	0.22	0.5	0.8	0.45	0.15	0.66	1.5
18a	2.6						
18b	1.9						
19	4.0						
$QXPT^{b}$	0.058	0.105		0.1	0.112	4.0	2.5
6-FQXPT <sup>b</sup>	0.009	0.17		0.16	0.015	0.2	0.2
efavirenz	0.03	0.12		0.16	0.04	0.15	0.3
nevirapine	0.4	9		7	10	36	18

<sup>*a*</sup> Inhibition of HIV-1 RT activities. All data listed were compared to the corresponding test results for nevirapine, performed at the same time. The  $K_i$  is stated as the mean of at least three experiments. <sup>*b*</sup> From ref 9. NA = not active.

Y188L, a mutation of RT that appears in patients treated with both nevirapine and AZT, was also included.

The first question addressed was whether, by expanding the SAR studies of the QXPT class of antiviral agents, the antiviral activity, the broadening of the spectrum of antienzymatic activity, and, possibly, the pharmacokinetic properties could be improved over the known thioureas. SAR studies were conducted by appropriately varying the substituents on the benzo-fused ring and by introducing different heteroaryl systems in place of the pyrrologuinoxaline of the lead **6** (6-FQXPT).

As illustrated in Table 1 for compounds 12a-g and **15a**–**c**, the potency of inhibition of RTwt and mutated RTs is strictly dependent on the nature of the substituents on the aromatic system. By comparing the inhibitory activity ( $K_{is}$ ) of compounds 12a-c to that of 6-FQXPT (6), it clearly appears that the electronwithdrawing fluorine is preferred at C-6 (6-FQXPT), this being 4-fold more potent against RT wt than its 7-, 9-, and 7,9-fluorinated analogues (12a-c). While against the V106A mutant **12a**–**c** show a comparable activity to that of 6-FQXPT (6), the 7,9-difluoro analogue 12c proved to be inactive against the K103N mutant, maintaining significant potency vs Y188L mutant. The analogue 12d (7-ClQXPT) is slightly less potent than 6 (6-FQXPT) vs RT wt, but it is about 4-fold more active than 12a (7-FQXPT) against L100I, V106A, K103N, and Y181I. In general, the halogenated analogues show greatly improved antienzymatic properties over their unsubstituted counterpart QXPT.<sup>9</sup> Furthermore, within this series, the antienzymatic potency and the broadening of spectrum of activity are also influenced by the nature of the heterocyclic skeleton (**12e**-**g** vs 6-FQXPT).

By comparing the *K*<sub>i</sub>s (Table 1) of unsubstituted QXPT and 6-FQXPT for inhibition of RTwt, L100I, K103N, and V106A with those of **12f** (ImQXPT), the key role played by the pyrrole ring with respect to the imidazole

(unsubstituted QXPT<sup>9</sup> vs **12f**) is evident. On the contrary, the pyrazole ring in place of the pyrrole (**12g** vs unsubstituted QXPT<sup>9</sup>) greatly improves the inhibitory activity against Y181I and Y188L mutants (**12g**,  $K_i =$ 0.52 and 0.40  $\mu$ M, respectively). We also investigated the analogue **12e** (PyrPZPT) in which the benzo-fused ring of **6** is replaced by a pyridine system. In this case **12e** shows increased anti-Y181I/Y188L activity, maintaining high potency against RTwt. Taking into account the inhibitory potency against RTwt and the other mutants, especially vs Y188L, it appears that **12e** (PyrPZPT) represents one of the most potent secondgeneration RT inhibitors.

The O-alkylated regioisomers **15a**–**c** were further investigated and included in Table 1. Compounds **15a** and **15b** proved to be potent broad-spectrum RT inhibitors, being as active as their N-alkylated isomers (**6** and **12d**). In particular, **15b** demostrated higher potency vs Y181I and Y188L than its N-alkylated isomer **12d**.

When subjected to preliminary pharmacokinetic studies, the most representative thioureas proved to be not bioavailable after oral administration, probably by rapid metabolic inactivation and clearance (first-pass hepatic elimination) (see Preliminary Pharmacokinetic Studies section). Since this unfavorable behavior was shown by thioureas characterized by different heteroaryl systems, the subsequent question addressed was whether through an alteration of the linker between the heteroaryl and the 5-bromopyridine groups an antiviral agent with higher hydrophilicity and, of consequence, higher oral bioavailability, maintaining the 3D pharmacophoric arrangement for an optimal interaction with the NNRTI binding site, could be developed. Starting from QXPTs, by replacing the ethylthiourea moiety with a semicarbazide or a thiosemicarbazide group, two inactive compounds were obtained (18a,b). Furthermore, shortening the linker with hydrazine, and maintaining the distal pyridine ring, compound 19 was obtained, characterized by a low antienzymatic activity. Conversely, replacing the pyridinoyl with a pyrazinoyl group, we identified the new reverse transcriptase inhibitor 8. As shown in Table 1, although less potent than thioureas (8 vs unsubstituted QXPT<sup>9</sup> and 6), compound 8 displays antienzymatic activity against RT wt and mutants higher than that of nevirapine, and it represents a great improvement in oral bioavailability over known ethylthioureas (see Preliminary Pharmacokinetic Studies section). Replacement of the pyrazine nucleus of 8 with pyridine (19) and substituted pyridines (data not shown), leading to inactive compounds, demostrated the critical role played by the distal nitrogen of the pyrazine ring.

Computational studies were performed to develop docking models of **8** and **15a** into the HIV-1 NNRTI binding site which could guide further lead optimization efforts. The coordinates of the RT/nevirapine complex solved by Ren et al.<sup>17</sup> were employed to model the target protein. All calculations were performed using molecular mechanics (Tripos force field<sup>18</sup>), conformational analysis, docking, and graphics routines available within the SYBYL program.<sup>19</sup> Details of these procedures are described in the Experimental Section, whereas results are summarized below.

According to our model of RT/8 complex (see Figure 1), the pyrazine N4 atom receives a hydrogen bond from



**Figure 1.** Model of the RT/**8** complex. Dotted lines highlight intramolecular hydrogen bonds. To improve clarity, only residues within 4 Å from the inhibitor are displayed and Glu138 and Pro236 have been excluded.



**Figure 2.** Overlay of the putative RT-bound conformations of **8** (yellow) and **15a** (red) extracted together with Lys101 from the RT/**8** and RT/**15a** complexes.

the backbone NH group of Lys101. This interaction rationalizes the loss of inhibitory activity resulting from replacement of the pyrazine N4 with a CH fragment to yield the pyridine derivative (**19**). An additional intermolecular hydrogen bond is hypothesized to occur between the hydrazide CONH and the Val179 carbonyl oxygen. The pyrroloquinoxaline and pyrazine rings are arranged in a butterfly-like arrangement adopted by most NNRTIS.<sup>20</sup> The tricyclic system is hosted in cavity made up by the aromatic side chains of Tyr181, Tyr188, and Trp229.

Figure 2 shows a superposition of the putative RTbound geometries of **8** and **15a**<sup>7b</sup> extracted together with Lys101 from the corresponding complexes. The pyrroloquinoxaline systems dock similarly although they are not coincident atom by atom. None of the pyrazine nitrogens of **8** match the pyridine nitrogen of **15a**. Each of the thioureidic NH groups of **15a** makes a hydrogen bond: one with the pyridine nitrogen (intramolecular) and the other one with the Lys101 carbonyl oxygen

**Table 2.** Inhibition of HIV-1 Infection in CEM-SS Cells,<sup>*a*</sup> Inhibition of HIV-IIIB Infection in C8166 Cells, and Cytotoxicity on NSO Murine Cell Line, Daudi Human (DH) Cell Line, 3T3 Fibroblasts (3T3F) Murine Cell Line, and Normal Human Lymphocytes (HL)

	CEM-SS cells			C8166 cells			TC <sub>50</sub> (mM)			
compd	$IC_{50}{}^{b}$ ( $\mu$ M)	${\rm EC}_{50}{}^{c}$ ( $\mu { m M}$ )	$\mathbf{SI}^d$	$IC_{50}^{b}$ ( $\mu$ M)	${\rm EC}_{50}{}^{c}$ ( $\mu { m M}$ )	$\mathbf{SI}^d$	NSO <sup>e</sup>	$\mathrm{D}\mathrm{H}^{e}$	3T3F <sup>e</sup>	HL <sup>e</sup>
12a	>200	0.063	>3170	15	0.032	468	0.02	0.019	0.02	0.02
12b	>2	0.0019	>110	21	0.2	105	0.52	0.51	0.5	0.51
12c	>2	0.085	>23				NT	NT	NT	NT
12d	>2	0.0018	>1080	7	0.061	123	0.97	0.96	0.98	0.97
12e	>2	0.0052	>380	11	0.124	89	0.7	0.69	0.69	0.68
12f	>2	0.013	>150	88	0.32	275	0.68	0.68	0.69	0.67
15a	>2	0.0029	>690	125	0.020	6250	>1	>1	>1	>1
15b				93	0.025	3720	>1	>1	>1	>1
8				18	0.4	40	0.87	0.86	0.88	0.87
$QXPT^{f}$				17	0.38	45	0.8	0.8	0.8	0.8
6-FQXPT <sup>f</sup>	>0.02	0.00047	>43	200	0.025	7990	1.0	1.0	1.0	0.96
8-CITIBO				50	0.030	1670				
AZT	>1	0.0019	526	>100	0.4	>250				

<sup>*a*</sup> Testing was performed by the National Cancer Institute's Developmental Therapeutics Program, AIDS antiviral screening program. All the data listed were compared to the corresponding test results for AZT which served as the treated control, performed at the same time. <sup>*b*</sup> The IC<sub>50</sub> value is the test compound concentration which results in a 50% survival of uninfected untreated control CEM-SS and C8166 cells (e.g., cytotoxicity of the test drug). <sup>*c*</sup> The EC<sub>50</sub> value is the test compound concentration which results in a 50% survival of uninfected untreated control CEM-SS and C8166 cells (e.g., cytotoxicity of the test drug). <sup>*c*</sup> The EC<sub>50</sub> value is the test compound concentration which produces a 50% survival of HIV 1 infected cells relative to uninfected untreated controls (e.g. in vitro anti-HIV-1 activity). <sup>*d*</sup> SI = selectivity index (IC<sub>50</sub>/EC<sub>50</sub>). <sup>*e*</sup> Blank was subtracted from O.D. measured at 570 nm. All data reported in table are measured O.D. × 1000. Standard deviations are not reported in table but never exceeded 5% of the mean value. O.D.s measured in the control row were as follows: NSO 0.125 ± 0.015; DH 0.129 ± 0.018; 3T3F 0.131 ± 0.014; HL 0.114 ± 0.01. <sup>*f*</sup> From ref 9.

(intermolecular). Differences in the binding modes of **8** and **15a** imply different lead optimization strategies in the hydrazide- and thiourea-based series.

**Secondary Tests.** A selected set of active compounds in enzymatic assays was evaluated for its ability to inhibit the HIV-1 wt cytopathicity on T4 cells (CEM-SS), to inhibit syncytia formation on C8166 cells infected with wild-type HIV-1<sub>IIIB</sub> virus. Furthermore, compounds **12a** and **8**, representative of the two series of antienzymatic agents, were further tested on monocytemacrophages infected with HTLV-III Ba-L (a laboratory adapted monocytotropic strain) in order to confirm their potential as antiviral agents. MTT assays were also conducted as a measure of the cytotoxic effect of the new compounds on four different cell lines, and synergistic antiviral activity with AZT has been calculated for representative antiviral agents.

Cell Culture Assays and MTT Assays. The results are presented in Table 2. For compounds bearing the quinoxaline system (12a-d), potent antiviral activity in CEM-SS and C8166 cells was found, comparable to that of 6, although 12b, in C8166 cells, proved to be less active than the counterparts 12a and 12d. Replacement of the benzo-fused ring with a pyridine led to an analogue with great antiviral efficacy (**12e** vs QXPT), while replacement of the pyrrole ring of unsubstituted QXPT with an imidazole (12f) maintained antiviral activity (C8166 cells). The O-alkylated regioisomers of 12 (15a,b) were also tested on CEM-SS and C8166 cells. By comparing the data for 6 and 15a in Table 2 (CEM-SS and C8166), an apparent discrepancy arose. In fact, 15a shows low antiviral efficacy on C8166 cells while its EC<sub>50</sub> on CEM-SS cells is equal to 2 nM. In C8166 cells, the new hydrazine derivative 8 showed a lower antiviral efficacy with respect to those of QXPTs and related compounds, although it was comparable to that of AZT tested in the same experimental conditions. Furthermore, two representative derivatives (8 and 12a) were also studied on monocyte-macrophages infected by HTLV-III Ba-L (Table 3). HIV-1 infection of monocytemacrophages is an important event in the pathogenesis of AIDS. Monocyte-macrophages differ from T cells for

**Table 3.** Inhibition of HIV-1-p24 Antigen Production in Monocyte-Macrophages<sup>*a*</sup> by Compounds **8** and **12a** 

J 1	0 5 1		
compd	$\mathrm{IC}_{50}\;(\mu\mathrm{M})^{b}$	$EC_{50} (\mu M)^{c}$	$\mathbf{SI}^d$
<b>8</b> <b>12a</b> (7-FQXPT) 6-FQXPT <sup>e</sup>	25 49 17	0.2 0.004 0.002	125 12200 8800

<sup>*a*</sup> Antiviral activity was determined after 14 days after infection. For comparison, HIV-1-p24 antigen production in control cells was 22.789 pg/mL. All the data listed were compared to the corresponding test results for AZT which served as the treated control, performed at the same time. <sup>*b*</sup> The IC<sub>50</sub> value is the test compound concentration which results in a 50% survival of uninfected untreated control macrophages cells (e.g., cytotoxicity of the test drug). <sup>*c*</sup> The EC<sub>50</sub> value is the test drug concentration which produces a 50% reduction of HIV-1-p24 antigen production in infected cells relative to uninfected untreated controls (e.g., in vitro anti-HIV-1 activity). <sup>*d*</sup> SI = selectivity index (IC<sub>50</sub>/EC<sub>50</sub>). <sup>*e*</sup> From ref 9.

a number of aspects which could possibly affect the anti-HIV activity of some antiviral agents. In our case, **12a** proved to be extremely potent in this test, while **8** confirmed its significant efficacy, in the submicromolar range. Taking into account the data shown in assays using different infected cells, **12a**–**e**,**g** appear to be among the most potent second-generation anti-HIV agents, while the quinoxaline derivative **8**, although less potent than QXPTs, is a representative lead of a new class of antiviral agents.

The low cytotoxicity was also confirmed by screening compounds in a MTT assay (3-(4,5-dimethylthiazol-2yl)2,5-diphenyltetrazolium bromide) by using NSO murine, Daudi human, 3T3 fibroblast murine cell lines, and normal human limphocytes (Table 2). Most of the tested compounds show weak cytotoxicity.

**Synergistic Antiviral Activity with AZT (and DDI).** Resistance is a major concern in the chemotherapy of AIDS, and the recent therapeutic strategies converge to favor multidrug regimens which are more safe and effective against the HIV-1 infection, with simultaneous reduction on drug dosages, toxicity, and resistance. A further improvement in multidrug therapy would be to administer combinations of drugs able to interact synergistically. One of the clinically preferred

**Table 4.**  $EC_{50}$  and Combination Index for Different Anti-HIV-1 Compounds in Association with AZT (and DDI for **8**), Calculated Using Both the Mutually Nonexclusive and the Mutually Exclusive Assumption<sup>*a*</sup>

	EC <sub>50</sub> (nM)			$\mathrm{CI}^b$		
compound	of compound alone	of compound with AZT (DDI) <sup>c</sup>	of AZT (DDI) <sup>c</sup> with compound	mutually exclusive assumption	mutually nonexclusive assumption	
none			400 (1300)			
8	450	15 (140)	98 (0.84)	0.640 (1.18)	0.570 (0.97)	
12a (7-FQXPT)	66	57	85	0.700	0.600	
12e (PyrPZPT)	124	50	15	0.470	0.450	
12f (ImQXPT)	588	460	66	1.070	0.940	
6-FQXPT <sup>d</sup>	25	2	3	0.091	0.090	

<sup>*a*</sup> C8166 cells were infected with HIV–IIIB. <sup>*b*</sup> CI > 1, CI = 1, and CI < 1 indicate antagonistic, additive, and synergistic activities, respectively. The CI values were calculated at 50% antiviral activity using both the mutually nonexclusive and the mutually exclusive form of the equation. <sup>*c*</sup> EC<sub>50</sub> and the combination index for **8** in association with DDI are shown in brackets. <sup>*d*</sup> From ref 9.

combinations among those drugs acts on different receptor sites within the same viral protein.<sup>21</sup> Accordingly, to evaluate the potential of the new series of inhibitors in a multidrug approach, we investigated the synergistic antiviral activity of 8, 12a, 12e, and 12f in combination with the nucleoside analogue AZT (and DDI for 8) (Table 4). The compounds tested proved to be able to potentiate the antiviral activity of AZT, and AZT was able to potentiate their activity. The extent of the synergistic activity was quantified according to Chou and Talalay by the calculation of the combination index (CI). Compounds 12a, e showed a synergistic anti-HIV-1 activity (CI < 1) when tested in combination with AZT, notwithstanding the mutually exclusive or nonexclusive assumption formulas that were used (Table 4). On the contrary, 12f showed an additive antiviral activity (CI = 1). It is noteworthy that the new antiviral agent 8 showed a strong synergistic anti-HIV-1 activity in combination with AZT. Synergistic activity was less evident when 8 was tested in combination with DDI. Data shown in Table 4 confirm these new reverse transcriptase inhibitors to be potentially useful in combination therapies.

**Preliminary Pharmacokinetic Studies.**<sup>22–26a,b</sup> In previous studies we showed that the ethylpyridylthioureidic derivative **6** (6-FQXPT)<sup>9</sup> rapidly reaches the systemic circulation and enters the brain after subcutaneous dosing in mice. However, its plasma and brain concentrations were low and variable and rapidly fell below the limit of quantitation, possibly because of rapid and extensive biotransformation. Findings were similar with its 9-fluoro-substituted analogue **12b** (data not shown). After oral dosing (20 mg/kg), compounds **6** and **12b** were undetectable in mouse plasma within the limits of sensitivity of the analytical procedure (about 0.1 nmol/ml, using 0.2 mL of plasma).

In the same experimental conditions (20 mg/kg orally, suspended in PEG400:Tween 80, 80:20, v/v) the 7,9difluoro-substituted analogue **12c** showed low oral bioavailability too. The same was achieved with the more lipophilic O-alkylated regioisomer **15a** (i.e., plasma concentrations were less than 0.1 nmol/mL within 2 h of oral dosing). Possibly the water insolubility and high lipophilicity of these ethylpyridylthioureidic derivatives prevent their efficient absorption and/or lead to extensive presystemic biotransformation.

Synthetic and pharmacokinetic efforts were then focused on the development and characterization of novel NNRTIS: by modifications of the structure of QXPTs, the new lead **8** was identified (see Table 1). This



**Figure 3.** Plasma and brain concentration—time curves of compound **8** after subcutaneous and oral administration in mice. Each point represents the mean plasma (solid line) and brain concentrations (n = 3) of subcutaneously (closed circles) and orally (open circles) administered **8**.

hydrazide derivative was expected to possess a higher solubility and a lower lipophilicity compared with the thioureidic analogues of type 12 and 15 as suggested by the following calculated *n*-octanol/water logP values of 8, 12c, and 15a: 1.1, 3.7, and 4.9, respectively.<sup>26b</sup> Figure 3 shows the plasma concentration-time curves of compound 8 after subcutaneous and oral dosing in mice. Subcutaneous absorption was relatively slow, as indicated by the plasma  $t_{max}$  which averaged 120 min after a 20 mg/kg dose. The mean  $C_{\text{max}}$  was 12.9  $\pm$  3.1 nmol/ml (n = 3). Under the assumption of complete absorption of 8 from the subcutaneous injection site into the systemic circulation, the mean total body Cl averaged 22.5 mL/min/kg. In conjunction with a relatively high Vd of 3 L/kg, this resulted in a mean elimination  $t_{1/2}$  of 1.5 h, although this could only be calculated with limited accuracy because of the short sampling interval.

Absorption of **8** was faster after oral dosing, with a mean plasma  $C_{\text{max}}$  (6.5 ± 1.7 nmol/ml) at 30 min. Mean plasma concentrations of the compound declined from the peak with elimination  $t_{1/2}$  of 1.4 h, similar to that after subcutaneous injection. The relative bioavailability, calculated from the AUC of the compound after an oral (21.7 nmol/ml/h) and subcutaneous (48.7 nmol/ml/h) dose of 20 mg/kg, was about 45%.

Since the brain is a site of infection and viral replication for HIV-1, effective brain entry is a desirable property of new NNRTIS. It was therefore of interest preliminarily to examine the passage of **8** across the blood-brain barrier and the relationship between brain and plasma concentrations. Reflecting the time profiles in the systemic circulation, **8** achieved mean brain  $C_{\text{max}}$ at approximately the same time as in plasma after either subcutaneous  $(3.0 \pm 1.4 \text{ nmol/g}, \text{ at } 120 \text{ min})$  or oral dosing  $(0.9 \pm 0.1 \text{ nmol/g}, \text{ at } 30 \text{ min})$ . After the peak, mean brain concentrations disappeared almost in parallel with plasma concentrations, with similar elimination  $t_{1/2}$  looking at the whole brain concentrations (Figure 2). Mean brain concentrations remained consistently 15-25% of the plasma concentrations within 4 h of dosing, yielding a mean brain-to-plasma distribution ratio (plasma/brain AUC) of about 0.2 regardless of the route of administration.

### Conclusions

In summary, we have prepared a number of highly potent antiviral agents with an ethylthioureidic backbone, structurally related to 6-FQXPT. Within the series of QXPTs, the SAR studies led to the identification of 12e, a potent second-generation antiviral agent, in which the benzo-fused ring of the earlier-described QXPTs is replaced by a pyridine ring. Furthermore, in the course of this work we have also investigated the structural dependence of the poor bioavailability and rapid clearance of QXPT compounds, in order to identify antivirals with better oral bioavailability over the known thioureas. Alterations of the molecular structure of our QXPTs 6 and 12 enabled us to develop the quinoxaline derivative 8, a prototype antiviral agent of a new class of RT inhibitors. Its unique structural feature is the presence of a hydrazine group linking the heterocyclic system to the aroyl moiety. Compound 8 represents a new generation inhibitor of HIV-1 replication with potent synergistic antiviral activity with AZT. Preliminary pharmacokinetic studies in the mouse suggest that the QXPT antiviral agents, like other thioureidic antivirals, offer low oral bioavailability and/ or rapid clearance. In contrast, the new quinoxaline derivative 8 has much better pharmacokinetic properties, being rapidly absorbed from the gastrointestinal tract, with mean  $C_{\text{max}}$  of about 7  $\mu$ M after a 20 mg/kg dose. Although oral bioavailability is about half that after subcutaneous dosing, it remains to be seen whether this is due to an extensive first-pass metabolism or to incomplete absorption from the gastrointestinal tract. Compound 8 also has an intermediate-low Cl compared to liver blood flow and its Vd exceeds the total body water volume of the mouse, suggesting extensive uptake into tissue. Although its brain-to-plasma distribution ratio averaged 0.2 after a single dose, it remains to be seen whether it is similar across species because of differences in blood flow, brain size, protein binding, and other mechanisms such as active transport that govern the extent of brain uptake.

## **Experimental Section**

For general experimental information, see ref 11. Lactams **9a**–**g** used for the synthesis of N-alkylated and O-alkylated compounds were synthesized by using a procedure described in ref 10, while azides **13h**, **i** were previously reported in ref 9.

**General Procedure for Preparation of N-Alkylated Quinoxalinones 10a–g and of O-Alkylated Quinoxaline 13a.** This procedure is illustrated for the preparation of 5-(2azidoethyl)-7-fluoropyrrolo[1,2-*a*]quinoxalin-4(5*H*)-one (**10a**) and 4-(2-azidoethoxy)-7-fluoropyrrolo[1,2-a]quinoxaline (13a). To a suspension of sodium hydride (69.7 mg, 2.81 mmol) in anhydrous DMF (10 mL) was added the pirroloquinoxaline 9a (0.6 g, 2.81 mmol) in one portion. The reaction mixture was stirred at room temperature for 2 h, and then tosylethyl azide (667 mg, 2.81 mmol) was added. After 24 h of stirring at 80 °C the solvent was removed under vacuum, and the residue was dissolved in ethyl acetate. The organic phase was washed with brine, dried, and concentrated. The residue was purified by flash chromatography (ethyl acetate and hexanes, 1:3, as eluant) to afford 13a and 10a as white solids (compound 13a eluted first). After recrytstallization (ethyl acetate and hexanes, 1:3) the N-ethylazide 10a was obtained as colorless prisms. 10a: (59%) mp 133-134 °C; IR (Nujol) 2104, 1653, 1611 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  3.72 (m, 2 H), 4.41 (m, 2 H), 6.66 (m, 1 H), 6.95 (m, 1 H), 7.15 (m, 1 H), 7.25 (m, 1 H), 7.66 (m, 2 H). Anal. Calcd for C<sub>13</sub>H<sub>10</sub>FN<sub>5</sub>O: C, 57.56; H, 3.72; N, 25.82. Found: C, 57.57; H, 3.69; N, 25.65.

**13a**: (20%) colorless prisms, mp 66–67 °C; IR (CHCl<sub>3</sub>) 2109 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  3.70 (m, 2 H), 4.78 (m, 2 H), 6.76 (m, 1 H), 6.94 (m, 1 H), 7.08 (m, 1 H), 7.38 (m, 1 H), 7.68 (m, 1 H), 7.78 (m, 1 H). Anal. Calcd for C<sub>13</sub>H<sub>10</sub>FN<sub>5</sub>O: C, H, N.

**General Procedure for Preparation of the Aminoethyl Derivatives 11a–g and 14a–c.** This procedure is illustrated for the preparation of 5-(2-aminoethyl)-7-fluoropyrrolo[1,2-*a*]quinoxalin-4(5*H*)-one (**11a**).

The azide **10a** (112 mg, 0.4 mmol) was suspended in dry, freshly distilled methanol (4 mL). The flask was purged with argon, and propane-1,3-dithiol (120  $\mu$ L, 1.2 mmol) and anhydrous triethylamine (166  $\mu$ L, 1.2 mmol) were added. The solution was stirred at room temperature for 24 h, the solvent was then removed under vacuum, and the residue was purified by flash chromatography (elution with methanol and triethylamine, 9:1) to obtain **11a** as an amorfous solid (89%). The amine was used in the next step without further purification. IR (Nujol) 1635 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.5 (br s, 2 H), 3.05 (m, 2 H), 4.32 (m, 2 H), 6.63 (m, 1 H), 6.95 (m, 1 H), 7.10 (m, 1 H), 7.20 (m, 1 H), 7.55–7.65 (m, 2 H).

**General Procedure for Preparation of QXPT Related** Compounds 12a-g and 15a-c. This procedure is illustrated for the preparation of *N*-[2-(7-fluoro-4-oxopyrrolo[1,2-*a*]quinoxalin-5-yl)ethyl]-N-[2-(5-bromopyridyl)]thiourea (11a). To a solution of aminoethylquinoxalinone **11a** (127 mg, 0.5 mmol) in anhydrous THF (2 mL) was added 2-(5-bromopyridyl)isothiocyanate (107 mg, 0.5 mmol), and the reaction mixture was stirred at room temperature for 1 h. The resulting precipitate was collected by filtration through a sintered glass filter and washed with another portion of THF. Recrystallization from THF gave 12a (211 mg) as colorless prisms (91%): mp 256-257 °C. IR (Nujol) 3420, 1653 cm<sup>-1</sup>; <sup>1</sup>H NMR  $(DMSO-\hat{d_6}) \delta 3.95 \text{ (m, 2 H)}, 4.48 \text{ (m, 2 H)}, 6.70 \text{ (m, 1 H)}, 7.04$ (m, 3 H), 7.83 (dd, 1 H, J = 11.15, 2.3 Hz), 7.94 (dd, 1 H, J = 8.85, 2.1 Hz), 8.14 (m, 3 H), 10.74 (b s, 1 H), 11.35 (br s, 1 H). Anal. Calcd for C19H15BrFN5OS: C, H, N.

N-(Pyrrolo[1,2-a]quinoxalin-4-yl)hydrazine (17). A solution of 16 (1.04 g, 5.6 mmol) and a catalytic amount of N,Ndimethylaniline in phosphorus oxychloride (16.9 mL, 184 mmol) was refluxed under argon for 5 h. After cooling, the excess phosphorus oxychloride was distilled off, and the residue was partitioned between dichloromethane and sodium hydrogencarbonate solution. The organic solution was washed with brine, dried, and concentrated. The residue was purified by column chromatography (dichloromethane/EtOAc, 9:1) and recrystallized from hexanes to give the 4-chloro intermediate (0.67 g, 59%) as colorless prisms: mp 262–264 °C. This chloro derivative was dissolved in methanol (15 mL), and the solution was cooled at 0 °C. Then a solution of hydrazine monohydrate (140  $\mu$ L) in methanol (3 mL) was slowly added. After the mixture was stirred for 2 h at room temperature, the solid was filtered off and the solution was concentrated to give 17 as a white solid which was recrystallized from methanol (590 mg, 86%): mp 159-160 ° C; IR (Nujol) 3300 cm<sup>-1</sup>; <sup>1</sup>H NMR  $(DMSO-d_6) \delta 4.53$  (br s, 2 H), 6.68 (m, 1 H), 6.89 (m, 1 H), 7.30 (m. 2 H), 7.51 (d, 1 H, J = 7.6 Hz), 8.05 (d, 1 H, J = 7.6 Hz), 8,20 (m, 1 H), 8.71 (br s, 1 H). Anal. Calcd for C<sub>11</sub>H<sub>10</sub>N<sub>4</sub>: C, H, N.

4-(5-Bromopyridin-2-yl)-1-(pyrrolo[1,2-a]quinoxalin-4vl)semicarbazide (18a). To a solution of triphosgene (54 mg, 0.85 mmol) in anhydrous dichloromethane (1 mL) was added a solution of 2-amino-5-bromopyridine (86 mg, 0.5 mmol) and diisopropylethylamine (192 µL, 1.1 mmol) in anhydrous dichloromethane (1 mL). After the mixture was stirred for 5 min at room temperature, a solution of 17 (100 mg, 0.5 mmol) in anhydrous dichloromethane (1 mL) was added. The resulting solution was stirred at room temperature overnight. The solvent was evaporated, and the residue was taken up in EtOAc. The organic layer was washed with brine, dried, and concentrated. The residue was purified by chromatography (dichlorometane and 10% methanol and 1% ammonium hydroxide) to give a yellow solid which was recrystallized (EtOAc) to give 73 mg (41%) of 18a as yellow prisms: mp 114-115 ° C. IR (Nujol) 3410, 1653 cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO- $d_6$ )  $\delta$  6.06 (s, 1 H), 6.45 (d, 1 H, J = 6.6 Hz), 7.22 (m, 4 H), 7.59 (m, 1 H), 8.02 (m, 3 H), 9.98 (br s, 1 H), 10.11 (br s, 1 H), 11.64 (br s, 1 H). Anal. Calcd for C<sub>17</sub>H<sub>13</sub>BrN<sub>6</sub>O: C, H, N.

**4-(5-Bromopyridin-2-yl)-1-(pyrrolo**[1,2-*a*]**quinoxalin-4-yl)thiosemicarbazide (18b).** Starting from **17**, the title compound was obtained following the synthetic procedure described in ref 9. Compound **18b** was obtained as yellow prisms (41%): mp 236–237 ° C. IR (Nujol) 3410, 1620 cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO- $d_6$ )  $\delta$  6.80–7.59 (m, 5 H), 7.80–8.70 (m, 4 H), 9.97 (br s, 1 H), 10.62 (br s, 1 H), 10.91 (br s, 1 H). Anal. Calcd for C<sub>17</sub>H<sub>13</sub>BrN<sub>6</sub>S: C, H, N.

N-(Pyrazinecarbonyl)-N-(pyrrolo[1,2-a]quinoxalin-4yl)hydrazine (8). To a magnetically stirred suspension of pyrazinecarboxylic acid (31 mg, 0.25 mmol) in anhydrous dichloromethane (1 mL) were added, portionwise within 1 h, triphenylphosphine (132 mg, 0.5 mmol) and 2,2'-dipyridyl disulfide (Aldrithiol-2) (111 mg, 0.5 mmol), and the disappearance of the starting material was monitored by TLC (EtOAc). Then a solution of 17 (50 mg, 0.25 mmol) in anhydrous dichlometane (3 mL) was added, and the reaction mixture was stirred overnight at room temperature. The solvent was removed, and the residue was taken up in a mixture of EtOAc and 5% hydrochloric acid. The organic phase was washed with 1 N sodium hydroxide and brine, dried, and concentrated. The residue was chromatographed (dichlorometane and 10% methanol and 1% ammonium hydroxide) to give a yellow solid which was recrystallized (EtOAc) to give 45 mg of 8 as yellow prisms: mp 216-218 ° C; IR (Nujol) 3420, 1653 cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO- $d_6$ )  $\delta$  6.83 (s, 1 H), 7.20–7.30 (m, 3 H), 7.44 (m, 1 H), 8.13 (m, 1 H), 8.34 (s, 1 H), 8.87 (s, 1 H), 8.98 (s, 1 H), 9.25 (s, 1 H), 9.65 (br s, 1 H), 10.96 (br s, 1 H). Anal. Calcd for C<sub>16</sub>H<sub>12</sub>N<sub>6</sub>O: C, H, N.

*N*-(**Pyrazinecarbonyl**)-*N*-(**pyrrolo**[1,2-*a*]**quinoxalin-4yl**)**hydrazine (19).** Following a synthetic procedure similar to that described for **18a**, using pyridine-2-carboxylic acid, the title compound was obtained in 68% yield as colorless prisms: mp 148–149 °C; IR (Nujol) 3420, 1635 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  6.86 (m, 1 H), 7.10–7.25 (m, 3 H), 7.40–7.50 (m, 2 H), 7.83 (m, 1 H), 8.19 (m, 1 H), 8.46–8.74 (m, 2 H), 9.03–9.35 (m, 2 H), 10.16 (br s, 1 H). Anal. Calcd for C<sub>17</sub>H<sub>13</sub>N<sub>5</sub>O: C, H, N.

**Molecular Modeling.** Molecular modeling studies were performed using the SYBYL software package<sup>19</sup> running on a Silicon Graphics Iris Indigo R10000 workstation. The crystal structure of the RT/nevirapine complex solved at 2.2 Å resolution by Ren et al. was retrieved from the Brookhaven Protein Data Bank<sup>27</sup> (entry code 1VRT).

Intramolecular and intermolecular energies were calculated using the molecular mechanics Tripos force field including the electrostatic contribution. Atom centered partial charges were calculated according to the Gasteiger–Hückel method.<sup>28,29</sup> Geometry optimizations were realized with the SYBYL/MAXI-MIN2 minimizer by applying the BFGS algorithm<sup>30</sup> and setting a root-mean-square gradient of the forces acting on each atom at 0.1 kcal/mol Å as a convergence criterion.

Models of compounds 8 and 15a were built according to

standard distances and valence angles of the Tripos force field. By manually modifying the torsion angles about the CAr–N and N–N rotatable bonds of **8**, we found a butterfly-like orientation of this ligand superimposable on nevirapine about the aromatic systems (corresponding to the butterfly wings): the pyrroloquinoxaline B ring and the pyrazine ring of **8** were matched with the pyridine A and B rings of nevirapine, respectively. The fitting points were, more specifically, pseudo-atoms placed at 1 Å along the normals to the plane of each aromatic ring passing through the ring centroid.

The so-obtained pharmacophore-based conformation of **8** was then placed into the NNRTI binding cleft extracted from the crystal structure of the RT/nevirapine complex. The binding site comprised 71 amino acids located within a distance of 10 Å from any non-hydrogen atom of the bound inhibitor. Water molecules were deleted. Hydrogens were added to the unfilled valences of the amino acids, and the Lys, Asp, and Glu side chains were modeled in their ionized forms.

The trial RT/**8** complex was first geometry-optimized by keeping fixed the coordinates of the protein. The resulting (highly distorted) geometry of the ligand was energy-minimized alone while retaining the conformation of the spacer through appropriate constraints applied to torsional angles. The optimized output geometry of the ligand was then placed into the NNRTI binding site through an overlay on the corresponding input geometry about non-hydrogen atoms. Finally, the complex was fully energy-minimized by keeping fixed the protein backbone until an attractive energy of 25 kcal/mol was reached. This protocol did not produce significant changes in the starting structure.

A model of the RT/**15a** complex was obtained by following a computational procedure similar to the one described above.

To check whether the RT-bound conformations of **8** and **15a** were energetically feasible, they were fully geometry optimized in vacuo. The resulting nearest local minimum conformers revealed to be more stable by 1.3 and, respectively, 3.1 kcal/ mol, meaning that the input geometries were not affected by steric conflicts.

HIV-1 RT RNA-dependent DNA Polymerase Activity Assay. Inhibition Assay. RNA-dependent DNA polymerase activity assay was assayed as follows: a final volume of 25  $\mu$ M contained reaction buffer (50 mM Tris-HCl pH 7.5, 1 mM DTT, 0.2 mg/mL BSA, 4% glycerol), 10 mM MgCl<sub>2</sub>, 0.5  $\mu$ g of poly(rA)oligo(dT)<sub>10:1</sub> (0.3  $\mu$ M 3'-OH ends), 10  $\mu$ M [<sup>3</sup>H]-dTTP (1 Ci/mmol), and 2–4 nM RT. Reactions were incubated at 37 °C for the indicated time. A total of 20  $\mu$ L aliquots were then spotted on glass fiber filters GF/C which were immediately immersed in 5% ice-cold TCA. Filters were washed twice in 5% ice-cold TCA and once in ethanol for 5 min and dried, and acid-precipitable radioactivity was quantitated by scintillation counting.<sup>31</sup>

Reactions were performed under the conditions described for the HIV-1 RT RNA-dependent DNA polymerase activity assay. Incorporation of radioactive dTTP into poly(rA)/oligo-(dT) was monitored in the presence of increasing amounts of the inhibitors to be tested. Data were then plotted according to Lineweaver–Burke and Dixon. For  $K_i$  determinations, an interval of inhibitor concentrations between  $0.2K_i$  and  $5K_i$  was used. Experiments have been done in triplicate. Experimental errors ( $\pm$ SD) were  $\leq$  10%.

**In Vitro Anti-HIV Assays. Cell Culture Assay. 1. CEM Cell Line.** The ability of the test compounds to protect HIV-1 infected T4 lymphocytes (CEM cells) from cell death was determined following the reported procedure.<sup>32</sup> All compounds were compared with a positive (AZT-treated) control performed at the same time under identical conditions.

**2.** Other Cells. Mature macrophages were obtained by incubating in 48-well plates (Costar, Cambridge, MA)  $10^6$  PBMC/mL of complete medium (containing Roswell Park Memorial Institute (RPMI)-1640, penicillin 100 units/mL, streptomycin 100 mg/mL, L-glutamine 0.3 mg/mL and 20% heat inactivated fetal calf serum). After 7 days of incubation in 5% CO<sub>2</sub> at 37 °C, nonadherent cells (lymphocytes) were removed by extensive washing. Using this method, the yield,

after removal of the nonadherent cells, was 10<sup>5</sup> macrophages per well. Test was conducted as described elsewhere.<sup>33</sup> Macrophages obtained with this method are >95% pure as detected by nonspecific esterase activity. C8166 is a CD4<sup>+</sup> T-cell line containing an HTLV-I genome of which only the tat gene is expressed.<sup>34</sup>

**3. Virus.** A laboratory lymphocyte-tropic strain of HIV-1 (HIV-1-IIIB) was used to infect C8166, while macrophages were infected with a laboratory monocyte-tropic strain of HIV-1 (HTLV-III-Ba-<sub>L</sub>, also called HIV-Ba-<sub>L</sub>).<sup>35,36</sup> Titration to determine the infectivity was performed in human macrophages as previously described. The titer of the virus stocks, expressed as 50% tissue culture infectious dose (TCID<sub>50</sub>), was determined as previously described.<sup>37</sup>

**4. Antiviral Agent.** Zidovudine (AZT) was purchased from Sigma Chimica.

**5. Toxicity.** Chemicals toxicity in C8166 was evalueted with a procedure involving a colorimetric assay (MTT assay) that monitors the ability of viable, but not dead, cells to reduce 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT) to a blue formazan product, which can be measured spectrophotometrically.<sup>38,39</sup>

**6. Assay of Antiviral Activity.** Antiviral activity of the tested compounds in acutely infected C8166 cultures was performed following an already described procedure.<sup>9</sup> The assay to evaluate anti-HIV drug efficacy in acutely infected mature macrophages has been previously described.<sup>33</sup> The antiviral activity of the compounds was assessed by measuring HIV-p24 antigen production in the supernatants of infected cultures as previously described<sup>35</sup> by using a commercially available HIV-antigen kit.

**7. Immunofluorescence Virus Binding Assay.** Calculation of the 50% effective dose ( $ED_{50}$ ) and 50% inhibitory dose ( $ID_{50}$ ) was performed. The  $ED_{50}$  and  $ID_{50}$  values were calculated from pooled values in the effective dynamic range of the antiviral and toxicity assays (5–95%) using the median effect equation as previously described.<sup>40</sup>

**Toxicity Tests. 1. Cell Lines.** All cell lines were obtained from ATCC. The cells were cultured in RPMI 1640 supplemented with 5% FCS, 0.1 mM glutamine, 1% penicillin, and streptomycin. Cells were grown in Nunc clone plastic bottles (TedNunc, Roskilde, Denmark) and split twice weekly at different cell densities according to standard procedure. 3T3 cells were grown as a monolayer and were split by using tripsin. Perypheral blood mononuclear cells (MNC) were separated from heparinized whole blood obtained from a healthy donor on a Fycoll-Hypaque gradient as previously described.<sup>41</sup> MNC thus obtained were washed twice with RPMI 1640 supplemented with 10% FCS, glutamine, and antibiotics, suspended at 200.000 viable cells/mL in medium containing, as mitogen, 5  $\mu$ g/mL PHA (Sigma) and used in toxicity tests.

**2. Chemicals.** MTT (3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide) was purchased from Aldrich. It was dissolved at a concentration of 5 mg/mL in sterile PBS at room temperature, and the solution was further sterilized by filtration and stored at 4 °C in a dark bottle. SDS was obtained from Sigma. Lysis buffer was prepared as follows: 20% w/v of SDS was dissolved at 37 °C in a solution of 50% of each DMF and demineralized water; pH was adjusted to 4.7 by adding 2.5% of an 80% acetic acid and 2.5% 1 N HCl solution.

**3. Toxicity Experiments.** Cells were plated at different concentrations on flat bottom 96-well microplates (0.1 mL/ well). Lymphocytes were plated out at 20 000 cells/well. 3T3 cells (murine fibroblast line) were plated at 10 000 cells/well. NSO cells (plasmocytoma murine cell line) were plated out at 3000 cells/well, and Daudi cells (human lymphoblastoid cell line) were plated at 300 cells/well. Twelve hours after plating, different concentrations of each compound were added to each well. After 48 h, MTT assay was performed to analyze cytotoxicity of the different compounds. Some experiments were performed by using confluent cells: compounds were added to the 3T3 monolayer 3 days after plating. Tests were then run as described above.

4. MTT/Formazan Extraction Procedure. A total of 20

 $\mu$ L of the 5 mg/mL stock solution of MTT was added to each well; after 2 h of incubation at 37 °C, 100  $\mu$ L of the extraction buffer was added. After an overnight incubation at 37 °C, the optical densities at 570 nm were measured using a Titer-Tech 96-well multiscanner, employing the extraction buffer as the blank.

**Synergy Calculations.** The multiple drug effect analysis of Chou and Talalay<sup>42</sup> was used to calculate combined drug effects.

**Drug Analysis.**<sup>8b</sup> Plasma and brain concentrations of the test compounds were determined by high-performance liquid chromatography (HPLC). An internal standard (IS) was used in each assay. Brain concentrations of **8** were not corrected for the compound contributed by residual blood because, in terms of concentrations, this amounted only to 5-10% of the brain concentrations.

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**Supporting Information Available:** Experimental details relative to synthesis and physical and chemical data of compounds **11–15**, to synergy calculations, and to drug administration and plasma and brain sampling/drug analysis, and a table with elemental analyses of compounds **8**, **12a–g**, **15a–c**, **18a**,**b** and **19**. This material is available free of charge via the Internet at http://pubs.acs.org.

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