

Titanium-Carbide MXenes for Work Function and Interface Engineering in Perovskite Solar Cells

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ABSTRACT

In order to improve the efficiency of perovskite solar cells (PSCs), careful device design and tailored interface engineering are needed to enhance optoelectronic properties and the charge extraction process at the selective electrodes. Here, we use two-dimensional transition metal carbides (the MXene $Ti_3C_2T_x$) with various termination groups (T_x) to tune the work function (WF) of the perovskite absorber and the TiO_2 electron transport layer (ETL), and to engineer the perovskite/ETL interface. Ultraviolet photoemission spectroscopy measurements and Density Functional Theory calculations show that the addition of $Ti_3C_2T_x$ to halide perovskite and TiO_2 layers permits to tune the materials' WFs, without affecting other electronic properties. Moreover, the dipole induced by the $Ti_3C_2T_x$ at the perovskite/ETL interface can be used to change the band alignment between these layers. The combined action of WF tuning and interface engineering can lead to substantial performance improvements in MXene-modified PSCs, as shown by the 26% increase of power conversion efficiency and hysteresis reduction with respect to reference cells without MXene.

The rapid development of perovskite solar cells (PSCs) carried out in the last decade demonstrated the potentiality of this PV technology to compete on equal footing with traditional inorganic PV or to work in synergy with established silicon technology in tandem cell configuration.¹ The remarkable efforts devoted to optimize the PSCs in term of device architecture², perovskite composition³ and charge collecting electrodes⁴ produced a large wallet of emerging materials comprising organic polymers,⁵ oxides,⁶ two-dimensional (2D) materials⁷ and others⁸ able to tune the device properties⁹ and eventually to boost their performance and stability.¹⁰ Usually, a typical PSC is composed of a perovskite active layer, sandwiched between two selective charge transport layers (CTLs) and electrodes for negative and positive charge extraction. These are selected carefully by taking into account their compatibility with the underneath layers,¹¹ their charge mobility¹² and the energy level alignment with the perovskite absorber.¹³ However, being PSCs composed of several layers, interfaces ultimately play a crucial role in ruling device performance and stability. Charge transfer at the interfaces,¹⁴ interface band alignment,¹⁵ interfacial vacancies,¹⁶ defects due to poor adhesion between layers¹⁷ and energy barriers have a strong impact on electrical device characteristics. From here, the pivotal role that interface engineering (IE)¹⁸ has recently gained in the PSCs field.¹⁹ In this regard, 2D materials can be inserted as inter²⁰ or buffer layers²¹ by modifying the chemical/physical properties of involved layers at

the interface and eventually to improve the charge injection/collection at perovskite/CTLs interfaces. One of the main potentialities of 2D materials is the possibility to easily tailor their electronic structure, such as work function (WF)^{22,23} or band gap, by proper functionalization²⁴ or by quantum confinement.²⁵ Fine-tuning of the WF allows to obtain appropriate energy level alignment leading to an ideal energy offset between perovskite active layer and CTLs, eventually inducing a built-in potential for efficient charge collection at the electrodes. Recently, a new family of 2D transition metal carbides, nitrides and carbonitrides (MXenes)²⁶ with a general formula $M_{n+1}X_nT_x$ ($n = 1, 2, 3$), where M represents an early transition metal, X is a carbon and/or nitrogen atom and T_x the surface-terminating functional groups came out as a promising class of 2D materials in many applications^{26,27} owing to their physical and chemical properties.^{26,28–30} Rich chemistry and surface termination make MXenes unique 2D materials with huge possibilities to tune their electronic properties. In fact, MXenes offer the possibility to vary the WF by choosing the proper transition metal as well as the X element.³¹ Moreover, during the synthesis of MXenes, their surfaces are naturally functionalized, which changes the electrostatic potential near the surfaces and affects the electronic structure, significantly shifting the WF.³² As density functional theory (DFT) predicts, surface termination strongly influences the density of states³³ and the WF of MXenes³⁴ which can range from 1.6 eV (for OH-termination) to 6.25 eV (for O-termination).^{34,35} This opens new opportunities for MXenes applications in optoelectronics and in particular in photovoltaics where already some initial studies have been presented for organic solar cells³⁶, Si solar cells,³⁷ dye-synthesized solar cells³⁸ and PSCs.^{39, 40} In the PSC case, $Ti_3C_2T_x$ MXenes have been incorporated into the perovskite absorber showing an improved morphology and an enhanced PCE (+12%) with respect to the reference cell without MXenes³⁹ or into the SnO_2 electron transporting layer (ETL)⁴⁰ to provide superior charge transfer paths that permits to enhance the PCE (+6.5%) with respect to the reference cells. In order to reveal the role of MXenes in perovskites and provide a clear strategy on the use of MXenes for WF and IE in PSCs, we synthesized $Ti_3C_2T_x$ MXene and performed an extensive characterization of the MXene structure and electronic properties (**Figure 1**).

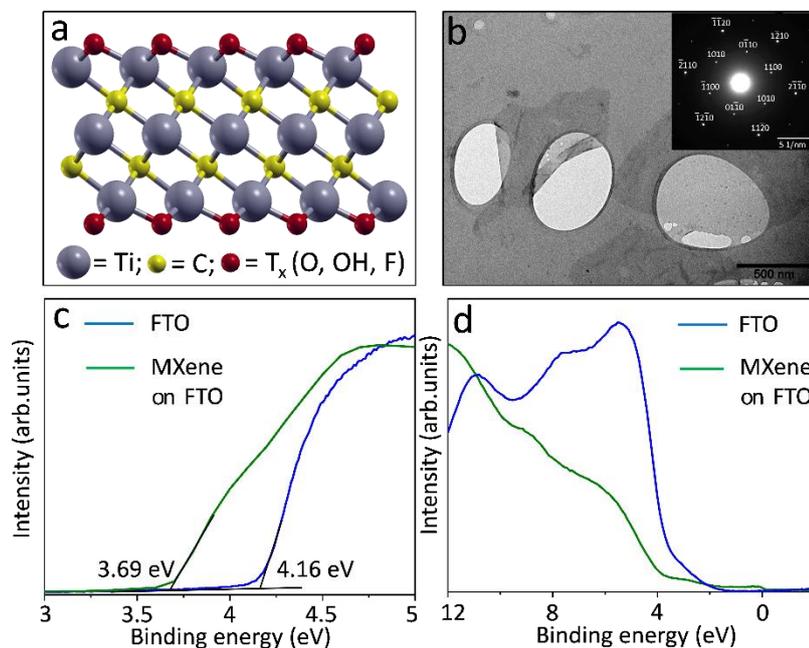


Figure 1: Characterization of $Ti_3C_2T_x$ MXene. **a**, Schematic structure of $Ti_3C_2T_x$ MXene. Surface terminations (T_x) are a mixture of F, O, and OH. **b**, TEM image of $Ti_3C_2T_x$ MXene flakes. The corresponding selected area electron diffraction (SAED) pattern is reported in the inset. UPS spectra measured with photon energy of 40.81 eV on the MXene flakes and FTO substrate supporting them are reported in left panel **c** and right panel **d** showing secondary electron cut-off and valence band region, respectively.

After chemical etching and exfoliation by sonication, single-layer $\text{Ti}_3\text{C}_2\text{T}_x$ nanosheet (**Figure 1a**) consists of two carbon atoms which bind three titanium ones as elementary units. The outer Ti layers tend to terminate with functional groups (Tx) such as O, OH, and F (**Figure 1a**), which are randomly distributed on the MXene surface. The MXene structure is characterized by the presence of single atomically-thin transparent flakes of 2D titanium carbides with hexagonal symmetry which follows the parent MAX phase after exfoliation (**Figure 1b**). $\text{Ti}_3\text{C}_2\text{T}_x$ sheets have irregular edges and size distribution ranging generally from 1.5 to 2.5 μm . Details about MXene composition were extracted by XPS measurements reported in **Figures S1 and S2a**. In particular, the ratio between the F : OH : O functional groups is estimated to be 1.6 : 0.65 : 0.34. Based on these results, it can be concluded that the MXene surface randomly ends with the F, OH and O groups, with a prevalence of fluorine functional groups.

The electronic properties of $\text{Ti}_3\text{C}_2\text{T}_x$ was featured by WF measurement using ultraviolet photoelectron spectroscopy (UPS). We found that the WF of synthesized MXene flakes deposited on FTO glass is rather low and equal to 3.7 eV (**Figure 1c**). The spectra measured in the valence band region (**Figure 1d**) show that the intensity extends up to the Fermi level and severely increases for binding energy (BE) close to 4 eV, likely due to the contribution of the oxidized phases detected in the XPS spectra (see **Figure S1a,b**). In general, the reported UPS measured WF values for $\text{Ti}_3\text{C}_2\text{T}_x$ MXene range from 3.4 eV⁴¹ to 4.62 eV.⁴² Such large modulation of the WF could be attributed to the different etching environment dramatically affecting the surface termination. The WF value of synthesized MXene is in agreement with the XPS results and should be due to the substantial percentage of hydroxyl groups terminating the $\text{Ti}_3\text{C}_2\text{T}_x$ matrix. This finding is supported by theoretical calculations³² (see also the discussion in SI) and experimental observations reported in literature.⁴¹

The low WF of MXene flakes could be exploited to finely control the energy level alignment between the perovskite absorber layer and the CTLs. To this end, we probed the effect of the MXene employed as additive in perovskite and TiO_2 layers onto their WFs by combining UPS (**Figure 2**) and XPS (**Figure S2b**) measurements.⁴³ The WF of pristine perovskite, determined from the secondary electron onset, is 4.72 eV, which is shifted to 4.37 eV after the addition of MXene (see **Figure 2a**), i.e. 0.35 eV lower than that of the undoped perovskite. The valence band spectrum measured on the MXene-doped perovskite film and the energy gap appear quite like the pristine ones. (**Figure 2b and S3**). The energy diagram in **Figure 2c** schematizes the WF and Ionization Energy (IE) reduction of MXene-doped perovskite.

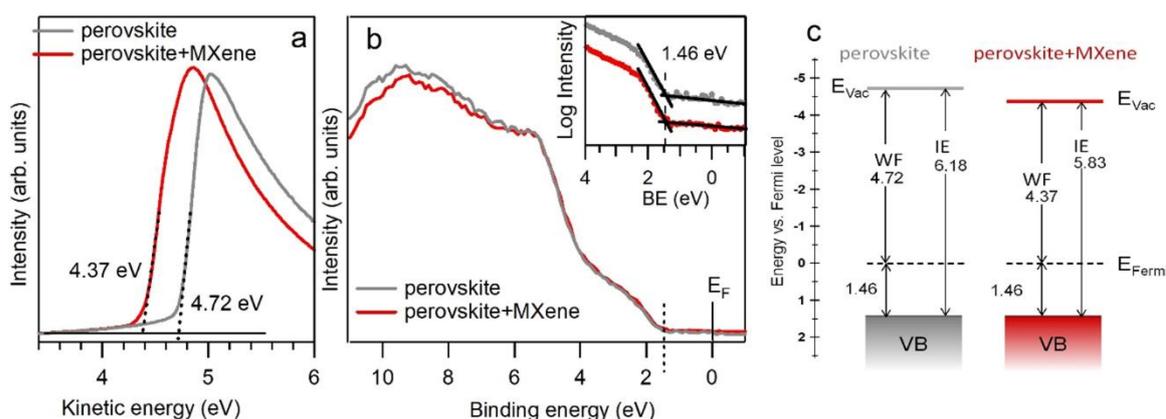


Figure 2: UPS curves of pristine and MXene-doped perovskite films. **a**, UPS spectra around the secondary electron cut-off. **b**, UPS spectra in the valence band region. For the pristine perovskite, the valence band maximum (VBM), determined by the intercept to zero of the intensity plotted in logarithmic scale (see inset in panel **b**) is at 1.46 eV below the Fermi level, in good agreement with previous findings.⁴³ **c**, Energy scheme for undoped and MXene-doped perovskite with respect to the Fermi level.

We exclude that in perovskite deposition the addition of MXene can change the perovskite crystal growth and the resulting morphology since no significant changes have been observed in the scanning electron microscopy (SEM) images reported in SI (**Figure S4**).

UPS characterizations were performed also for the TiO₂ ETL with and without addition of MXene (**Figure S5**). In this case, MXene doping leaves the shape and the maximum of the valence band almost unchanged. At the same time, the WF slightly decreases from 3.91 eV for the TiO₂ sample to 3.85 eV in the case of MXene-doped TiO₂. The small WF differences between the pristine and doped TiO₂ layers compared to the WF shift induced by MXene in perovskite can be related to the oxidation of Ti₃C₂T_x MXene (**Figure S6**). However, the WF decrease observed in the doped TiO₂ layer suggests that MXene does not oxidize completely when annealed with TiO₂ as confirmed by the XRD pattern measured for sintered MXene-doped TiO₂ (**Figure S7**). In order to understand the origin of the perovskite WF change induced by the interaction with MXene, we performed DFT calculations of the perovskite/Ti₃C₂T_x with OH and O termination. To make this calculation feasible, we consider the single cation CH₃NH₃PbI₃ (MAPbI₃) perovskite having experimentally verified that MXene doping induces a WF shift also for this perovskite (**Figure S8**). Calculation details are reported in SI (see **Figures S9** and **S10**). DFT calculations reported in **Figure 3** show that the charge transfer at the MAPbI₃/Ti₃C₂(OH)₂ interface induces the formation of an interface dipole, causing an important reduction of the WF and affecting the band alignment of the system. The transferred charge density is localized at the very first Pb-I layer, therefore involving less than 1 nm of the surface. The formation of an interface dipole can also be deduced from the vacuum potential slopes observed in **Figure 3a** and **3b**. In the case of OH terminated MXene, the slope is higher than O termination, implying a larger interface dipole. Moreover, in MAPbI₃/Ti₃C₂O₂ the potential slope (i.e. the interface dipole) has the opposite direction with respect to OH termination case. **Figure 3c** and **3d** show the projected bands of the MAPbI₃/MXene slabs, where the red colour represents the contribution from the inner atoms of the MAPbI₃ slab, falling in the grey boxes drawn in **Figure 3a** and **3b**.

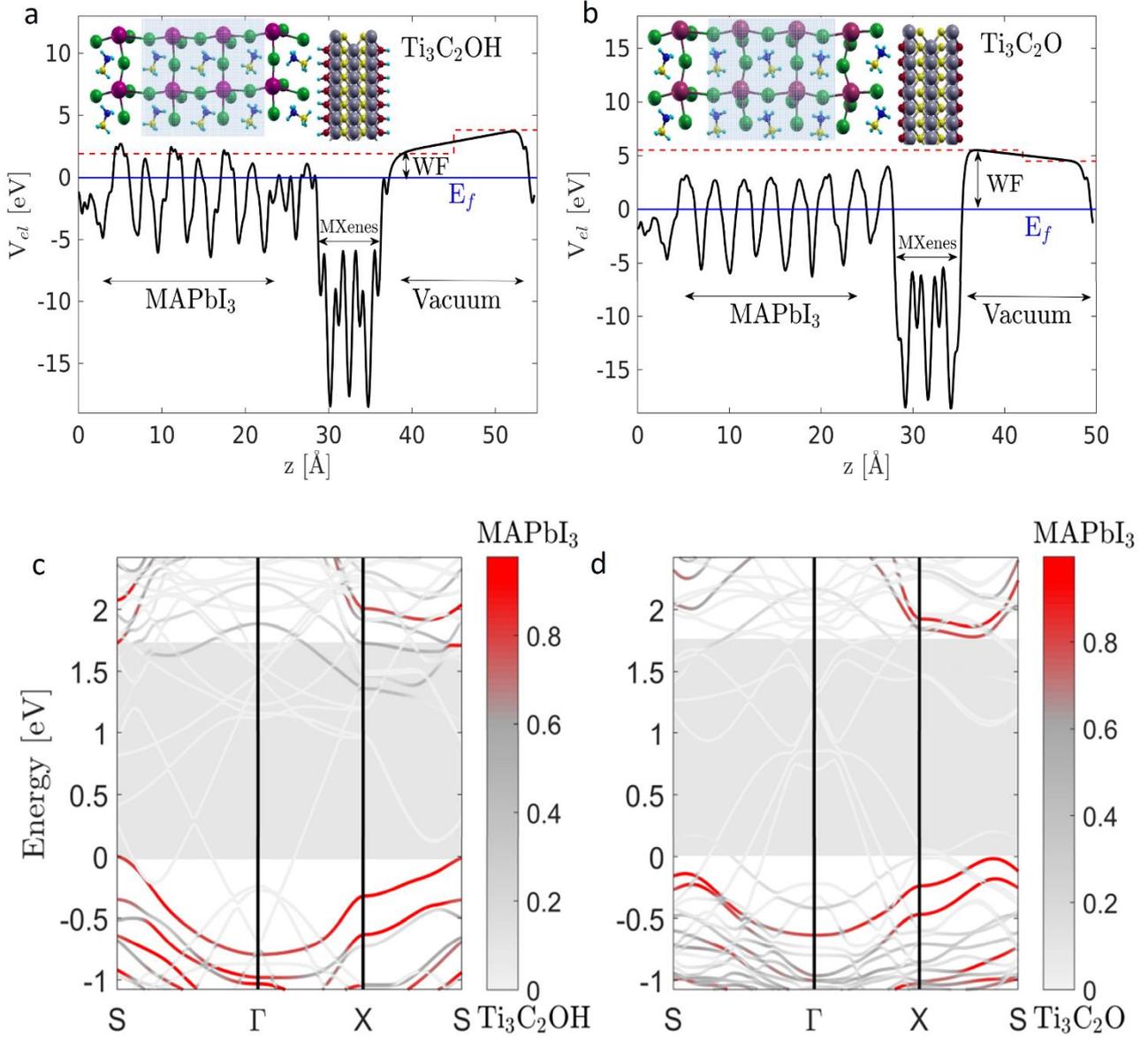


Figure 3: DFT calculation of the MAPbI₃/MXene structure. **a, b** Electrostatic potential averaged over planes perpendicular to the MAPbI₃/Ti₃C₂(OH)₂ and MAPbI₃/Ti₃C₂O₂ interface, respectively. The computed structures are shown within the plots, where green, magenta, blue, yellow, cyan, grey and red spheres represent I, Pb, N, C, H, Ti and O atoms, respectively. The red dashed lines represent the dipole corrected vacuum levels. The Fermi energy is set to zero, so that the vacuum potential, just away from the MXene surface, corresponds to the WF, of the MAPbI₃/Ti₃C₂T_x interface, as depicted in the panels. We can see that the WF derived for the OH terminated MXene configuration, $WF_{OH} \cong 2.1$ eV, is substantially smaller than the value obtained for the O terminated structure, where $WF_O \cong 5.7$ eV. Notably, similar calculations for the F terminated MXene do not show a significant variation of MAPbI₃ WF. **c, d**, Projected band structures of the MAPbI₃/MXene slabs for OH and O termination of the MXene, respectively. Contribution from the bulk part of the MAPbI₃ slab (grey box in panels **a, b**) are coloured in red. The valence band edge is set to zero. The bulk band gap of the MAPbI₃ is indicated by the shaded area.

A remarkable outcome emphasised in **Figures 3c** and **3d** is that, even in such small structures, the MAPbI₃ bandgap is barely affected by the interaction with MXene, remaining at 1.7 eV. This gap independence is a feature of MXene additives also seen in experiments. We conclude that the interaction is mainly electrostatic, owing to the dipole formation at the interface. A similar result is obtained when considering a symmetric

structure MXene/MAPbI₃/MXene (see **Figure S11**), demonstrating that the WF shift of perovskite also occurs when MXene is found at both ETL and HTL interfaces or within the bulk of the material, decreasing its averaged WF. In our calculations we have assumed that different cations (MA- methylammonium, FA- formamidinium, Cs- cesium) behave similarly, although they affect the lattice constants and the bandgap. For instance, in the case of FA, shown in **Figure S12**, the WF reduces from 4.5 eV of bare perovskite to 1.7 eV, in contrast to 2.1 eV obtained for MA. On the contrary, we expect that the WF of CsPbI₃+Ti₃C₂OH will be larger, owing to a larger bandgap that results in a smaller interfacial polarization. Our multication perovskite have nominal fractions FA=78.85%, MA=16.15%, Cs=5%, hence Cs plays a minor role.

We also observe that the WF reduction induced by the OH terminated surface is much more pronounced than the WF increase induced by O terminations. This is due to a rather strong non-linear behaviour already observed in literature³⁵ and also obtained in our calculations. Indeed, the perovskite WF using MXene with a OH:O termination ratio of 50:50 reduces to 3.1 eV, whereas for a ratio of 75:25 it is close to 1.9 eV, namely close to the value obtained for 100% OH. Based on the XPS data reported above (**Figure S1,S2**), which estimate 62% F, 25% OH and 13% O terminations, whereas the plain OH:O ratio would determine a $\Delta WF \sim 2$ we expect the reduction of WF to be 4 times smaller due to the abundance of fluorine, hence $\Delta WF \cong -0.5$ eV, much closer the value of -0.35 eV experimentally observed. Similar calculations reported in SI show the impact of MXene on the WF of TiO₂ (see **Figure S13**). Even in this case, the WF obtained for the anatase (101) surface with OH terminated MXene, $WF_{OH} \cong 1.6$ eV, is smaller than that with O terminated MXene, where $WF_O \cong 5.8$ eV.

Five different PSC structures were designed and characterized to investigate the impact of Ti₃C₂T_x MXene. In particular, the “reference” device has a typical layer sequence with compact (cTiO₂) and mesoporous (mTiO₂) TiO₂, namely glass/FTO/cTiO₂/mTiO₂/perovskite/spiro-OMeTAD/Au, while the other structures include MXene as a dopant or as interlayer as summarized in Table 1. The comparison of cross-section SEM images for reference and full-MXene device (type C), reported in **Figure S14**, exclude a direct impact of MXene on the device layer morphology.

Name	Device Structure
Reference	Glass/FTO/cTiO ₂ /mTiO ₂ /perovskite/spiro-OMeTAD/Au
Type A	Glass/FTO/cTiO ₂ /mTiO ₂ /perovskite+MXene/spiro-OMeTAD/Au
Type B	Glass/FTO/cTiO ₂ +MXene/mTiO ₂ +MXene/perovskite+MXene/spiro-OMeTAD/Au
Type C	Glass/FTO/cTiO ₂ +MXene/mTiO ₂ +MXene/MXene/perovskite+MXene/spiro-OMeTAD/Au
Type D*	Glass/FTO/cTiO ₂ /mTiO ₂ +MXene/MXene/perovskite+MXene/spiro-OMeTAD/Au

Table 1: Structure of the investigated PSCs. All device types (A, B, C and D) are based on MXene-doped perovskite. Type B and C devices include also MXene doping in cTiO₂ and mTiO₂. The structure C has an additional MXene interlayer at the mTiO₂/perovskite interface. Devices of type D, similar to C but with a standard cTiO₂ layer, have been fabricated to identify the role of MXene in the cTiO₂ layer. *Discussions related to structure D are reported in SI.

As a main result, all device structures employing MXene (type A, B, C **Figure 4** and D **Figure S15**) show an average improved PCE with respect to reference cells (+26.5% for type C) mainly due to an average improvement in V_{oc} (+5.7%) and FF (+13.7) with respect to the reference device. A less marked, but still important improvement in averaged J_{sc} was also observed (+4.8% for type C).

A remarkable efficiency of 20.14% is achieved in the case of full-MXene structure (Type C) with FF=77.6%, J_{sc}=23.82 mA/cm² and V_{oc}=1.09 V. Notably Type C PSCs exhibit a very good stabilized power under prolonged 1 SUN irradiation at maximum power point (MPP), as reported in **Figure S16a**. The incident photon to current conversion efficiency (IPCE) spectrum for Type C devices is reported in **Figure S16b**, together with the integrated short circuit current, while the related J-V characteristics are reported in **Figure S17a**. As shown by the J-V characteristics, a clear benefit in the use of MXene in PSC consists in a significant reduction of the J-V hysteresis, particularly relevant in the case of type C. The hysteresis index (HI), calculated as the ratio

between the PCE extracted from reverse J-V scan (open circuit-> short circuit) and the one from the forward scan (short circuit-> open circuit), reduces in average from 1.22 of the reference cell to 1.04 of the Type C MXene-based PSC (see **Figure S17b**). In addition, type C and reference devices were exposed to continuous 1 SUN irradiation for 30 minutes. While the type C device showed stabilized power output at MPP (P_{MPP}) after 25 minutes by retaining more than 83% of initial value (see **Figure S17c**), the reference cell demonstrated a monotonic P_{MPP} decrease trend after 10 minutes. The stabilization effect due to the MXene can be related to the improved charge extraction, since trapped charge at the interfaces are well-known to trigger the degradation.⁴⁴

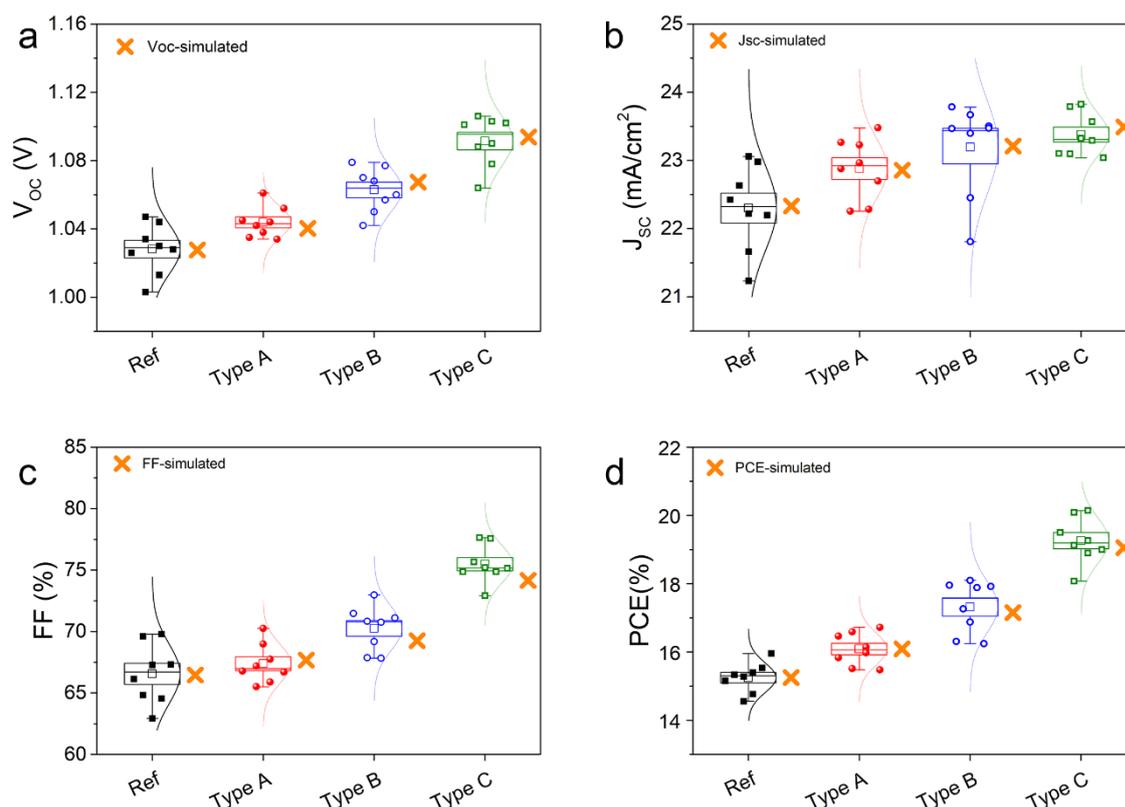


Figure 4: Photovoltaic parameter statistics for the investigated PSCs. **a**, Open circuit voltage (V_{oc}). **b**, Short circuit current density (J_{sc}). **c**, Fill factor (FF). **d**, Power conversion efficiency (PCE). Parameters are extracted from the J-V curves acquired at 1 SUN irradiation. The standard error (SE) is represented with a box while the average value is depicted as an empty squared dot. For each PSC structure, 8 devices have been fabricated with a cell active area of 0.09 cm². Results for device simulations are also displayed as orange crosses.

To further elucidate the role of MXene in each PSC structure, we performed transient photocurrent (TPC) and photovoltage (TPV) measurements combined with light power dependency of J_{sc} and V_{oc} , photoluminescence spectra, charge carrier lifetime and charge extraction measurements (**Figure S18**). The insertion of MXene within both mTiO₂ and perovskite layer reduces charge recombination rate at the TiO₂/perovskite interface (see SI and **Figures S18a, S18b, S18c and S18d** for details) while the MXene interlayer is able to reduce trap state density and/or passivate the mTiO₂ surface (see SI and **Figures S18e, S18f**). Finally, when MXene is added to cTiO₂ layer, energy barrier at the cTiO₂/mTiO₂+MXene interface is significantly reduced, by improving the electron collection at the photoelectrode (**Figure S19**).

In order to support the conclusion that the improvement of PSC performance are directly related to WF tuning, optimized interface alignment and reduction of charge recombination induced by MXene, we performed device simulations based on a drift-diffusion model⁴⁵.

According to the experimental devices, we model the reference solar cell as a FTO/cTiO₂/mTiO₂+perovskite/Spiro-OMeTAD/Au stack, where the mTiO₂ scaffold and perovskite are

modelled as an effective 550 nm absorbing perovskite layer with band edges at $E_C = -3.9$ eV and $E_V = -5.5$ eV.⁴⁶ Cell types A, B and C have been modelled by shifting the WFs of the MXene-doped layers according to the UPS measurements presented in **Figure 2** and **Figure S5**. In particular, we applied a rigid upshift of the band-edges of perovskite (type A) and perovskite and cTiO_2 layers (types B and C) of 0.35 eV and 0.05 eV, respectively. For device type C we assumed that the MXene interlayer, owing to its low WF, further improves the band alignment at TiO_2 /perovskite interface. This has been modelled by an additional upwards shift of the band edge of cTiO_2 . Additionally, we assumed a slightly smaller interface recombination at the TiO_2 /perovskite interface for type C compared to type B, based on the measured lifetimes and on the light intensity dependence of the V_{OC} (see **Figure S18c**, **Figure S18d** and discussion in SI). The full list of parameters employed in the simulations is reported in **Table S1**, while energy band profiles for the simulated device structures are reported in **Figure S20**. By comparing J-V simulations with experimental results (**Figure S21**) and simulated PV parameters with experimental ones (**Figure** and **Figure S22**), we can deduce that the proposed model is able to consistently reproduce experimental results for all PSC types.

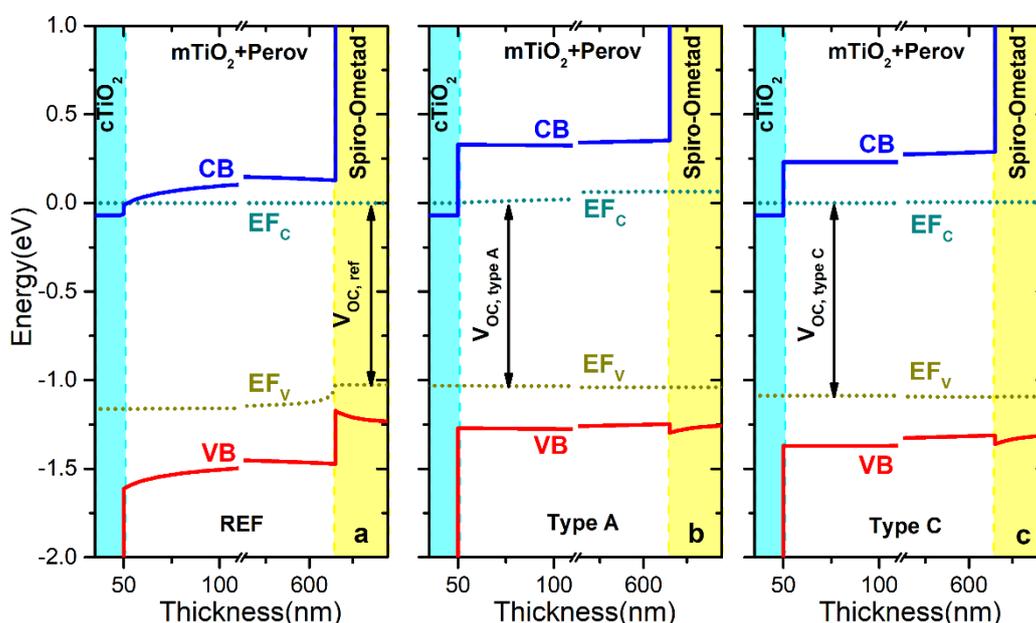


Figure 5: Band profiles of PSCs with and without MXene as obtained by physical simulation modelling. Panels show the Conduction Band (CB), Valence Band (VB) profiles at V_{OC} between corresponding quasi Fermi levels for CB (EF_C) and VB (EF_V). **a**, Reference PSC. **b**, Type A PSC. **c**, Type C PSC.

Figure 5 shows the band profiles and quasi Fermi levels for the reference (**Figure 5a**), Type A (**Figure 5b**) and Type C (**Figure 5c**) PSCs at open circuit conditions. The barrier at the perovskite/spiro-OMeTAD interface found in the valence band of the reference device, which is responsible for the V_{OC} drop, reduces in MXene-doped perovskite for both type A and Type C PSCs. The reduced barrier at the perovskite/spiro-OMeTAD interface tends to increase the V_{OC} , however in Type A PSC the up-shift of the perovskite bands increase the band discontinuity at the TiO_2 /perovskite interface with a consequent penalization of the final V_{OC} . This problem is partially solved in Type C (**Figure 5c**) where the additional up-shift of the TiO_2 band induced by the MXene doping and MXene interlayer reduces the TiO_2 /perovskite barrier and improves the V_{OC} . We should point out that the simulations with band shifting induced by the $\text{Ti}_3\text{C}_2\text{T}_x$ MXene cannot justify the increase of the J_{SC} observed experimentally (see **Figure 4b** and **Figure S22**). However, MXene-doped perovskite showed an increased light absorption and IPCE with respect to pristine perovskite (see **Figure S23**). We can speculate that this could be related to scattering⁴⁷ or plasmonic resonance⁴⁸ induced by MXene into the perovskites even though a clear picture for this phenomenon has not been identified. Accounting for this absorption coefficient increase in the simulations, we are able to recover the increase of J_{SC} (**Figures S21 and S22**). The increase of FF for Type C PSCs suggests a change of the transport properties at the

TiO₂/perovskite interface induced by the MXene interlayer (see **Figure S18** and discussion in SI). By modelling this effect with an increase of carrier mobility in the effective mTiO₂/perovskite layer we are able finally to recover also the increase of FF (**Figure S22**), as demonstrated by comparing J-V simulations with experimental results (**Figures S21 and S22**).

OUTLOOK

This work focuses on the demonstration that MXenes, in particular the Ti₃C₂T_x, owing to their WF tunability controlled by the surface termination groups can be exploited to modify the WF of perovskite and transporting layers with a consequent optimization of band alignments in PSCs. The WF shift has been observed for both single and multi-cation perovskite doped with Ti₃C₂T_x MXene by demonstrating the general applicability of our approach. We also show that MXene can be effectively employed to tune the interface between perovskite absorber and TiO₂ ETL to further enhance charge transfer between the two layers. By combining MXene-doped layers and MXene-engineered interfaces we are able to demonstrate a strong improvement of PSC efficiency (+26%) with respect to the reference cell, achieving a final maximum efficiency exceeding 20% and an almost complete suppression of the cell hysteresis. The possibility to vary on demand the WF of materials and tuning their band alignments with other layers forming an electronic device is of fundamental importance to enlarge the design parameter space and to improve device performance. MXenes have the potentialities to be a key player in WF tuning of perovskite absorbers (**Figure S24**) and CTLs in PSCs, without affecting other properties of the materials. Owing to this unique property, we believe the MXene WF tuning and MXene interface engineering developed in this work can inspire innovative efficient designs of PSCs and other perovskite based devices such as LED and detectors.

AUTHORS CONTRIBUTIONS

An.P, ADC, DS, and DVK conceived the work. AA and SP performed the experiments on solar cells and the electro-optical characterizations. An.P produced and characterized the MXenes. ADV, DR, Al.P and MA performed the theoretical simulations. RL and AL performed UPS and XPS analysis. ADC coordinated the research activity. The manuscript was written through the contributions of all the authors. All the authors have given approval to the manuscript final version.

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REFERENCES

1. Cai, M. *et al.* Cost-Performance Analysis of Perovskite Solar Modules. *Adv. Sci.* **4**, 1600269 (2016).
2. Hussain, I. *et al.* Functional materials, device architecture, and flexibility of perovskite solar cell. *Emergent Mater.* **1**, 133–154 (2018).
3. Kim, H.-S., Hagfeldt, A. & Park, N.-G. Morphological and Compositional Progress in Halide Perovskite Solar Cells. *ChemComm* **55**, 1192–1200 (2019).
4. Ameen, S., Akhtar, M. S., Shin, H.-S. & Nazeeruddin, M. K. Charge-Transporting Materials for Perovskite Solar Cells. *Adv. Inorg. Chem.* **72**, 185–246 (2018).
5. Fakharuddin, A. *et al.* Perovskite-Polymer Blends Influencing Microstructures, Nonradiative

- Recombination Pathways, and Photovoltaic Performance of Perovskite Solar Cells. *ACS Appl. Mater. Interfaces* **10**, 42542–42551 (2018).
6. Mingorance, A. *et al.* Interfacial Engineering of Metal Oxides for Highly Stable Halide Perovskite Solar Cells. *Adv. Mater. Interfaces* **1800367**, 1–10 (2018).
 7. You, P., Tang, G. & Yan, F. Two-dimensional materials in perovskite solar cells. *Mater. Today Energy* **11**, 128–158 (2019).
 8. Li, T. *et al.* Additive engineering for highly efficient organic-inorganic halide perovskite solar cells: Recent advances and perspectives. *J. Mater. Chem. A* **5**, 12602–12652 (2017).
 9. Chen, K., Schünemann, S., Song, S. & Tüysüz, H. Structural effects on optoelectronic properties of halide perovskites. *Chem. Soc. Rev.* **47**, 7045–7077 (2018).
 10. Yang, S., Fu, W., Zhang, Z., Chen, H. & Li, C. Z. Recent advances in perovskite solar cells: Efficiency, stability and lead-free perovskite. *J. Mater. Chem. A* **5**, 11462–11482 (2017).
 11. Isabelli, F. *et al.* Solvent systems for industrial-scale processing of Spiro-OMeTAD hole transport layer in perovskite solar cells. *ACS Appl. Energy Mater.* **1**, 6056–6063 (2018).
 12. Wang, Z. K. & Liao, L. S. Doped Charge-Transporting Layers in Planar Perovskite Solar Cells. *Adv. Opt. Mater.* **6**, 1–13 (2018).
 13. Courtier, N. E., Cave, J. M., Foster, J. M., Walker, A. B. & Richardson, G. How transport layer properties affect perovskite solar cell performance: insights from a coupled charge transport/ion migration model. *Energy Environ. Sci.* **12**, 396–409 (2019).
 14. Fakhruddin, A., Schmidt-Mende, L., Garcia-Belmonte, G., Jose, R. & Mora-Sero, I. Interfaces in perovskite solar cells. *Adv. Electron. Mater.* **7**, 1–44 (2017).
 15. Wang, S., Sakurai, T., Wen, W. & Qi, Y. Energy Level Alignment at Interfaces in Metal Halide Perovskite Solar Cells. *Adv. Mater. Interfaces* **5**, 1–30 (2018).
 16. Saidaminov, M. I. *et al.* Suppression of atomic vacancies via incorporation of isovalent small ions to increase the stability of halide perovskite solar cells in ambient air. *Nat. Energy* **3**, 648–654 (2018).
 17. Lee, I., Yun, J. H., Son, H. J. & Kim, T. S. Accelerated Degradation Due to Weakened Adhesion from Li-TFSI Additives in Perovskite Solar Cells. *ACS Appl. Mater. Interfaces* **9**, 7029–7035 (2017).
 18. Agresti, A. *et al.* Graphene Interface Engineering for Perovskite Solar Modules: 12.6% Power Conversion Efficiency over 50 cm² Active Area. *ACS Energy Lett.* **2**, 279–287 (2017).
 19. Petridis, C., Kakavelakis, G. & Kymakis, E. Renaissance of graphene-related materials in photovoltaics due to the emergence of metal halide perovskite solar cells. *Energy Environ. Sci.* **11**, 1030–1061 (2018).
 20. Taheri, B. *et al.* Graphene-engineered automated sprayed mesoscopic structure for perovskite device scaling-up. *2D Mater.* **5**, (2018).
 21. Arora, N. *et al.* Perovskite solar cells with CuSCN hole extraction layers yield stabilized efficiencies greater than 20%. *Science* **358**, 768–771 (2017).
 22. Konios, D. *et al.* Highly efficient organic photovoltaic devices utilizing work-function tuned graphene oxide derivatives as the anode and cathode charge extraction layers. *J. Mater. Chem. A* **4**, 1612–1623 (2016).
 23. Agresti, A. *et al.* Efficiency and Stability Enhancement in Perovskite Solar Cells by Inserting Lithium-Neutralized Graphene Oxide as Electron Transporting Layer. *Adv. Funct. Mater.* **26**, 2686–2694 (2016).

24. Agresti, A. *et al.* Two-dimensional (2D) Material Interface Engineering for Efficient Perovskite Large-area Modules. *ACS Energy Lett.* **4**, 1862–1871 (2019).
25. Najafi, L. *et al.* MoS₂ Quantum Dot/Graphene Hybrids for Advanced Interface Engineering of a CHNH₃PbI₃ Perovskite Solar Cell with an Efficiency of over 20%. *ACS Nano* **12**, 10736–10754 (2018).
26. Hantanasirisakul, K. & Gogotsi, Y. Electronic and Optical Properties of 2D Transition Metal Carbides and Nitrides (MXenes). *Adv. Mater.* **1804779**, 1–30 (2018).
27. Naguib, M., Mochalin, V. N., Barsoum, M. W. & Gogotsi, Y. 25th Anniversary Article : MXenes : A New Family of Two-Dimensional Materials. *Adv. Mater.* **26**, 992–1005 (2013).
28. Zhang, C. J. *et al.* Transparent, Flexible, and Conductive 2D Titanium Carbide (MXene) Films with High Volumetric Capacitance. *Adv. Mater.* **29**, 1–9 (2017).
29. Lipatov, A. *et al.* Elastic properties of 2D Ti₃C₂T_x MXene monolayers and bilayers. *Sci. Adv.* **4**, eaat0491 (2018).
30. Akuzum, B. *et al.* Rheological Characteristics of 2D Titanium Carbide (MXene) Dispersions : A Guide for Processing MXenes. *ACS Nano* **12**, 2685–2694 (2018).
31. Khazaei, M., Ranibar, A., Arai, M., Sasaki, T. & Yunoki, S. Electronic properties and application of MXenes: a theoretical review. *J. Mater. Chem. C* **5**, 2488–2503 (2017).
32. Khazaei, M. *et al.* OH-terminated two-dimensional transition metal carbides and nitrides as ultralow work function materials. *Phys. Rev. B - Condens. Matter Mater. Phys.* **92**, 1–10 (2015).
33. Hu, T. *et al.* Chemical Origin of Termination-Functionalized MXenes: Ti₃C₂T₂ as a Case Study. *J. Phys. Chem. C* **121**, 19254–19261 (2017).
34. Liu, Y., Xiao, H. & Goddard, W. A. Schottky-Barrier-Free Contacts with Two-Dimensional Semiconductors by Surface-Engineered MXenes. *J. Am. Chem. Soc.* **138**, 15853–15856 (2016).
35. Schultz, T. *et al.* Surface termination dependent work function and electronic properties of Ti₃C₂T_x MXene. *Chem. Mater.* acs.chemmater.9b00414 (2019). doi:10.1021/acs.chemmater.9b00414
36. Yu, Z. *et al.* MXenes with tunable work functions and their application as electron- and hole-transport materials in non-fullerene organic solar cells. *J. Mater. Chem. A* (2019). doi:10.1039/C9TA01195A
37. Fu, H. C. *et al.* MXene-Contacted Silicon Solar Cells with 11.5% Efficiency. *Adv. Energy Mater.* **1900180**, 1–9 (2019).
38. Dall’Agnese, C., Dall’Agnese, Y., Anasori, B., Sugimoto, W. & Mori, S. Oxidized Ti₃C₂ MXene nanosheets for dye-sensitized solar cells. *New J. Chem.* **42**, 16446–16450 (2018).
39. Guo, Z. *et al.* High Electrical Conductivity 2D MXene Serves as Additive of Perovskite for Efficient Solar Cells. *Small* **1802738**, 1–8 (2018).
40. Yang, L. *et al.* SnO₂-Ti₃C₂ MXene electron transport layers for perovskite solar cells. *J. Mater. Chem. A* (2019). doi:10.1039/C8TA12140K
41. Peng, C. *et al.* High efficiency photocatalytic hydrogen production over ternary Cu/TiO₂@Ti₃C₂T₂ enabled by low-work-function 2D titanium carbide. *Nano Energy* **53**, 97–107 (2018).
42. Deng, W. *et al.* All-Sprayed-Processable, Large-Area, and Flexible Perovskite/MXene-Based Photodetector Arrays for Photocommunication. *Adv. Opt. Mater.* **1801521**, 1–9 (2019).
43. Philippe, B. *et al.* Valence Level Character in a Mixed Perovskite Material and Determination of the Valence Band Maximum from Photoelectron Spectroscopy: Variation with Photon Energy. *J. Phys.*

- Chem. C* **121**, 26655–26666 (2017).
44. Ahn, N. *et al.* Trapped charge-driven degradation of perovskite solar cells. *Nat. Commun.* **7**, 1–9 (2016).
 45. Auf Der Maur, M. *et al.* The multiscale paradigm in electronic device simulation. *IEEE Trans. Electron Devices* **58**, 1425–1432 (2011).
 46. Deepa, M. *et al.* Cesium power: Low Cs⁺ levels impart stability to perovskite solar cells. *Phys. Chem. Chem. Phys.* **19**, 4069–4077 (2017).
 47. Sarycheva, A. *et al.* Two-Dimensional Titanium Carbide (MXene) as Surface-Enhanced Raman Scattering Substrate. *J. Phys. Chem. C* **121**, 19983–19988 (2017).
 48. Chaudhuri, K. *et al.* Highly Broadband Absorber Using Plasmonic Titanium Carbide (MXene). *ACS Photonics* **5**, 1115–1122 (2018).
 49. Alhabej, M. *et al.* Guidelines for Synthesis and Processing of Two-Dimensional Titanium Carbide (Ti₃C₂T_x MXene). *Chem. Mater.* **29**, 7633–7644 (2017).

METHODS

MAX phase and MXene preparation

Ti₃AlC₂ MAX phase was synthesized as described elsewhere.⁴⁹ In short, commercial powders of Ti, Al, and TiC were mixed together to achieve 3.0:1.1:1.9 molar ratio of Ti:Al:C in glass jar for 24 h using zirconia balls. The mixture was put in corundum crucible and heated to 1400 °C for 2 h under Ar atmosphere in tube furnace (Carbolite). The obtained MAX compact was crashed and sieved through a 400 mesh size sieve (stainless steel, Fritsch) to get a powder with an average particle size less than 30 μm. Ti₃C₂T_x MXene was synthesized by A-layer selective etching from the Ti₃AlC₂ precursor via less aggressive method, known as the minimally intensive layer delamination (MILD).⁴⁹ We used 1g of fine-ground MAX phase powder and slowly added it in a plastic jar with as-prepared mixture of 1g LiF and 20 ml of 6M HCL. The etching lasted for 24 h at 30–35 °C, while being stirred by a magnetic stir bar. After that, the etched powder was repeatedly washed with deionized water and centrifuged until neutral pH (~6–7) was reached. The Ti₃C₂T_x colloidal suspension in acetonitrile (ACN) was obtained by washing and centrifuging MXenes in ACN for several times to remove water. The MXene suspension was ice-cooled bath sonicated at 35 kHz for 1 h (Laborette 17, Fritsch) under continuous argon bubbling followed by 1 h of centrifugation at 3500 rpm to extract single-layer flakes.

MXene characterization

A transmission electron microscope, TEM, (JEOL JEM-2100, Japan) with an accelerating voltage of 200 kV was used to characterize structure of Ti₃C₂T_x. TEM samples were prepared by dropping delaminated Ti₃C₂T_x suspension onto a copper grid and dried in the air. The chemical composition of the MXenes surface was characterized by X-ray photoelectron spectroscopy (XPS) using an Axis Supra (Kratos Analytical, UK) spectrometer. The peak fitting was performed using a Gaussian-Lorentzian peak shape after the subtraction of a Shirley background by the CasaXPS software (version 2.3.17). X-Ray diffraction (XRD) spectra of MXene and TiO₂ samples were collected using a Rigaku Miniflex 600 X-ray diffractometer with monochromatic Cu-K_α radiation (λ=1.5406 Å).

Device realization

Materials

Mesoporous transparent titania paste (30 NR-D), formamidinium iodide (FAI) and methylammonium bromide (MABr) are purchased from GreatCell Solar, while Lead(II) iodide (PbI₂), Lead(II) Bromide (PbBr₂), and cesium iodide (CsI) complex are ordered from TCI and GmbH respectively. Moreover Cobalt(III) FK209 is purchased by Lumtec, while 2,20,7,70-tetrakis-(N,N-dip-methoxyphenylamine)9,9'-spirobifluorene (Spiro-MeOTAD) is from Borun. All other materials including titanium(IV) isopropoxide (TTIP), lithium bis(trifluoromethanesulfonyl)imide (Li-TFSI), acetylacetone, ethanol, acetone, dimethylformamide (DMF), dimethyl sulfoxide (DMSO), acetonitrile (ACN), tert-butylpyridine (tBP), chlorobenzene (CB), are purchased from Sigma-Aldrich. All materials are used as received unless specified otherwise.

Cell fabrication

Patterned Fluorine-doped tin oxide (FTO, Pilkington, $8\Omega\text{cm}^{-1}$) coated glasses were firstly washed with a cleaning liquid, dissolved in deionized water and then cleaned by ultrasonic bath with acetone and ethanol for 10 minute each step. The compact TiO₂ (cTiO₂) blocking layer (40nm) was deposited by spray pyrolysis at 460 °C and annealed for 30 minutes the same temperature. The cTiO₂ solution was composed by acetylacetone (2 mL), TTIP (3 mL) and ethanol (45 mL) and doped by Ti₃C₂T_x MXene dispersion (0.07mg/ml) insertion into the standard cTiO₂ solution with a volume ratio of 1%. (cTiO₂+MXene) The same MXene suspension, with the 0.42% volume ratio was used to dope the solution of mesoporous TiO₂ (mTiO₂+MXene) paste 30 NR-D in ethanol (1:10 weight ratio). This MXene-doped mesoporous TiO₂ (mTiO₂+MXene) solution was deposited on the substrate by spin-coating at 3000 rpm for 20 s and subsequently annealed in air for 30 min at 480°C by obtaining a 120 nm thick scaffold layer. The MXene interlayer was realized by spraying the MXene suspension (0.03 ml/cm²) by an airbrush onto the 80°C pre-heated substrate. The sample was immediately transferred to a N₂-filled glove box where was deposited the perovskite layer. The precursor solution, composed by a mix of FAI (1 M), PbI₂ (1.1 M), MABr (0.2 M), PbBr₂ (0.2 M) and CsI (1.5 M stock solution in DMSO) in the mixture of anhydrous DMF/DMSO (4:1 vol/vol) was stirred at room temperature for 30 min, and then was doped by adding 0.014mg of MXene in 1 ml perovskite precursor solution. The MXene doping amount has been optimized in term of cell PCE, as reported in **Figure S25**. The MXene-doped perovskite solution was spin coated onto the samples with a one-step deposition and anti-solvent method, consisting in a two steps program at 1000 and 5000 rpm for 10 s and 30 s respectively. During the second step, 200 µL of CB was poured on the spinning substrate 7 s prior to the end of the program. Immediately after spin coating, the substrates are annealed at 100°C for 1 h to form a perovskite crystal structure and by obtaining a compact perovskite layer 450nm thick. After that, a spiro-OMeTAD (73.5 g·L⁻¹) in CB solution doped with tBP (26.7 µL·mL⁻¹), LiTFSI (16.6 µL·mL⁻¹) and a Cobalt(III) FK209 complex (7.2 µL·mL⁻¹) was spin-coated at 2000 rpm for 20 s by achieving a hole transport layer of about 140nm. The precursor solution concentrations of LiTFSI are 520mg/ml in ACN and 375.8 mg/ml in ACN for Cobalt (III) FK209 respectively. Finally, the device was completed by the high-vacuum thermal evaporation of a gold counter electrode (~100 nm) on an active area of 0.09 cm² defined by a shadow mask.

Pristine and MXene-doped perovskite characterization

The UPS and XPS measurements were carried out at the Material Science Laboratory of the LNF-INFN (Frascati, Italy) on pristine and MXene-doped perovskite films deposited on a TiO₂/FTO substrate by exciting the sample with the HeIIα (40.81 eV) and Mg Kα (1253.6 eV) radiation, respectively, and collecting the photoelectrons with a Omicron EA 125 hemispherical analyzer. For the WF measurements a bias of 8 V was applied to the sample. All spectra were calibrated versus the Fermi level at zero binding energy measured on a clean metal surface.

Scanning Electron Microscopy: Electrodes were imaged by aim of a field-emission SEM (JOEL JSM-7500 FA). The acceleration voltage was set at 5 kV. Images were collected using the in-lens sensors (secondary electron in-lens image) and the secondary electron sensor (lower secondary electron image). No coating was applied.

Device characterization

Electro-Optical measurements: IPCE, ARKEO, I-V characteristics.

Current-Voltage (I-V) characteristics of masked and encapsulated devices are acquired in air atmosphere by using a solar simulator (ABET Sun 2000, class A) at AM1.5 and 100 mW cm⁻² illumination conditions, calibrated with a certified reference Si Cell (RERA Solutions RR-1002). Incident power was measured with a Skye SKS 1110 sensor. The class was measured with a BLACK-Comet UV-vis spectrometer. The devices were not preconditioned before the J-V measurements and a mask of 0.09 cm² was placed atop the solar cells; any variation of short circuit current density was observed in the case of smaller measurement masks. Both reverse and forward I-V scans were performed by using a scan rate of 20 mV/s and a dwell time of 200 ms. Any spectral mismatch factor correction was applied to PCE measurements since the class A Abet Sun Simulator employed for the I-V characterization has an error <1.3% in the spectral region of interest between 400 and 900 nm, that does not affect significantly the performance measurements accuracy.

Illumination intensity dependence of V_{OC} and dark JV measurements are performed with a modular testing platform (Arkeo - Cicci research s.r.l.) composed by a white LED array (4200Kelvin) tuneable up to 200 mWcm⁻² of optical power density and a high speed source meter unit (600 Ksamples/s) in a four wire configuration. A spring contact based sample holder is used to improve the repeatability of the experiments. Transient measurements are performed in a high perturbation configuration by acquiring the entire V_{OC} rise profile after switching the light intensity from ~0 to 1 Sun.

IPCE spectra acquisition are carried out by means of a homemade setup composed by a monochromator (Newport, mod. 74000) coupled with a xenon lamp (Oriel Apex, Newport) and a source meter (Keithley, mod. 2612). A home-made LabVIEW program controls the spectra acquisition.

DFT calculations and device simulations

First-principles calculations based on density functional theory (DFT) within the local density approximation (LDA) are performed using the Quantum Espresso package. Scalar-relativistic norm-conserving pseudopotentials are employed, with the exchange-correlation energy parameterized by Perdew-Zunger. Device simulations are performed by using TiberCad multiscale simulation software (TiberLAB s.r.l.).

DATA AND CODE AVAILABILITY. The experimental data, Quantum Espresso scripts for DFT calculation and TiberCad scripts for device simulations that support the findings of this study are available from the corresponding authors upon reasonable request.

REFERENCES

50. Alhabeb, M. *et al.* Guidelines for Synthesis and Processing of Two-Dimensional Titanium Carbide (Ti₃C₂T_x MXene). *Chem. Mater.* **29**, 7633–7644 (2017).